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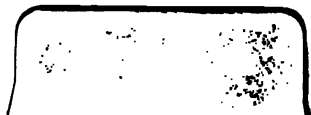
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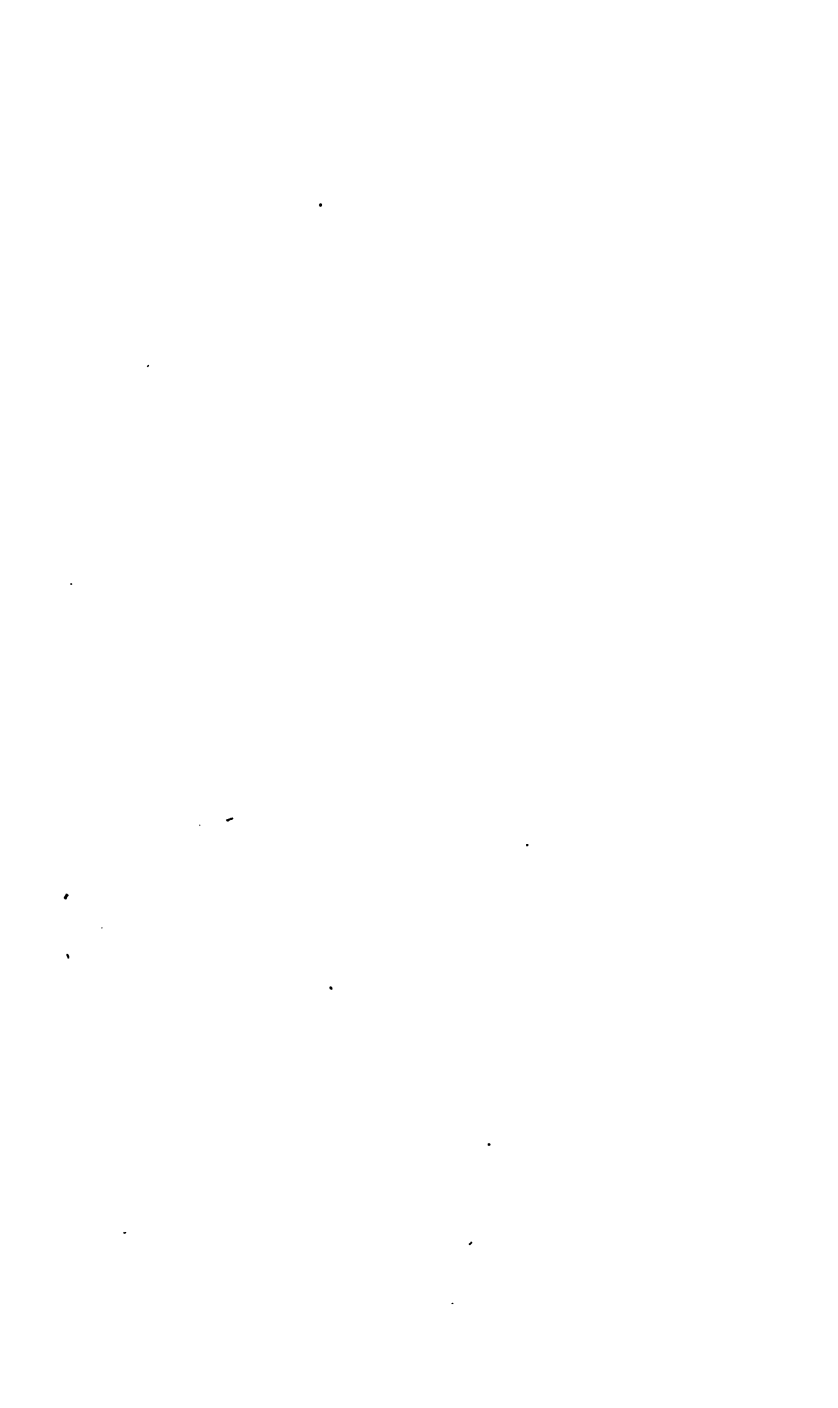
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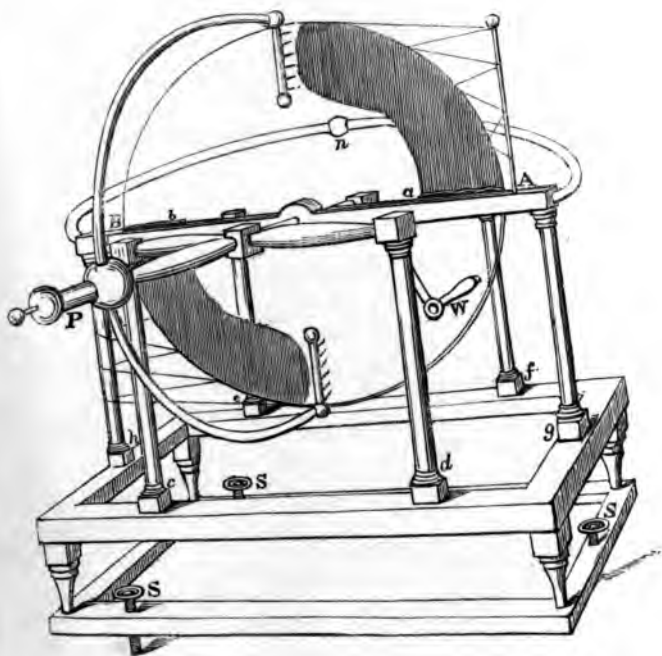
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AND  
THE PURPOSES TO WHICH IT HAS BEEN APPLIED.

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## P R E F A C E.

HAVING consented to prepare a small Elementary Treatise on Electricity for Mr. Weale's Series of Rudimentary Works, the writer has been desirous to confine himself within the limits of such a production as may be read with advantage by those who have begun to turn their attention to scientific pursuits. As a science, Electricity has, within a comparatively few years, progressed so rapidly, and proved so fertile in discoveries important to the public interest, and to further advances in natural knowledge, that it is by no means an easy task to bring within the circumscribing space of a short treatise, the many subjects Electricity is found to embrace. All, therefore, the author has attempted, is a sound practical and theoretical view of the subject, without involving the elementary student in any complicated or abstruse detail; but commencing with the simplest facts, illustrated by such elementary experiments as are easy and accessible, without great labour or cost. The scientific reader must not expect, therefore, to find much in this little book with which he is not familiar; yet the student, it is hoped, will have still benefited by its perusal, and further find the prosecution of his studies advanced, by the gradual development of some of the most important and broad principles of Electricity found in this unpretending Elementary Volume.

W. SNOW HARRIS.

Plymouth,  
16th December, 1848.



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## RUDIMENTARY ELECTRICITY.

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### I.

Phenomena observable in subjecting certain Substances to a peculiar species of Excitation by Friction—Origin of Electrical Terms—Electrics and Conductors—Insulation, Attraction, and Repulsion—Positive and Negative Electricity—Induction, or Electrical Influence—Other sources of Electrical Excitation—Manipulation.

1. *Phenomena of Electrical Excitation.*—When certain substances are subjected to a peculiar kind of excitation by friction they become apparently endowed with an attractive power; and if the excitation be powerful and be carried on in a dark room, faintly luminous flashes, and sometimes luminous sparks, attended by a weak crackling noise and a peculiar odour, may be observed to arise from the surface of the excited body.

*Exp. 1.* Let an ordinary stick of sealing-wax or a roll of brimstone be freely rubbed with a piece of dry woollen stuff, or soft white silk; it will become attractive of light particles of matter, such as fragments of metallic leaf, a downy feather, or small pieces of paper.

*Exp. 2.* Take a dry tube of glass, about an inch in diameter and about eighteen inches in length; rub the tube freely from one end to the other, through a dry soft silk handkerchief, held in the hand: small pencils of light and faintly luminous flashes and sparks, attended by a subdued crackling noise and a peculiar smell, will appear in the dark to dart from the surface of the glass, producing a marked sensation when near the hand or face. If the tube be now

presented to any light body, a powerfully attractive force will appear to be exerted by the tube. Downy feathers, fragments of cotton wool, attached to delicate threads, small balls turned of elder-pith, fragments of metallic leaf, leaf gold more especially, are the substances best adapted for such experiments.

2. These effects will be more sensibly produced if the tube be gently warmed by passing a current of warm air through it, and if the silk used to excite it be touched over with a compound of tin and sulphur, termed 'aurum musivum,'\* or mosaic gold, used by statuaries and painters. The rough side of oiled silk, oiled only on one side, will, when rubbed over with this substance, produce a powerful effect.

3. *Early knowledge of Electrical Attraction.*—This apparently attractive force superinduced upon bodies subjected to friction has been known from the earliest periods of which we have any scientific records. It was noticed full 600 years before our Christian era, by the learned Greek Thales, the father of the Ionic Philosophy, as being a curious property of amber. He was so struck with it that he considered the amber as having become endowed with a species of animation.

4. *Origin of Electrical Terms.*—Amber appears to be the first substance in which an attractive power was observed to arise by simple friction; hence, when other substances were found to possess a similar property, they were considered as being *amber-like*, and were said to be *electrical*, after the Greek 'ελεχτρον' (*electron*), or Latin 'electrum,' signifying amber.

It may be here observed that the common terms employed in this branch of science are almost entirely based on these Greek and Latin words. Thus the mysterious agency, whatever it be, the source of the observed attractive power, has been concisely expressed by the term *electricity*.

The substances in which an attractive power becomes de-

\* This composition is prepared chemically by first saturating the tin and sulphur with mercury and a little sal-ammoniac, and then subjecting the amalgam to a sand heat. It may be obtained readily at any chemist's.

veloped by friction are called *electrics*;—the attractive power observed, *electrical attraction*;—the peculiar state or condition of the given substance, *electrical excitation*;—whilst the excited body is said to be *electrified*.

Again, any particular contrivance for the mere exhibition of electrical force is called an *electroscope*, and those for the more precise measurement of such forces, *electrometers*.

5. Although the list of electrical substances in remote periods of science was extremely limited, being principally confined to amber, jet, and agate, yet in more recent times it has become so general as to include almost every known substance, as susceptible in a greater or less degree of electrical excitation. The substances, however, more especially termed *idio-electrics*, that is to say, those which under ordinary circumstances readily evince electrical properties by friction, may be brought within the following limits:—

TABLE I.—LIST OF ELECTRICAL BODIES.

- Shell lac—brimstone—amber—jet.
  - Resinous bodies of every kind, including pitch and wax.
  - Gums of every kind, including camphor and caoutchouc.
  - Glass and all vitreous and vitrified substances.
  - The diamond, agate, and most other precious stones.
  - Tourmaline, and other crystalline transparent argillaceous and silicious gems and stones.
  - Bituminous substances.
  - Silk of every kind and form.
  - Dried animal furs and skins—hair—wool—feathers—paper—porcelain.
  - Turpentine and various oils and fatty fluids.
  - All dry gases.
  - Atmospheric air.
  - Steam of high elasticity.
  - Ice at 0° of Fahrenheit.
6. *Friction the remote but not the immediate cause of attraction.*—The state of excitation of any of the more perfect



of these electric substances, such as shell lac, glass, brimstone, &c., appears confined entirely to those particles of the body immediately under the rubbed part, and is not apparent even in these, so long as the friction continues to be applied. Thus if a square of common window glass be subjected to friction on a portion of one of its surfaces, that portion, together with a similar portion of the opposite surface immediately under the rubber, will, on withdrawing the rubber, exhibit electrical excitation, but no other part of the glass will do so.

*Exp. 3.* Place a square of window glass, *A B*, fig. 1, made very dry, and slightly warm, upon two wine-glasses, as supports; and apply to its upper surface a flat circular rubber *R*, which may consist of a common cork bung, cut evenly and flat, and covered with silk, and having a rod of wood or metal *H*, fixed in its centre as a handle.

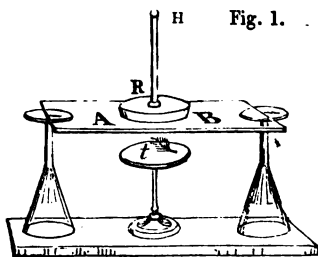


Fig. 1.

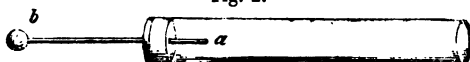
Place any light substance *t* immediately under the rubber, and about an inch, more or less, from the glass. A fragment of Dutch gold, or a light feather, answers very well. Turn the circular rubber carefully round so as not to derange its place on the glass. Not the least attractive effect will be observed on the light body *t* during the friction. Let the rubber be now raised off the glass, and the substance *t* will be vigorously attracted, but only by that portion of the glass which had been rubbed. If the rubber be prepared as in Experiment 2, the attractive effect will be vigorous, and exerted on the gold fragment at a considerable distance; and if the glass and rubber be fixed in a frame of wood, the results are more conveniently obtained.

7. About the year 1729 an important discovery resulted from some attempts of Mr. Stephen Grey to render metallic substances attractive by friction. He completely failed in

these attempts, but he found that although these substances were not excitable in the ordinary way, yet they might acquire an attractive power by placing them in communication with an electrified body.

*Exp. 4. Communication of electricity to non-excitable substances.*—Let a stout brass wire, about  $\frac{1}{8}$ th of an inch in diameter and a foot in length, be passed through a common cork, fixed in one extremity of the glass tube employed in Experiment 2, so as to project about 2 inches within the tube. Fix a small wood or brass ball on the projecting extremity of the wire, as at *b*, shown fig. 2, and excite the tube in the way before described (2).

Fig. 2.



The wire *ab* and the ball *b* will both become attractive of light bodies in common with the tube, and the excited electricity will be communicated to the ball and wire from its

Fig. 3.



inner surface, and with so much force as to cause luminous sparks to pass

from the ball upon the *finger* or other conducting body placed near it.

8. This communication of excited electricity to non-electric substances was found on further inquiry to extend through a considerable length of such substances. Thus, when a very long wire *bd*, fig. 3, having a metallic or ivory ball *d* at its lower extremity, was attached to the projecting rod, light bodies were attracted by it near the ground during the excitation of the glass tube on the summit of a high building. The first experiments of this kind were made with common packthread, which the experimenters suspended horizontally by supports of metallic wire, for the convenience of increasing the length: now in this case it was found impossible to

transmit the attractive force through the thread, whereas on suspending the packthread by *cords of silk* or other electrics, the electrical action became transmitted through a distance of 765 feet. These silk cords, therefore, had satisfied some important condition of the experiment. When a silk line, however, was substituted for the packthread, the electrical excitation of the tube was no longer apparent in the ball at the remote extremity, in whatever way it was suspended. From this it became apparent, that the class of substances termed electrics had not only the property of electrical excitation, but they had also the power of confining or imprisoning, as it were, the communicated electricity upon other bodies, in which it could not, under the same circumstances, be so excited: on the other hand, the non-electric substances allowed the excited electricity to pass off to the walls of the building and to the ground.

9. *Electrical conduction*.—Hence arose a knowledge of a distinct series of phenomena dependent on what has since been termed *Electrical Conduction*, giving origin to a new classification of substances considered as conductors of electricity. The substances which properly come under this *conducting* or *non-electric* class are principally as follows:

TABLE II.—LIST OF ELECTRICAL CONDUCTORS.

Every metallic substance known.

Well-burned charcoal.

Plumbago.

Concentrated and diluted acids and saline fluids.

Water and moist vegetable matter.

Living animal matter.

Flame—smoke—steam.

10. The distinctive difference in the conducting and non-conducting property of bodies may be readily illustrated in the following way:

*Exp. 5.* Excite the glass tube and wire employed in the last experiment, and bring the ball of the wire into contact

with any of the substances in Table I., termed *electrics*, such as a rod of glass, a stick of sealing-wax, or brimstone rendered perfectly dry. The attractive power of the ball and wire, together with the tube, will not be in any sensible degree impaired. Let the electrified ball now touch the walls of the room or other conducting substance communicating with the ground: the attractive power will instantly vanish.

It is evident from these facts that all electric substances (5) are *non-conductors* or *insulators*, as they are also appropriately termed; whilst, on the other hand, *non-electric substances* are transmitters or conductors of electrical action.

11. *Insulation*.—When, therefore, any conducting substance is placed on an electrical support, such as a rod of glass or shell lac, it is considered to be insulated, and is termed an *insulated conductor*: when electrified by contact with any excited or other electrified body, it is said to be *charged*. The electrical charge thus communicated to an insulated conductor appears to be collected about its surface, and to be rather dependent on that than on the solid content. Thus, if two metallic spheres or cylinders, the one solid, the other hollow and extremely light, be suspended by silk lines, or placed on dry insulating supports, and be charged by contacts with the excited tube, fig. 2, the attractive energy of each upon any light substance presented to it will be found quite alike in each. In this experiment the insulators must be very dry and perfect.

The best insulating substances are of the vitreous and resinous class, Table I., such as shell lac, brimstone, dry glass rods, vitrified and crystalline bodies. To these may be added silk.

The best conducting substances are principally metallic bodies, saline fluids, and common charcoal.

12. It should, however, be here understood, that modern researches, especially those of Faraday, lead us to conclude that there are really no substances which perfectly conduct or

perfectly obstruct electrical action. The *insulating* and *conducting* power is in fact a difference of degree only; still the extreme differences are so great, that if classed in relation to such differences, those at the extremes of the series admit of being considered, the one as insulators, the other as conductors, whilst the intermediate terms are made up of substances which may be considered as imperfect, taken as either. Conversely, every substance is capable of excitation by friction; yet the differences in this respect are so great as to admit of some bodies being called electrics and others non-electrics, with an intermediate class between these extremes, which may be termed imperfect electrics.

*Series of conductors and insulators.*—Metals and concentrated acids are found at the conducting extremity of such a series,—shell lac, brimstone, all vitreous and resinous bodies, at the other or electric extremity, whilst the imperfect or intermediate substances comprise such matter as common earth and stones, dry chalk, marble, porcelain, paper, and alkaline matter.

13. The attractive power evinced by any electrical body in a state of excitation, although the first and usually the most evident electrical effect, is yet not the only force which seems to result from this curious condition of common matter. On a closer examination of the phenomena, a new class of facts present themselves, of remarkable interest. If the excitation be considerable, and the attracted body insulated, it will, after being drawn into contact with the electrified substance, rebound from it, with great violence, as if repelled by some new power, and will not be again attracted until it has had conducting communication with the earth, or some other mass of matter capable of reducing it to its original condition before the contact.

*Electrical repulsion.*—The attractive force becomes, in fact, superseded by an opposite or repulsive force, and thus two species of action appear to be obtained.

*Exp. 6.* Attach a slip of metallic leaf, about half an inch

in width and 4 inches long, to a short and thin slip of writing paper, which is easily done by a little moisture from the lip. Fix the paper to the extremity of a stick of common sealing-wax, or a slender rod of glass coated with sealing-wax, as represented in fig. 4: present this leaf, Fig 4.

thus insulated by the wax, to the metallic ball of the excited tube described in fig. 2: the leaf, after being vigorously attracted, will instantly on contact with the ball appear to be repelled, and will not be again attracted until reduced to its original state, either by contact with the earth or some other unelectrified mass. A simple reed, suspended by a short cotton or silk thread, may be used for this purpose with advantage.

A test or trial leaf or reed, when prepared and thus insulated, constitutes an extremely good electroscope, and is well adapted for the exhibition of simple electrical phenomena. Leaf gold, and silver leaf, may be occasionally employed for minute forces, but it is managed with difficulty. The best kind of metallic leaf for ordinary purposes is a coarse kind of leaf termed Dutch metal. The white metal is often the most compact, but the yellow metal is sometimes preferable.



*Exp. 7. Electrical attractions and repulsions.*—Excite a tube of glass as in Experiment 2, and bring it near a small ball of cork or elder-pith, resting on a table. The ball will appear to bound from the table towards the tube, from which it will be as instantly repelled; on touching the table it will be again attracted, and thus a rapid play of what appears to be a series of attractions and repulsions may be maintained for some time.

In all these experiments the excitation requires to be vigorous, and this may always be insured by the means before described (2).

14. On a further examination of these attractions and repulsions, we find a most important relation of electricity to common matter, leading to the conclusion that in every case of

electrical excitation, as well as in every other instance of electrical action, two equal and opposite forces or powers are called into play. These, when combined, condense or saturate each other, and thus neutralize the free action of either, as observed in the phenomena just described.

*Exp. 8. Two kinds of electrical power.*—Let a simple leaf, such as that employed in the last experiment, be made repellent of an excited glass tube, as before, so as to be thrown freely off it. Then, whilst in this state, present it to an excited roll of brimstone or sealing-wax. The leaf will appear to be vigorously attracted by this substance. Conversely, present the leaf first to the excited brimstone, and when made repellent of that, to the excited glass,—a powerful attractive effect will succeed.

It is here evident that one of these substances will, under given conditions of excitation, attract an electrified body made repellent of the other.

*Vitreous and resinous,*—or *positive and negative electricity.*  
—We are led from this interesting fact to infer the existence of two opposite electrical or elementary states of excitation in which forces are developed attractive of each other. These forces were by the more early inquirers supposed to depend on two different kinds of electricity, and were termed vitreous and resinous electricities, as being derived, the one from excited vitreous bodies, the other from excited resinous bodies: subsequent investigations, however, as we shall presently see, show that both these hypothetical electrical elements may be obtained from the same electric, merely by changing the substance employed to produce the friction. Hence, as involving less assumption, it has been agreed to designate the opposite electrical states of which these terms are really expressive by the common positive and negative signs employed in arithmetic and algebra, calling the one positive or plus, and the other negative or minus; the positive being given to the vitreous excitation, as developed in the friction of glass; the negative sign — being appro-

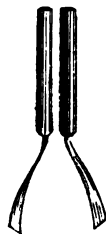
priated to the excitation of resinous bodies by silk or woollen stuff.

*The terms positive and negative an arbitrary selection.*—It is to be understood here, however, that the appropriation of these terms is altogether a matter of arbitrary selection. When any substance, therefore, evinces vitreous electricity, it is said to be electrified positively, or plus + : when it evinces resinous electricity, it is said to be negatively electrified, or minus — : when unelectrified, or in its ordinary state, it is said to be neutral.

15. From these phenomena it may be inferred, that substances in dissimilar states of electricity, that is, the one +, the other —, *attract* each other. Substances in similar electrical states, that is, both + or both —, *repel* each other ;— a concise and simple formula.

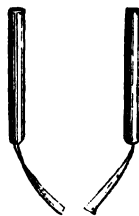
*Exp. 9. Similar electricities repel — opposite electricities attract.*—Prepare two similar test leaves of metal, as in Experiment 6 : let each be made repellent of an excited glass tube or of an excited roll of brimstone or sealing-wax : oppose these electrified leaves fairly to each other, and they will immediately diverge, as represented in the annexed fig. 5.

Fig. 5.



*Exp. 10.* Make one of the trial leaves repellent of excited glass and the other repellent of excited wax or brimstone ; oppose them, as before, and they will immediately converge and approach each other, as represented in fig. 6.

Fig. 6.



16. A glass tube and wire, such as described Exp. 4, page 5, is an efficient and convenient arrangement for obtaining vitreous or positive electricity ; and if the tube be well varnished over with a solution of shell lac in rectified spirits of wine, laid on the glass previously warmed, resinous or negative electricity may then be obtained from the wire and ball, and in



a similar way by exciting the surface with dry woollen stuff or soft white silk : the same result is obtained by making the tube sufficiently hot, and rubbing it over with sealing-wax, so as to give the surface a thin coating.

17. In treating of the phenomena of excitation we have as yet considered the development of one only of the two electrical forces ; on a closer examination, however, it is found that both forces are produced, although only one is usually apparent. This may be made evident by insulating the rubber, that is, the body with which the electric is said to be excited ; in this case the rubber will be found also attractive of light bodies, and its electrical state will be precisely the reverse of that of the excited body. The reason, therefore, why this force is not apparent in ordinary cases of excitation, is, in consequence of the rubber being usually in contact with conducting matter, such as the hand, by which its electricity is neutralized through conducting communication with the earth (8).

*Exp. 11. Both electricities produced in every case of electrical excitation.*—Wrap a broad slip of soft silk, or a wide silk ribbon, round a rod of glass or sealing-wax, to serve as an insulating handle : apply the roll thus produced as a rubber upon a strip of common window glass of equal width, made perfectly dry and a little warm : after slight friction, examine both the silk and the glass ; each will attract light substances, but they will be in opposite electrical states, as is seen by the one attracting where the other repels. This may be shown, as in the last experiment, by an insulated strip of metallic leaf, which may be suspended from some convenient support. The experiment may be repeated with a small roll of dry woollen stuff and a stick of sealing-wax or brimstone.

18. *Opposite electricities dependent on the conditions of excitation.*—The precise electrical state excited in various substances by friction depends not only on the kind of electric, but on the substance by which it is excited. Thus a rod of

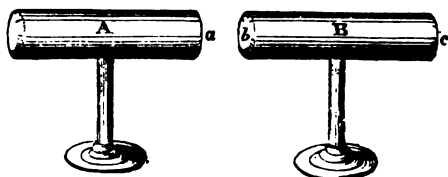
glass, rubbed by white silk, evinces that peculiar state of electricity which it has been agreed to call *vitreous* or *positive*; but when the glass is rubbed against the back of a cat, it evinces what has been called *resinous* electricity; that is to say, an opposite or *negative* state of electricity. Glass rubbed with silk gives positive electricity, but sealing-wax rubbed with the same silk evinces negative electricity. Conversely, silk rubbed with glass becomes negative, but rubbed with sealing-wax is positive. Hence it is quite apparent that bodies are differently excited electrically by friction with the same or different substances. It may, however, be instanced as a remarkable exception, elicited by such inquiries, that the back of a cat, by any experiment yet made, excites every substance negatively, and invariably produces positive electricity. Positive and negative electricity still, however, are simultaneous phenomena, and are actually both produced in every case of electrical excitation.

19. *Electrical influence*.—These first or more elementary facts being understood, we are prepared to enter upon a new class of phenomena, remarkable for their great interest and importance in the progress of electrical discovery. We have hitherto confined our attention to the attractive and repulsive powers developed in bodies either by a direct excitation (1), or by immediate communication (7); but there is yet another kind of electrification to be considered as depending on the influence of excited bodies upon conducting bodies, and exerted at very sensible and even considerable distances. This influence has been termed *Electrical Induction*, and the resulting effect, *Induced Electricity*.

*Exp. 12.*—Insulate upon supports of varnished glass two small cylinders of wood or metal, A B, fig. 7, from 3 to 5 inches in length and about 3 inches in diameter, and terminating in flat faces: place these cylinders within an inch or more of each other, and in a right line, as represented in the figure. Excite the glass tube and wire (fig. 2, page 5) powerfully (2), and communicate by repeated contacts of

the ball an electrical charge (11) to one of these conductors **A**. Present now to the distant extremity *c* of the cylinder **B** the slip of metallic leaf or other light substance, as de-

Fig. 7.



scribed in Experiment 6: it will be immediately drawn towards it. Present the leaf also to various parts of the cylinder **B**, between the extremities *b c*: the attraction will diminish considerably as we pass from the distant extremity *c*, until it becomes nearly nothing.

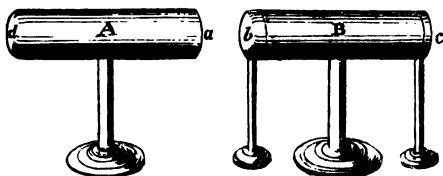
If we withdraw the electrified body **A**, these phenomena are no longer apparent, thus proving that the electrified state of the cylinder **B** was an induced or temporary condition, depending on the influence of **A**, whilst operating upon it at a distance.

*Exp.* 13.—Place the two cylindrical conductors near each other, as before, and communicate a charge to one of them **A**; make the test leaf repellent of **A** by contact with it, and then present it to the distant extremity *c* of the cylinder **B**. It will be also repellent of that extremity, showing that the electrical states are the same, but the repulsive effect will diminish considerably as we approach the near extremity *b*, where it will appear to be nearly lost.

20. If the precise condition of the conductor **B**, whilst under the influence of the charged conductor **A**, be carefully investigated, its extremities *b* and *c* will be found actually in opposite electrical states; for although one of these only, in the distant extremity *c*, is apparent whilst under the influence of the charged conductor **A**, yet by a slight variation in the experiment the opposite electricity may be also discovered.

*Exp. 14. Electrical induction exemplified.*—Let the terminating faces *b c*, fig. 7, of the cylindrical conductor **B** consist of thin slices of about  $\frac{1}{4}$ th of an inch or more in thickness, supported upon slender glass rods, as shown in figure 8, and applied so as to fit closely against the opposite ends of **B**. Charge cylinder **A** as before (Exp.

Fig. 8.



12), and remove the terminating face *b* by its insulating rod whilst under the induction of **A**. This face *b* will now attract the trial leaf when the leaf is repellent of the charged cylinder **A**, and repel it when the leaf is attractive of **A**, thereby showing that it is in an opposite electrical state; that is to say, if **A** be charged positively, *b* is charged negatively; and if **A** be negative, *b* is positive (15). Proceed now to remove in a similar way the distant face *c*, we have then the converse of this result (Exp. 13); that is to say, the face *c* will be found, as before, attractive of the trial leaf when that leaf is attractive of **A**, and repellent of it when repellent of **A**; thereby showing that its electrical state is the same (15): hence its being formed of a thin slice, and applied as an extremity of the cylinder **A**, does not at all interfere with the result.

It may not be out of place here to observe, that the electrical disturbance in the near face *b* will be most strikingly shown if removed under the influence of **A** when the distant face *c* of the cylinder **B** is extended in length, or is in conducting communication with the earth.

21. This electrical disturbance by induction, although more particularly observable in the neutral conductor, is, however,

not exclusively confined to it: on further investigation we find a sort of reflected induction on the charged body itself, causing a considerable change in its previous condition.

*Exp. 15.* Construct the cylinder  $\Lambda$  with moveable terminating faces  $a$   $d$ , as in the case of the cylinder  $\mathbf{B}$ , or what comes to the same thing, charge the cylinder  $\mathbf{B}$ , and observe by the trial leaf the repellent effect of each of its terminating faces; after which oppose to it the conductor  $\Lambda$  in a neutral state, placed in conducting communication with the ground. Under these conditions, the repellent effect towards the distant extremity  $c$  will be considerably diminished, and towards the near extremity  $b$  evidently increased; and if we remove each of the terminating faces, as before, by their insulating supports, the repellent power of the near face  $b$  will be found greater than that of the distant face  $c$ , which face may become under some conditions quite neutral, and evince even an opposite negative state.

22. This peculiar kind of electrical action, termed *Induction*, appears essential to the phenomena both of attraction and repulsion (3) (13): it is invariably attendant on both, and in all probability precedes them; that is to say, in every case of electrical attraction or repulsion the bodies are first rendered attractable or repellent, and then attracted or repelled; and without this previous preparation, neither of these effects are produced. Thus it is found that the attraction of an excited electric, or of a charged conductor, is less forcible upon electrics little sensible of induction, than upon conductors highly susceptible of such change.

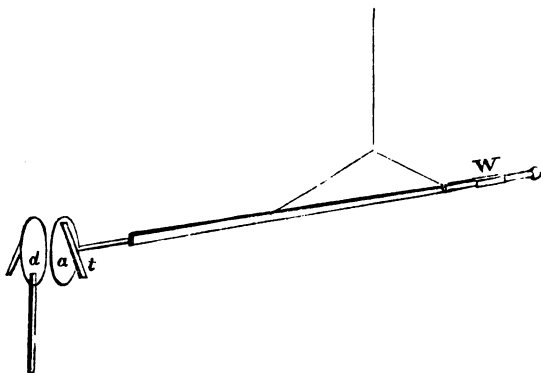
*Exp. 16. Attraction and repulsion attended by induction.*

—Cover a small sheet of writing paper, on each side, with smooth gilt paper, so as to produce, when pasted and dry, a stiff plate; cut out a circular disc of about 3 inches in diameter, and attach to one side of it a slip of metallic leaf, so as to hang freely about its surface, as in the annexed fig. 9, which is easily done by uniting it



to a thin slip of cork, cemented to the disc by a little strong paste; attach this disc to a thick filament of glass, at right angles to the direction of the leaf; by means of a little sealing-wax, fix the glass in the extremity of a light arm of deal, suspended by a fine thread of silk from the ceiling, as shown in fig. 10; balance the system by means of a small loop

Fig. 10.



for suspension, and a light sliding weight  $w$ , as in the figure. The system being thus free to move about a centre, present to the disc  $a$  an insulated electrified disc  $d$ , which may be charged negatively, if required, by means of the tube, fig. 2, page 5, if it be covered with sealing-wax or a thick varnish of shell lac (16): the leaf  $t$  will be observed to leave the surface and spring away from it, whilst a similar leaf attached on the opposite surface of  $d$ , and in a state of divergence, will tend to collapse: at this instant the discs will appear to attract each other.

*Exp. 17.* Charge both the discs  $a$  and  $d$  with the same electricity, and, whilst the leaves are divergent, present the disc  $a$  to the disc  $d$ , as before: the leaves will diverge still more energetically, and the bodies will then repel each other.

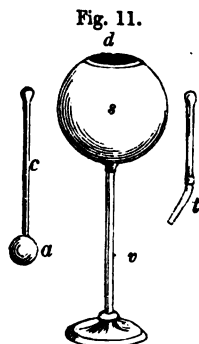
*Exp. 18.* Charge the discs with opposite electricities, and present them to each other, as before: the leaves pre-

viously divergent will now tend to fall back, and will collapse considerably; the discs will then freely attract each other.

23. It is to be further observed, as a most important and characteristic feature of electrical action, that neither induction, attraction, or charge, have any dependence whatever on the solid contents of conducting bodies, or even on the kind of substance of which such bodies consist. Thus, whether an insulated conductor (11) be a thin hollow body of silver, tin, wood, or any other conducting substance, the electrical charge it is capable of receiving, and the subsequent induction or attraction it exerts, is in each case precisely the same: the only condition or element which enters into such developments is time. Inferior conductors, such as wood, require in some cases a very small portion of time for the accumulation and yielding up of electricity, whilst the action of the most perfect conductors, such as metals, is as it were instantaneous. The celebrated French Philosopher, Coulomb, found that an equal division of electricity took place between two insulated conducting substances of equal surface and similar form, whatever differences existed in the kind of substance of which they were composed, and although one were a mere shell of matter and the other perfectly solid.

The following experiments are highly instructive and important.

*Exp.* 19. Let  $s$  be a thin hollow metallic sphere about 4 inches in diameter, and having a circular opening at  $d$ , about an inch and a half across. Place this sphere on a long insulating support  $v$ . Let  $a$  be a small ball of brass about  $\frac{3}{4}$ ths of an inch in diameter, either solid or hollow, and insulated on a slender rod of glass  $c$ . Charge the small sphere  $a$  with electricity so as to render it powerfully attractive and repel-



lent of an insulated trial leaf electroscope  $t$  (13), and then plunge it within the interior of the shell  $s$ , but without touching the edge of the opening at  $d$ . Having brought this electrified ball in contact with the interior of the shell, withdraw it again by its insulating handle carefully, in a similar way. It will be found that every particle of the charge will have left the small sphere  $a$  to appear on the outer surface of the sphere  $s$ , which will have now become attractive and repellent of the trial leaf  $t$ : by a few repetitions of contact of the inner surface of  $s$  and the small sphere  $a$ , after being charged, the exterior charge upon  $s$  may be rendered very powerful; for notwithstanding the previous communication of electricity to  $s$ , the small sphere  $a$  will be continually and completely robbed of its charge by contact with the interior of the shell.

When this experiment is repeated with spheres of wood and metal, hollow or solid, but of the same dimensions, precisely similar results are obtained; and it may be likewise shown, that the charge communicated to an insulated sphere of metal, either solid or hollow, will be equally shared by contact with a sphere of wood, either hollow or solid. In the use of solid spheres for such experiments we require an interior cylindrical cavity at  $d$ , sufficient to receive the insulated charged body.

24. It may be inferred therefore, as before observed (11), that the accumulative force of electricity has little dependence on the solid content or kind of substance of a conductor, but is influenced rather by the *extent* of surface; and such is found to be the case. Thus of two spheres the surfaces of which are to each other in a given proportion, say as 1:2, the larger surface will require twice the charge to evince the same activity; for the result depending on the surface, it follows that if we dispose a given quantity of electricity upon twice the surface, we have only one-half the electricity in any given point; and hence the action upon an electroscope is dependent on the other half, and is consequently



diminished. Now when we double the charge, there is again the same quantity of electricity in any given point; consequently the activity is the same as before. It has been ascertained by conclusive experiments, that the force of electricity thus accumulated is as the square of the quantity *directly*, and as the square of the surface *inversely*.

25. *Other sources of electrical excitation.*—We have as yet considered ordinary electrical excitation as arising exclusively from friction; it will, however, be requisite to understand, that although friction under some form is for common purposes the most available source of this peculiar state of certain bodies, yet it is not the only source: a great variety of operations, both natural and artificial, are also causes of electrical disturbance; such, for example, as pressure, simple contact, and other mechanical processes, — changes of temperature, — changes of form, — chemical changes, — magnetic influence, and some others. The following are interesting illustrations.

*Exp. 20. Excitation by change of form and temperature.*

—Liquefy a little common brimstone by a gentle heat over a fire in a covered earthenware vessel, and pour it into a dry wine-glass; put a short rod of glass into the liquid, to serve as an insulating handle: when cold, remove the solid brimstone cone from the glass, and it will be found, by means of the trial leaf, to be in a state of active excitation, together with the glass, which will be in an opposite or positive state. Chocolate and several substances become electrically excited in this way in passing from a fluid to a solid state.

*Exp. 21. Excitation from changes of temperature only.*—

Expose a piece of tourmaline to a moderate heat by placing it on a common watch-glass, carefully held over a spirit lamp for a short time. This curious stone, in cooling, will exhibit a strong electrical action, as if excited by friction. Several crystallized gems become electrical in this way.

*Exp. 22. Excitation by chemical action.*—Pour some dilute sulphuric acid on coarse iron filings contained in a

green glass bottle : during the chemical action which ensues the bottle will show signs of electrical excitation, which may be seen by presenting to it a trial leaf.

*Exp. 23. Excitation by pressure and contact.*—Press a black and white silk ribbon fairly together by drawing them between the finger and thumb ; they will adhere strongly, but without evincing any electrical sign externally : proceed now to detach these ribbons by pulling the surfaces from each other ; each ribbon will, as in the case of Exp. 3, page 4, be now found attractive of light bodies, and the ribbons will again, if allowed, fly together.

Laminæ of Muscovy talc or mica will exhibit, when forcibly separated, strong electrical signs.

*Exp. 24. Excitation by contact only.*—Prepare a series of about a thousand or more small circular discs of silver, zinc, and paper ; place them in a glass tube previously well dried, taking care that the discs are all placed in the same order, as *silver—paper—zinc—silver—paper—zinc*. Secure the extremities of the tube with corks or metallic caps, and let short wires pass through them, so as to press upon the terminating plates of the series, as in the annexed figure 12. Then, by the mere contact or association of

Fig. 12.



these bodies in consecutive groups, each extremity *a*, *b*, of this *dry pile*, as it is called, will evince opposite electrical states ; the zinc extremity being positive, and the silver extremity negative : a pile of this kind has also been termed the *electrical column*.

*Excitation by the contact of metals and fluids.*—Another powerful and peculiar source of electrical excitation consists in a similar arrangement of metals with water or moistened cloth interposed, commonly termed the *voltaic pile*. Copper and zinc are the metals usually employed. If a series of about 50

circular or square discs of these metals, of about an inch in diameter, be grouped in the order of *zinc—moistened cloth—copper*, both the zinc and copper extremity will exhibit attractive and repellent effects, such as already described (15). The metallic plates, when square, may be joined at their faces, and placed so as to form a series of cells in a trough of wood, as in fig. 13. In this arrangement, if river water be poured into the cells, the electrical power of the terminating plates is considerable; the zinc extremity being electrified positively, and

Fig. 13.



the copper negatively; and if the interposed fluid be a solution of common salt in water, a faint shock will be experienced on touching the opposite extremities of the series.

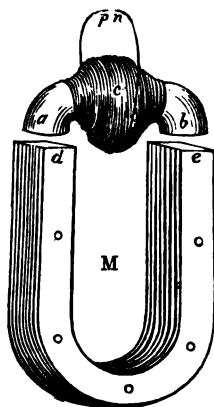
*Excitation by living animal matter.*—Certain fish, such as the torpedo and gymnotus, have a similar electrical power, derived from a peculiar organ acted on by a fluid, and which enables them at will to excite a peculiar and very strong electrical action.

A peculiar species of electrical excitation may be caused by the influence of magnetism on metallic wires, exerted either by the natural magnet termed the loadstone, or by a combination of steel bars forming a compound artificial magnet.

*Exp. 25. Excitation by magnetic influence.*—Coil some hundred feet of copper wire, covered with silk, carefully round a soft iron bar, allowing the extremities of the bar to project freely beyond the coil. Fix the ends of the coil extremely near each other, and bring the ends of the bar into contact with the poles of a very powerful magnet. A sort of electrical disturbance will ensue throughout the whole extent of the wire, which may be made to manifest itself in the form of a spark between the ends of the

coil, and to exhibit other electrical indications ; and this effect will be produced whenever we break or make contact with the magnet. Let *m*, fig. 14, be a battery of bent magnetic bars ; *a b*, a bar of soft iron having from 300 to 500 feet of copper wire coiled round it at *c*, the extremities of which, *p n*, come very near each other. Then, when a contact is made between the projecting ends *a b* of the bar of soft iron and the ends or poles *d e* of the magnet *m*, electrical indications will ensue between *p* and *n*. To facilitate the effect, one extremity of the coil is plunged in a small cup filled with mercury, and the opposite end made to close as it were upon the surface, so as to be very near it.

Fig. 14.



We are indebted to the admirable researches of Dr. Faraday for this beautiful fact.

With an arrangement of this kind, and with a coil of about 3000 feet of wire, M. Pixii, of Paris, obtained ordinary electrical attractions and repulsions on leaf gold. In this case the magnet was made to revolve rapidly on an axis, under the projecting ends of a curved bar of soft iron, carrying the coil. The poles of the magnet and the ends of the bar being very accurately placed, so as to produce an insensible contact, every time the poles of the magnet passed the ends of the bar, a continued current of most brilliant sparks ensued between the extremities of the coil through the medium of mercury. Similar results were subsequently obtained by Mr. Saxton, who, by a most ingenious mechanical arrangement, caused the iron and coil to revolve instead of the magnet.

26. *Electrical Manipulation.*—A few remarks on the elec-

trical manipulation requisite to the success of the several experiments we have hitherto described, may not be here altogether out of place, observing that these experiments have been purposely selected as being of the most simple and easy kind.

In the first place we have to remember, that since water is found in the class of conductors (9), and that the success of electrical experiments mainly depends on good insulation, it is most important to select a dry atmosphere for such experiments; or otherwise to render the air dry artificially by an Arnott's stove, which is admirably adapted to this purpose. It will be further desirable in certain cases to perfect the insulations by exposing them to the warmth of a small iron, heated to redness, and curved in such way as nearly to encircle the insulating support, as represented in the annexed fig. 15.

Fig. 15.



Moreover, the surfaces of glass rods used as insulators should be carefully varnished over with a solution of shell lac in rectified spirit, laid on the glass before a fire, the glass being previously warmed. If these precautions be not taken, the accumulated electricity will speedily disappear by conduction over the surface of the insulators (8). Silk threads intended for insulation should be treated in the same way.

27. The principal articles requisite to the student for the prosecution of early experimental inquiries are the following: A few tubes of glass, varying from  $\frac{1}{2}$  to  $1\frac{1}{2}$  inch in diameter; some glass rods from  $\frac{1}{8}$ th to  $\frac{1}{2}$  an inch in diameter; a few filaments or stout threads of glass, of about the  $\frac{1}{20}$ th to the  $\frac{1}{10}$ th of an inch in diameter; silk thread of various sizes; fine threads of unspun silk from the silk-worm, for the suspension of light bodies; shell lac, common sealing-wax, a roll or two of brimstone, and one or two other electrics of the resinous class (5); a little soft white silk, coarse oiled silk, fine leather, and some dry woollen stuff and dry hare-skin, for excitation; a piece of aurum musivum; straw reed of

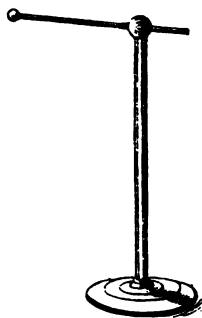
various diameters; light cotton and linen thread; a few small balls of wood and cork of various sizes, from  $\frac{1}{4}$  to  $\frac{1}{2}$  an inch diameter; some small balls formed of the pith of elder, varying from the  $\frac{1}{30}$ th to the  $\frac{1}{3}$ rd of an inch in diameter, and which are easily produced by turning the pith about in small hemispherical cavities formed in any soft body capable of attrition, such as soft sand-stone; sundry other light bodies, such as cotton wool, downy feathers, &c.; a few books of Dutch metal and gold and silver leaf; metallic rods of various sizes,—brass wire answers very well; a few hollow metallic balls,—these are usually made of brass, and polished; a solution of shell lac in spirits of wine.

28. Electrical excitation is also greatly promoted by the use of an amalgam of zinc, tin, and mercury, in the proportion of 1 part tin, 2 parts zinc, and 4 parts mercury. The zinc should be first exposed to a melting heat in an iron ladle, the tin then added, and finally the mercury, gently heated in another small iron ladle, should be poured gradually upon these liquefied metals, stirring the mixture carefully at the same time. After allowing the amalgam to cool down, it should be poured, just before solidifying, into a wooden or iron box, and be constantly agitated by shaking until cold. It will commonly be found in the form of a powder, or an easily pulverized mass. This should be now triturated in an iron mortar, and sifted through a small muslin sieve, so as to obtain an extremely fine powder: this powder, when employed to promote excitation, should be rubbed up in a mortar with a little lard, just enough to hold the particles together, and be then spread on the rubber with a common palate knife. It is principally employed for the excitation of glass, which may be powerfully excited with this amalgam, spread on the rough side of oiled silk or on basil leather.

29. An insulating stand for the suspension of light bodies will be found very convenient to the student; also one or two insulating tables. The former is easily obtained by running a slender rod of glass through a ball of wood or cork, and

fixing the ball on an extremity of a slender glass rod, as represented in the annexed fig. 16. A horizontal arm of this kind may be well employed for suspending one of the trial leaves, fig. 5, page 11, whilst observing the action of the other. Wine glasses with long stems, made dry and varnished, are convenient, when inverted, as small insulating tables: a common watch-glass, cemented by sealing-wax at the extremity of a long slender rod of glass, fixed on a firm base, as in fig. 16, is a very effective insulation.

Fig. 16.



30. In conclusion it may be observed, that in all cases of excitation it is desirable to preserve the opposite surface in a perfectly dry state, and protected from moisture: thus plates of glass will excite more freely when their opposite surfaces are varnished with shell lac, or what is better, coated over with sealing-wax. The electrical tube or ball described in Exp. 4, fig. 2, sect. 7, being extremely available to the purpose of the student, should be thus treated: the interior surface should be either varnished or coated with sealing-wax;—this is effectually done by first pulverizing the sealing-wax, and then placing it within the tube; after which apply a strong heat sufficient to dissolve the wax, and allow it to float over the surface of the glass. A tube thus prepared with the amalgam just described (28) will never fail, when dry, to evolve positive electricity in great quantity; and, when coated on the outside with sealing-wax or a thick coating of shell lac varnish, negative electricity also, as before observed (16). In this case, however, it must be excited with woollen, or soft white silk, or hare-skin.

## II.

Prevailing Theories of Electricity—Hypothesis of Symmer and Du Fay—  
The Franklinian Theory—Eales's Theory—Faraday's Views and Theory  
of Electrical Induction.

31. Having now considered and illustrated by simple experiments some of the more obvious and elementary phenomena of ordinary electricity, those which first demand attention, and upon the clear apprehension of which all further progress in this department of science mainly depends, we may now briefly advert to the prevailing theories, or rather philosophical speculations, which have been advanced in explanation of them.

If we examine the notions entertained by some of the more early Electricians, we find them to be, in many instances, of a somewhat rude and unsatisfactory character. Boyle imagined that an electric emitted a sort of glutinous effluvium which laid hold of small bodies and brought them back to the excited electric. The celebrated Newton, however, in two Queries at the end of his 'Optics,' appears to imagine, with his accustomed diffidence, that electric bodies, when excited, emit a very attenuated elastic fluid or exhalation, in consequence of the vibratory motion of their parts.

That electrical phenomena depend on some infinitely subtle and powerful agency pervading the material world, is in the present state of natural knowledge almost certain. That this agency, although differing essentially from any form of matter of which we have the least cognizance, is itself material, is also highly probable. Admitting this first principle, two theories of electricity, of modern date, principally claim attention.

32. The first of these originated in the discoveries of Du Fay and Symmer (14). This theory supposes electricity to be an infinitely attenuated fluid pervading the most compact bodies, being compounded of two more primary elements possessing



distinct and opposite properties. These, which are called *vitreous* and *resinous* electricities, are also assumed to be highly elastic fluids of perfect fluidity and elasticity, each repulsive of its own particles, but attractive of the particles of the other; so that when combined in virtue of this attraction, they completely condense or neutralize each other, and electrical repose or quiescence is the result. When, however, a disunion of these elements takes place, each becomes active, and hence the source of the phenomena of electrical excitation, which consists in a separation and abstraction of one of the elements, leaving the other in excess or uncompensated. The direct consequence of this is that the elementary electricity in excess, whichever it be, will tend to disturb the neutral electricity of any substance near it by operating upon the opposite elementary electricity which it contains; so that the balance of the two electricities in combination in the neutral body is overset, its electricity being separated, as it were, into its constituent elements. This constitutes *induction* (19). Now *attraction* is an immediate consequence of this induction, since the opposite electricities tend to unite, so that the bodies, if free to move, approach each other (22). When the opposite electricities are already in excess in two substances, the previous induction just mentioned is not requisite, and is hence in this case very limited. The bodies tend towards each other as before. *Repulsion* is supposed to arise from the excess of either of the elements in both the repellent bodies, the particles of which, by the hypothesis, repulse each other.

This theory does not recognize any peculiar affinity or attractive force between either of the electricities and common matter: it assumes, however, that there is an intimate association of the quiescent or compound element termed *electricity* with the particles of common matter, and that this electricity is indefinitely attenuated and imponderable. When resolved into its constituent elements, and a portion of one of them abstracted, then the element in excess is found by the repulsion of its particles to exist in a thin stratum on the surface of the

body where it is retained, within, as it were, a hollow vase of air, by the pressure of the atmosphere.

33. The second view of electrical action, differing essentially from the former, originated at the same time with Dr. Franklin in America, and with Dr. Watson in England, about the year 1747. It has been with justice called more especially the *Franklinian Hypothesis*, from the beautiful and extensive researches of this celebrated Philosopher, and the admirable way in which he brought them to bear on the generic phenomena of electricity. This theory assumes the existence of a single elementary homogeneous fluid of extreme tenuity and elasticity, without weight, existing in a state of equable distribution throughout the material world; and is assumed to be repulsive of its own particles, but attractive of all other matter. When distributed in bodies in quantities proportionate to their capacities or attraction for it, such bodies are said to be in their natural state. There is in this case a sort of equilibrium of distribution, and the result is electrical repose. When, however, we increase or diminish the quantity in any substance, we disturb this equilibrium, and a powerful action ensues, arising out of the attractive force of the body to regain its natural share of the electric fluid, if its original quantity be diminished, or to throw it off upon other bodies, if increased.

34. According to this theory, *excitation* is the result of a change in the relative attractive forces of the rubber and electric for electricity, when brought into contact; in which case the attraction of the one for electricity is increased, and that of the other diminished. Still, however, whilst in contact, the equilibrium of repose is not apparently disturbed (6), since the two substances may be taken as one: when, however, we disunite them, the original attractions are restored whilst the new distribution remains, and the result is that one of the bodies becomes overcharged with the quantity abstracted from the other, and which is consequently minus or undercharged; that is to say, one is charged *positively*, the other *negatively*.

Electrical Induction is the result of the tendency of the electric fluid to an equilibrium of distribution. If a body be overcharged, it endeavours to throw off the superabundant quantity upon any other body near it, causing the electricity of the neutral body to recede to its more distant parts, so as to make, as it were, room for it. (19) (20.) If undercharged, or deficient of its natural share, then the common matter of the body attracts towards it the electric fluid in any other neutral body near it, causing a flow, as it were, of the natural electricity of the neutral body from its more distant parts (20); in either case disturbing the electrical distribution of the neutral body, and electrifying its opposite extremities,—the one positively, the other negatively (20).

*Attraction* is an immediate consequence of this influence of induction, from the tendency of the opposite or plus and minus states to unite and equalize the distribution; the attractive force of the matter of one of the opposed surfaces for electricity being increased, the other diminished. In fact, the one is a condition to supply the electricity required by the other.

*Electrical Repulsion* (13) is the result of the repulsion of the particles of electricity collected in the repelling bodies when these are overcharged, and the result of the attraction of the denser fluid in the medium surrounding the body when undercharged.

35. Both these theories, or rather hypotheses, explain readily many complicated phenomena of electricity, yet they must be admitted to be extremely defective in their general application to the mass of facts which they are required to elucidate: one of the great difficulties in the way of both is the phenomena of electrical divergence (15), since it may be shown, that if assumed to depend on a purely repulsive force existing in the molecules of any electric fluid, then it is a species of repulsion essentially different from any repulsive agency of which we have the least experience,—its action being at great distances, and operating between distinct and circumscribed collections

of repulsive matter disposed on the surfaces of bodies. But it may be shown on common mechanical principles, that no motion could possibly take place in this way. Some of the French philosophers, aware of this fact, have accordingly endeavoured to account for electrical divergence by a mechanical action on the surrounding air. The Franklinians, on the other hand, when they found it impossible to apply their principles to the divergence of *negatively* electrified bodies, that is to say, bodies from which the agency has been abstracted but upon which the repulsion is assumed to depend, resorted first to an attraction in the surrounding medium as the cause of the separation of such bodies. Many were in consequence led to explain every case of electrical divergence on this principle, and to deny the existence of a repulsive force in electricity altogether. As it became, however, difficult to support this view, they were finally led to assume that the repulsion in such cases arose from an actual repulsive force between the particles of common matter when deprived of their natural share of electricity; all of which, it must be allowed, are only so many plausible excuses for a defective theory.

These difficulties have been still further increased from the circumstance that very perfectly insulated bodies have by modern experiments been found capable of retaining their charge in an extremely rarefied medium, such as that produced by the most perfect air pumps, and that electrical divergence and attraction can be exhibited in such a medium, much in the same way as in a dense atmosphere.

36. Such are the two theories generally received in explanation of ordinary electrical phenomena. We have, however, to notice in addition to these, an hypothesis of electrical action by Mr. Henry Eales, a philosophical Irish gentleman of great intelligence, who wrote on this subject in the year 1771, and which, upon a fair review, may probably, with very slight amendment, be found far less embarrassed than either of the preceding theories.

Mr. Eales assumes, in common with other Philosophers, the

existence of an extremely subtle agency, termed *electricity* or the *electric fluid*, and, as appears by his letters to the Royal Society in 1757 and 1758, he was one of the first to discover the two elementary electricities of which the electric fluid is said to be composed. These two electricities, or electrical powers, as he terms them, consist of two distinct elastic media, which equally condense and attract each other, and are at the same time equally attracted by all other matter. These powers are supposed to be repulsive of their own particles, and to have each, when separated from the other, a power of expansion in all directions, the limit being determined by the attractive force of the matter of the electrified substance acting against the electrical force of expansion. In this way an electrical stratum is conceived to exist about any substance charged with electricity, held to its surface by the attractive force of the matter of which the substance consists. Excitation, as in the first theory, consists in a separation and abstraction of one of these elementary forces, leaving the other in excess. *Induction* is the result of the expansion of the electricity in excess towards any neutral body,—attracting the one element, repelling the other element of the electricity of the neutral body, and so disturbing its electrical state (20). Attraction is the consequence of this by the combination of the opposed electricities (22), which condense each other at a distance, and so drag the bodies together as it were through the medium of their respective electrical strata held by attraction to their surfaces. A similar attractive effect results from the expansions of the electricities of two substances oppositely electrified (22). If the two electrified expansions be similar, as in the case of similarly charged bodies, then the meeting of these at the point of contact will cause the bodies to separate up to a distance at which the repulsive action of the elementary electrical particles is balanced by the attractive force of the matter of the bodies for these particles. Such were the views of Mr. Eales; and it must be allowed that they involve less difficulty than either of the others. That two opposite elec-

trical forces or powers are called into play in every case of electrical action, is quite certain, from whatever source they may be conceived to be derived ; and it is hence only required to explain how these forces act in bringing about the perceived effects. Mr. Eales had certainly the merit of originating a doctrine of opposite electrical forces ; but being a prolix, and in some respects a rather unmethodical writer, his views have not met with that attention which they really merit.

37. In very recent periods of electrical research much light has been thrown on the nature of induction by the beautiful investigations of Dr. Faraday, whose extensive discoveries in this department of science have far surpassed in importance those of almost every other inquirer, both in past or modern times. This celebrated Philosopher conceives electrical induction to depend on a physical action between contiguous particles, and never taking place at a distance without operating through the molecules of intervening non-conducting matter. In these

intermediate particles a separation of the opposite electricities takes place, and they become disposed in an alternate series or succession of positive and negative points or poles : this he terms a polarization of the particles, and in this way the force is transferred to a distance. Thus, if in fig. 17, P represent a positively charged body, and *a b c d* intermediate particles of air or other non-conducting matter, then the action of P is transferred to a distant body N by the separation and electrical polarization of these particles, indicated by the series of black and white hemispheres. Now if the particles can maintain this state, then insulation obtains ; but if the forces communicate or discharge one into the other, then we have an equalization or combination of the respective and opposite electricities throughout the whole series, including P and N.

Fig. 17.



38. The theoretical views of Faraday, although connecting electrical forces with the particles of common matter, are especially confined to the general laws of the arrangement of these

forces. He does not stop to associate together by incomprehensible mechanical relations a subtle hypothetical electrical fluid and gross matter, but sets out at once with a positively existing condition of things. According to Faraday, the effects may depend upon a single electric fluid, as in the hypothesis of Franklin, or upon the association of two fluids, as in the theories of Symmer and Du Fay and Eales; or they may not depend upon any thing which can properly be called the electric fluid, but on certain affections or vibrations in the matter in which they appear. His theory is entirely confined to certain arrangements of the electrical forces as exemplified in induction, and showing how they are arranged. He assumes that *all particles* of matter are more or less conductors;—that not being in their quiescent state arranged in a polarized form (fig. 17), they become so by the influence of contiguous and charged particles,—they then assume a forced state, and tend to return by a powerful tension to their original normal position;—that being more or less conductors, the particles charge either bodily or by polarity;—that contiguous particles can communicate their forces more or less readily one to the other: when *less* readily, the polarized state rises higher, and *insulation* is the result; when more readily, *conduction* is the consequence. Hence conductors and insulators are bodies whose particles have naturally more or less power to communicate the electrical forces, just as they possess other natural properties.

Induction of the ordinary kind is the action of a charged body upon insulating matter, or matter the particles of which communicate the electrical forces to each other in an extremely minute degree,—the charged body producing in it an equal amount of the opposite force; and this it does by polarizing the particles (fig. 17). Such are the theoretical results of Faraday's elaborate investigations; and it must be allowed that he has placed them, by a most conclusive sequence of experiments, carried out with admirable skill and profound reasoning, far within the limits of mere conjecture.

### III.

Electroscopes—Condenser—Electrical Machines—Hydro-electric Machine—Electrophorus—Theoretical views of the action of Electrical Machines—The Electrical Jar or Leyden Phial—Theoretical views of the Electric Jar—Electric Batteries—Electrometers—Quadrant Electrometers—Torsion Electrometer—The Bifilar Balance—The Scale-beam Electrometer—The Discharging Electrometer—The Unit Measure—The Thermo-Electrometer.

39. Instruments for indicating the presence and quality or kind of electricity are termed *Electroscopes* and *Condensers*.—Electroscopes consist of any delicately suspended light body, capable of yielding to the smallest degree of force. Downy feathers, reeds, small balls of the pith of elder, fragments of cotton wool, &c., when suspended by fine filaments of silk or cotton from any convenient support, such as that shown in fig. 16, constitute the readiest and most simple form of this instrument. The metallic leaf already described (13), attached to a slip of paper and suspended in any convenient way, is perhaps the best electroscope of this form: when constructed of leaf gold or silver, and shielded by plates of glass, it is sensible to the least possible force. Of the more elaborate kinds of electroscope, the following are worthy of attention.

*The balanced needle.*—This form of electroscope may be advantageously constructed as follows: upon a short bent

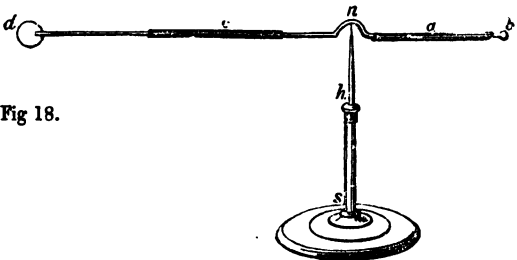


Fig 18.

brass wire *a n c*, fig. 18, are placed two light reeds *a b* and



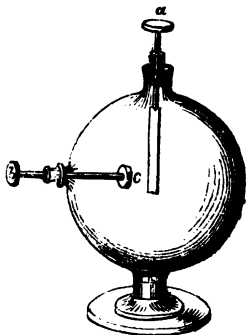
$c d$ , so as to form arms of unequal length: the long arm  $c d$  carries a light disc  $d$  of stiff gold paper, of about  $\frac{1}{2}$  an inch in diameter; the short arm  $a b$  carries a balance weight  $b$ , which may be a small shot or a ball of melted sealing-wax. The whole is delicately poised by a central point  $n$  upon a short piece of brass rod  $h$ , supported on a varnished rod of glass  $h s$ . The balance may be readily adjusted by sliding the reeds  $a b$  and  $c d$  upon the bent brass wire, so as to elongate or contract the opposite arms.

When this needle is used to detect the mere presence of electricity by its attractive force, a metallic thread is hung from the brass rod  $h$  so as to uninsulate the needle, and the excited or other electrical substance presented to the disc  $d$ . If we require to determine the kind of electricity, we withdraw the metallic thread from  $h$ , insulate the needle, and electrify the disc  $d$  either positively or negatively: on again presenting the electrical substance to the disc  $d$ , it will be either attracted or repelled, according as the electricity is of the same or opposite kind to that with which the disc is charged (15).

*Single gold leaf electroscope.*—An instrument of this kind was first described by Dr. Hare, of Philadelphia. It is best constructed in the following way. A slip of gold leaf, about 3 inches in length and  $\frac{3}{10}$ ths of an inch wide, is suspended from the extremity of a small brass rod, within a glass cylinder or sphere, as in the annexed

Fig. 19.

figure 19. Immediately opposite the lower end of the leaf a similar small rod of brass passes through the side of the sphere, carrying a gilded disc of wood or paper  $c$ , of about half an inch in diameter: both these slender brass rods terminate in small plates of brass or gilded wood,  $a$  and  $b$ , also about half an inch in diameter. The metallic rod at  $a$  slides through fine pieces of cork, within a short var-



nished tube of glass, supported by a wooden cap, or otherwise a piece of cork placed in the straight neck of the glass. The second rod  $b c$  slides through fine cork in a similar way, either inserted in the opening originally formed with a neck in the side of the sphere, or in a cap of wood, cemented about the hole drilled through the glass. The whole is sustained on a convenient foot-piece. A common spherical lamp-glass with a hole drilled in its side, or a small chemical receiver, in which an opening in the side with a neck is already present, will be found convenient substitutes for a glass receiver made expressly for the purpose, as represented in the figure.

If we require to detect by means of this electroscope, which is to the last degree sensible, the presence of electricity, a thread of metal is hung on the rod  $b$ , and the electrical body brought into contact with the disc  $a$ . When the distance between the leaf and the disc  $c$  is made very small, the most minute force of attraction is apparent.

If we require to determine the *kind* of electricity, we slide the wire of the disc  $c$  so as nearly to touch the leaf, and then electrify gently the leaf or disc  $c$ , either positively or negatively, with moderately excited glass or wax. The leaf will then be repelled, and stand off from the disc (13). Under these circumstances, if we present the electrical body to one of the plates  $a$  or  $b$ , the leaf will either diverge more freely, or collapse towards the disc  $c$ , according as the electrical state of the substance under examination is of the same or of an opposite kind to that with which the leaf has been previously charged (15). In the use of this instrument, care must be taken to preserve a dry atmosphere within the glass receiver, which may be, if required, removed from its foot, and held for an instant over a warm iron. The glass also, about the openings through which the brass rods pass, should be carefully varnished. With proper care, it is quite astonishing how sensitive the instrument becomes to the smallest electrical force.

*Electroscopes of Divergence.*—Two small balls of the pith of elder, attached to a thread of silk or cotton, and hung

over any appropriate insulating support, as in fig. 20, constitute an electroscope of divergence. If the balls be electrified either positively or negatively, they diverge (15); and any substance in an oppositely electrical state will, if brought near them, cause them to collapse more or less (15), whilst a substance in the same electrical state increases the divergence. These have been termed 'Canton's Balls,' from their being first employed by that skilful Electrician. The Earl of Stanhope constructed an electroscope of divergence by suspending two delicate reeds, terminating in pith balls, quite parallel to each other, as shown in fig. 21. This electroscope is more sensitive from the parallelism of the reeds, since the divergence will be more sensible when the legs of the electroscope hang parallel than when their upper extremities are in contact.

Mr. Cavallo obtained a delicate suspension by hanging the balls on short pieces of silver wire, formed into a loop at the upper ends, and moving on a ring of the same wire, as a point of suspension: this method gives extreme freedom of motion.

*Bennett's gold leaf electroscope.*—This is one of the most perfect and beautiful of this class of instrument. It consists of two slips of gold leaf, attached to short slips of paper, and suspended parallel to each other within a glass receiver by means of a rod of brass passing through a varnished glass tube, insulated in its neck, as represented in fig. 22. The end of the

Fig. 20.

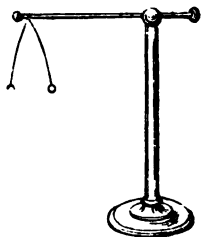
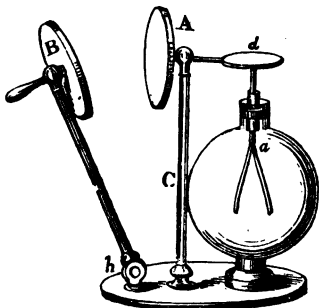


Fig. 21.



Fig. 22.



brass rod *a* is split open, and formed into forceps for holding the paper slips to which the leaves are attached, whilst the opposite extremity carries a small disc of metal or of gilt wood *d*, as in the single leaf electroscope, fig. 19.

When we require to ascertain either the presence or kind of electricity, the electrified substance is brought in contact with the plate at *d*. The leaves then separate, as shown in the figure; the separation being permanent, at least for a short period of time. The kind of electricity is ascertained by presenting a gently excited rod of glass or sealing-wax to the cap *d*; the leaves will then either collapse or have the divergence increased, according as the electricity with which they have been previously charged is of the same or of the opposite kind to that of either of these excited bodies. It may be requisite to observe, that the sensibility of this instrument is such as to require extreme care in the progress of such experiments. We should only operate with very gentle electrical forces, just sufficient to produce a decisive effect on the leaves of the instrument, and no more. Such is its sensibility, that a slight flap with a silk handkerchief on the plate *d* will render the leaves divergent. The Rev. A. Bennett, the inventor, found it powerfully affected by the mere projection of powdered chalk upon the plate from a common brush, or by wind blown upon it from a common bellows. It may be employed with advantage for Experiments 9 and 11.

The slips of gold are easily managed in their application to this instrument, if the proper means be resorted to, otherwise the process of applying or replacing them is tedious and difficult. The gold leaf should be laid on a leather cushion, and handled with a flat clean palate knife, made very dry. The slip is cut, or rather divided, by the edge of this knife drawn with pressure over it parallel to one of the sides of the leaf; a small short slip of gilt paper, gently moistened with the lip at one end, is then applied to the gold slip, by which it may be readily raised off the cushion. In placing the two leaves in the instrument, it is desirable to separate them a little by a

very thin slip of gilt cork, so as to allow of their hanging parallel and free, without absolutely touching. The sensibility of the instrument is by this means increased, and the leaves are not liable to adhere, which otherwise frequently happens.\*

All these electroscopes will be found extremely useful in electrical inquiries, and may be applied in all the preceding experiments as the ingenuity and judgment of the student may suggest.

40. *The electrical condenser.*—The delicacy of the instrument last described may be greatly increased by the contrivance attached to fig. 22, which originated with the celebrated Professor Volta, of Pavia, and is an application of the principle of approximated surfaces (20). We may infer from Exp. 15 (21) that an electrified conductor has a certain portion of the charge masked or neutralized when brought near a similar neutral body in conducting communication with the earth. Under these circumstances, the effect of the unopposed surface upon the electroscope may be not only very greatly diminished, but may be even reduced to nothing, or nearly so. In order, therefore, to affect the electroscope as before, an additional quantity of electricity is required, in consequence of the exclusive or compensating action between the opposed surfaces. Imagine, then, this additional quantity thrown upon the already charged conductor A, fig. 7, page 14, and the electroscope attracted in the same way as before. It is evident that if we now, under these circumstances, remove the opposed compensating surface, that is to say, the body B, fig. 7, we shall immediately set free the previous quantity which it held in abeyance, and the electroscope will become attracted with the united forces of the whole, and so evince an increased effect. The near approximation of such a surface therefore enables an insulated conductor to absorb or receive

\* The management of leaf gold and other metallic leaf being of importance to the Electrician, he should provide himself with a hard leather cushion, a palate knife, and wooden forceps, such as used by the gold beaters. The gold leaf employed should be firm, and of the best quality.

a greater quantity of electricity under the same amount of activity, as shown by an electroscope, than it could otherwise support. In other words, it acquires an increased electrical capacity, which is in fact the principle of *Volta's Condenser*. This instrument consists of two metallic discs, A and B, fig. 22, the former being insulated by the glass rod c, and connected with the electroscope, while the latter is supported on a brass rod, moveable on the hinge h, so as to admit of being brought very near the disc A, or withdrawn therefrom. These discs being placed nearly in contact, and any insensibly electrified substance applied to the disc A, the opposite disc B is turned back, as in fig. 22, when the electroscope immediately diverges with the electricity absorbed by the condenser A during the time B was opposed to it, but afterwards rendered free when B is removed.

An efficient and powerful condenser of a simple kind may be formed by placing a circular disc of wood, covered with tin-foil or metallic leaf, very near a smooth table, which is easily done by allowing it to rest on three small fragments of common window glass coated over with sealing-wax, or otherwise varnished. This disc may be about a foot or more in diameter, and  $\frac{1}{4}$  inch or a little more in thickness; it should be furnished with an insulating handle, D, fig. 26, page 52. If it be now required to examine any substance in so low an electrical state as not to affect a delicate electroscope, we bring it into contact with the disc, then remove the disc by its insulating handle, and apply it to the electroscope, which immediately diverges, and by which both the presence and quality of a very minute quantity of electricity may be ascertained.

If, instead of transferring the plate immediately to the electroscope, we apply it to a second smaller condensing plate, as A, fig. 22, having B opposed to it, and repeat this operation, after successive contacts with the substance under examination it is quite astonishing how small a quantity of electricity may be detected on the removal of the disc B.

This process of multiplying the effects of very small forces has been extensively employed by Electricians in very refined electrical inquiries, and has given rise to instruments termed 'multipliers' and 'doublers,' which, although exhibiting great skill and ingenuity in construction, are still liable to considerable objection, being of themselves, from their extreme delicacy, liable to induce a low state of excitation, and hence manifest equivocal results.

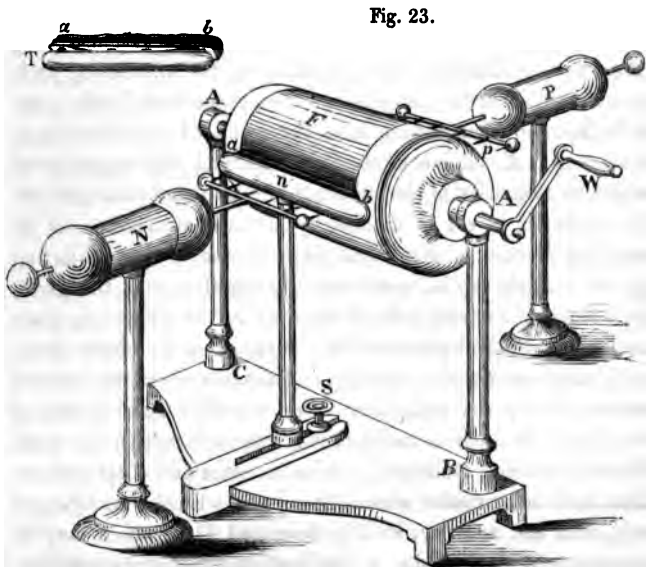
#### INSTRUMENTS FOR EXCITING AND COLLECTING ELECTRICITY.

41. Any instrument for the excitation and collection of electricity is termed an *electrical machine*. Those machines in which the excitation is produced by friction consist, 1st, of some electric to be excited; 2ndly, a rubber or cushion by which the excitation is effected; 3rdly, an insulated conductor for collecting the excited electricity. Thus the glass tube, fig. 2, page 5, with its wire and ball excited by silk held in the hand, is in fact an electrical machine,—the tube being the electric, the hand and silk the rubber, and the attached metallic wire and ball the insulated conductor for collecting the excited electricity. If the tube be about an inch and a half in diameter and about 2 feet in length, and be rubbed with the rough side of black oiled silk, smeared over with an amalgam of tin, zinc, and mercury (28), the excitation is by no means inconsiderable; strong electrical sparks will be thrown off from the conducting wire and ball, and other powerful effects will be obtained. In the early periods of this branch of science, glass tubes, brimstone, amber, and other electrics, subjected to friction by means of the hand, were commonly resorted to as an easy means of generating electricity by friction; but as further advances were made, and the vast importance of the subject became apparent, a more complete and powerful kind of apparatus was naturally sought for. The first attempts of this kind consisted in the revolution of globes of glass, aided by some kind of *mechanism*. These were made to turn round against fixed

cushions, the generated electricity being collected upon insulated conductors. We are indebted to a philosophical Burgomaster of Magdeburg, the ingenious Otto Guericke, for the first idea of a machine of this kind. Having mounted a globe of brimstone on an axis, he caused it to revolve against the hand, and thereby obtained a rapid and powerful excitation.

A great variety of electrics have been treated in this way, giving rise to machines of different forms and construction, some of them of a very complicated and unwieldy kind. All these, however, have become finally resolved into one or two forms of apparatus now in use, in which the electric to be excited is either a hollow cylinder or a circular plate of glass. We shall, for the sake of perspicuity, confine ourselves to a description of a few of the most perfect of this class.

42. The annexed fig. 23 represents a cylindrical electrical



machine of an improved kind, in which A A is a hollow glass



cylinder, having wide open ends fitted with caps of wood. These caps have projecting pivots, which serve as the extremities of a horizontal axis, and turn in pivot-holes drilled in spherical pieces of wood or metal,  $\Delta A$ , cemented on vertical glass pillars,  $\Delta B$ ,  $\Delta C$ , the whole being fixed on a firm wooden basis,  $c B$ . A flat cushion, or rubber of coarse oiled silk  $r$ , fig. 23, stuffed with wool or hair, and backed by a wooden cylinder  $n$ , covered with tin-foil, is fixed at  $ab$  on a vertical glass pillar  $s n$ , attached by a sliding piece  $s$ , to the base  $c B$ , by which it is caused to bear with greater or less force against the glass cylinder. There is a screw at  $s$ , for confining the slide in any given position. The rubber  $ab$  is supported loosely against the wooden cylinder by two brass pins, and is furnished with a long flap of thin silk  $f$ , oiled on one side only, sewed to its outer edge, which rests upon and passes over the upper surface of the cylinder, the rough side being next the glass. Immediately opposite the termination of this silk flap there is a small cross arm of brass  $p$ , carrying three or four short points for catching up the vitreous or positive electricity from the glass as it emerges from beneath the silk flap, and transmitting it to the insulated cylindrical conductor  $p$ . A similar cross arm transmits the negative or resinous electricity of the rubber to the insulated conductor  $n$ : the cylinder is set in motion by a winch  $w$ , having an insulating handle of glass, and fixed to one of the projecting pivots. Each cap of wood consists, together with the pivot, of a single piece, and is fitted securely in the projecting glass necks by attached pieces of fine cork, so as to admit of the cap being occasionally removed, and the interior of the cylinder rendered very dry and clean. For a perfect development of the power of such a machine, it is requisite to free the glass from all dirt and moisture, both on its inner and outer surface. The cushion or rubber should be covered with the amalgam of tin, zinc, and mercury, already described (28), which may be spread on it by the aid of a little lard, and the surrounding air should be rendered as dry as possible.

43. Under these favourable conditions we obtain the following most interesting phenomena. Directly we put the glass cylinder in motion, 1st, on removing the conductors P and N, brilliant scintillations and lines of light will dart round the under surface of the cylinder, between the termination of the silk flap and the rubber; brushes of light and luminous coruscations will frequently appear to fly off into the air from under the flap, presenting in the dark an extremely beautiful appearance. If a pointed metallic rod be held in the air in front of the silk flap, a bright star of light will appear to settle on it, even at a considerable distance from the glass. 2ndly, when the conductors are placed as shown in the figures, and the knuckle of each hand presented, one to the conductor P, and the other to the conductor N, a current of bright and powerful sparks will pass between the knuckles and each of the conductors; and if we observe at this time the points on the cross arms, those on the positive conductor P will appear as simple luminous points or stars, whilst those on the conductor N will be divergent, and present to the eye the appearance of a luminous brush. If we suspend a chain of small metallic beads, strung together on a silk thread, between two metallic rods fixed on the conductors P N, it will appear as a sort of brilliant necklace of luminous beads, the effect of which is frequently dazzling to the eye.

When we require to collect vitreous or positive electricity, we present the substance to be charged or electrified to the positive conductor P, and connect the negative conductor N, or otherwise the cushion itself, with the earth, by which electricity becomes continually supplied to the revolving glass cylinder. In this case the large conductor N may be removed. When we require negative or resinous electricity, we reverse this. The body to be charged is presented to the negative conductor N, and the positive conductor P connected with the earth, so as to relieve the surface of the glass cylinder of the electricity thrown on it, and enable it to develop a continual excitation in the rubber, and conse-

quently a constant supply of negative electricity to the conductor *N*.

44. It has been usual in constructing these machines to form the end of the glass cylinders into narrow necks, and close them up; but it is far preferable to have them wide and capacious, so as to admit of drying and wiping out the interior. The condensation of moisture on the interior of the surface is fatal to the action of the machine, from the conduction it affords to the return of the induced electricity over the interior of the glass to the cushion. In fact, if the conductors *P* and *N* be joined by a curved wire, all the phenomena we have just described will vanish. It was probably from this circumstance that some of the early Electricians found their glass globes and cylinders improved in power by coating the interior with some resinous composition, by which the non-conduction of the vitreous surface was rendered more perfect, and moisture less liable to condense on it. Electrical machines of this construction are made with glass cylinders varying from 4 inches to a foot or even to 20 inches in diameter, and from 6 to 18 inches in length.

45. *The plate-glass electrical machine.*—About the year 1776, a celebrated German, Van Marum, was led to employ a circular disc of shell lac as being a convenient form of electric for excitation. This was soon followed by a disc of plate-glass. In a similar way, Ingenhouz, Van Marum, Ramsden, and Cuthbertson, appear to have been the originators of electrical machines of this kind. The machine, as constructed by Cuthbertson, a celebrated philosophical instrument maker, consists of a circular disc of plate-glass, from 2 to 7 feet in diameter, mounted on a horizontal axis of metal, and sustained by a vertical frame of mahogany, to which are fixed two pairs of cushions or rubbers, one pair embracing the upper margin of the plate, and the other pair the lower margin. This plate is turned round by a winch fixed at the extremity of the axis. The electricity is confined by flaps of oiled silk, extending from each pair of cushions round one-fourth nearly of the circum-

ference of the plate, and is then collected by two rows of points opposite its horizontal diameter, and communicating with an insulated metallic conductor.

In this arrangement we have only the vitreous or positive electricity of the plate. To obtain the resinous or negative electricity of the cushions, it is requisite to connect them by a slip of metal let into the frame-work, and to place the whole machine on pillars of glass. In this case the conductor, which was before insulated, must communicate with the ground.

This machine is prepared for excitation as already described (42). Its action is very intense, and the same phenomena as observed with the glass cylinder are observable on turning the plate,—the effects being doubled by the addition of another set of cushions, although it has been doubted whether the friction on *both* sides of the plate is attended by any greater effect than a well-applied friction on one side only. This form of electrical machine is undoubtedly extremely powerful, and well adapted to extensive researches in electricity; but for ordinary philosophical purposes, the cylindrical machine, from the simplicity of its construction and use, is probably more convenient.

46. Van Marum, about the year 1785, constructed, with the aid of Mr. Cuthbertson, an electrical machine on this principle, of enormous power, and which was afterwards placed in Teyler's Museum at Haarlem. It consisted of two circular plates of French glass, each 65 inches in diameter, fixed upon the same axis, and excited by four pairs of cushions, nearly 16 inches in length. The conductor was supported on three pillars of glass, sending out collecting branches between the plates. Two men, and sometimes four, were employed to turn the plates. When in full force, a single spark from the conductor was found to melt a leaf of gold; a thread became attracted at a distance of 38 feet; and a pointed wire exhibited the appearance of a luminous star at a distance of 28 feet from the conductor. Persons within 10 feet of the plates experienced a

sort of creeping sensation over them, as if surrounded by a spider's web.

47. The collection of negative electricity from machines under this kind of arrangement being attended with inconvenience, several attempts have been made to vary it. Van Marum was hence led to place a single plate at the extremity of a strong horizontal axis; and having insulated the cushion or glass pillars on each side of its horizontal axis, collected either positive or negative electricity by a double-branched wire in connection with the conductor, which could be turned either upon the rubber or the glass,—whilst another wire behind the plate could be placed in a similar way, so as either to supply the rubber, or discharge the electricity excited in the glass.

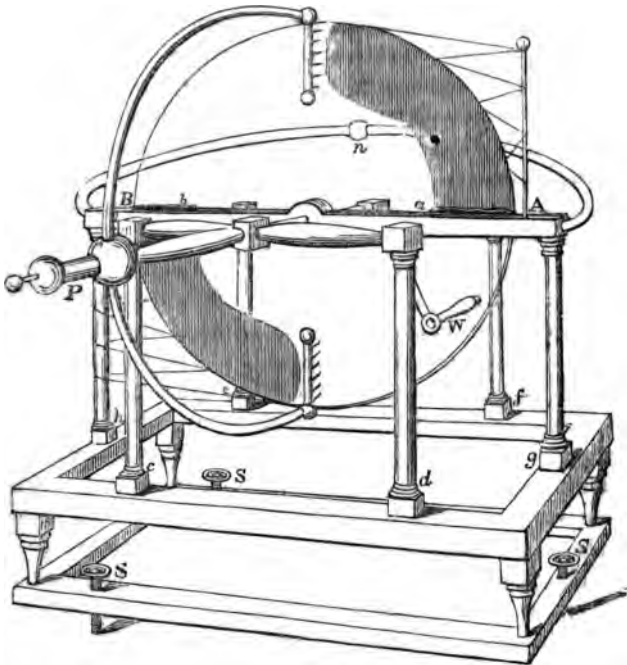
In other arrangements a single pair of cushions only is employed, with a large conductor immediately opposite. The great plate used at the Royal Polytechnic Institution in Regent Street, London, is mounted in this way. It is 7 feet in diameter, is turned by a small steam engine, and is extremely powerful.

48. The following arrangement, fig. 24, will be found effective, and calculated to meet every difficulty in the use of the glass plate for positive and negative electricity. In this construction the plate is mounted on a metallic axis, supported between two transverse bars of mahogany, and with four vertical mahogany pillars, *c d e f*, as represented in the figure. These pillars are inserted into a strong rectangular frame below, and two before and two behind the plate, forming a firm and secure basis. The rubbers *A a* and *B b* are placed on each side, and insulated on stout pillars of glass, *A g* and *B h*, also fixed to the hollow rectangular frame. A metallic conductor *P*, with two branches, is sustained in a vertical position in front of the frame, on a strong support of glass, whilst a curvilinear brass tube, *A n B*, passing behind the plate, connects the cushions, and forms the negative conductor.

The plate is turned round by an insulating winch *w*, constructed of a strong cylindrical bar of glass, and the whole is

supported by four legs on a second rectangular frame, having three levelling screws, *s s s*, fixed in it, so as to render the

Fig. 24.



axis level and the machine secure on the ground. This machine is employed in the same way as that already described (42). With a plate of about 2 or 3 feet in diameter, very extraordinary power is obtained.

In all machines of this construction it is requisite to attach cords of silk to the flaps, and pass them round fixed supports, in order to prevent the flaps from being dragged over the plate in turning it round.

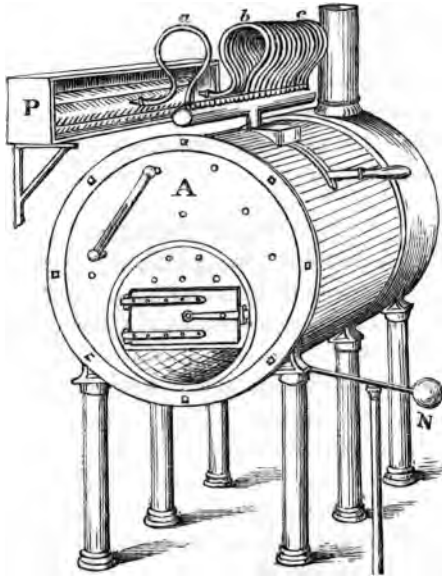
49. It may not be amiss to observe that the selection of plate-glass for these machines is a point of the greatest conse-

quence. A high degree of polish and a perfect manufacture being essential, the French plate of the old kind appears to be the most powerful. There is little doubt but that a common vitrified surface, such as that of the best crown glass, is very desirable, and better qualified for electrical excitation: hence plates of common window glass, joined by an intervening cement of black sealing-wax, will be found well adapted to the purposes of electrical excitation. A plate machine constructed in this way, with a plate of about 2 feet in diameter, had very remarkable power.

50. *Hydro-electric machine.*—This species of electrical machine is of very recent date and construction, and is the result of an accidental discovery in 1840, by an observing workman in charge of a fixed steam engine at Sighill, near Newcastle. Owing to a leak in the cement about the safety-valve, there arose a considerable escape of steam. The engine-man, being about to adjust the weight of the valve, was surprised by the emission of a powerful spark of electricity, which he found always to arise from the metal-work connected with the boiler, and from the boiler also, if he attempted to touch it while steam was continuing to escape, especially when one of his hands was immersed in the vapour. Mr. Armstrong, a scientific gentleman at Newcastle, having been informed of this result, lost no time in investigating the phenomenon. By an insulated brass rod, having a metallic plate at one extremity and a ball at the other, he obtained 60 or 70 sparks per minute on bringing the ball near the boiler, whilst the plate was immersed in the issuing vapour, and after a series of highly interesting inquiries, succeeded in producing a vaporizing electrical machine, depending on the excitation of particles of water driven by steam through small orifices. This machine consists of a steam boiler A, fig. 25, insulated on stout pillars of glass. The steam is caused to issue from a general steam pipe through bent iron tubes *a b c*, terminating in jets of wood, and of which there are a great number: an insulated projecting conductor N is placed in connection with

the boiler, for the convenience of collecting the excited electri-

Fig. 25.



city; and a second conductor **P**, formed of a metallic case, furnished with several rows of points, is placed immediately in front of the jets, to receive and carry off the opposite electricity of the steam, and prevent its return upon the boiler, by which the excited forces would be neutralized. Faraday, who also investigated this question with his accustomed tact and penetration, has shown, by a series of masterly experiments, that the electricity thus produced does not depend on the mere issue of steam through small orifices, or upon any chemical or other change which may be supposed to arise from evaporation or condensation, but is the result of the friction of condensed particles of water whilst being driven by the still issuing steam through the jets; so that in fact these watery particles perform the office of the glass plate of the common



machine, and give out vitreous electricity. The wood jets and pipes act as the rubber, and give out resinous electricity, whilst the steam itself is the mere mechanical means by which the friction is effected.

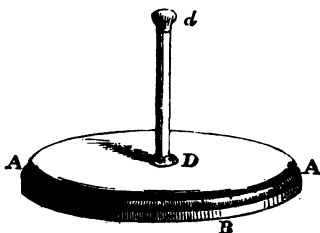
The electricity produced by this apparatus is enormous in quantity. The sparks from the conductor *N*, upon an uninsulated metallic ball, are dense and rapid, presenting frequently the appearance of a continuous flame, and will readily set fire to inflammable matter.

#### THE ELECTROPHORUS.

51. We are indebted to the ingenuity of Professor Volta for this peculiar species of electrical machine, based on excitation by friction, but the immediate action of which depends on the principle of induction (20).

The arrangement consists of a circular or other electrical plate *A A*, fig. 26, usually of shell lac. This plate is placed on a conducting disc *B*, of about the same size, and termed the sole; a second circular conducting disc *D*, termed the cover, and furnished with an insulated handle *D d*, is placed upon the upper surface of the electrical plate. When this instrument is employed for the production of electricity, we first remove the cover *D* by its insulating handle, and then excite the electrical plate *A* by flapping it briskly with dry silk or with some dry fur-skin; we then replace the cover *D* as before: under these circumstances the cover, being insulated, does not take up the electricity of the plate, but is acted upon inductively (20), that is to say, its neutral condition is changed, and the parts distant from the under surface placed in the same electrical state as the surface of the excited plate: hence these parts are in a condition to

Fig. 26.



charge any substance brought near them. We find accordingly, that on bringing a conductor near the cover, an electrical spark is elicited, which tends to neutralize the induced state of its parts distant from the surface in approximation with the electrical plate beneath; and if such conductor be insulated, it will become charged with the same electricity; that is to say, if the electrical plate be negatively excited, the distant parts of the cover *D* will be negative by induction (20), and any conductor brought near it will yield up positive electricity, and remain also negatively electrified. If we now remove the cover *D* by its insulating handle, we immediately restore its original capacity or conditions of neutrality, whilst the positive electricity it had acquired under induction remains. Hence it is now charged with positive or vitreous electricity, and will electrify any insulated conductor positively when brought near it.

It is quite evident that in this machine the excited electric loses nothing of its electricity by all this, the excitation being merely subservient to induction. Hence, after restoring the neutrality of the cover by relieving it of its charge, we may again repeat the operation, and carry off in this way successive charges to an almost unlimited extent without any new excitation of the electric plate: it is hence a sort of permanently excited electrical machine.

The electrical plate of this instrument, as constructed by Volta, consisted of shell lac, common resin, and Venice turpentine, in equal parts. The sole may consist of a circular plate of wood, covered with tin-foil. The cover, or superior conductor *D*, may be also a light disc of wood, covered with some stout metallic leaf. Small plates of glass, sealing-wax, or brimstone, are admirably adapted to experimental purposes, when mounted as electrophorus plates, with the attendant conductors. The machine itself has been termed *electrophorus*, from the circumstance of its prime conductor, the cover, *bearing off* as it were by repeated removals definite quantities of electricity which it had absorbed under the influence of the excited electrical plate.

**ELECTRICAL MACHINES NOT DEPENDENT ON FRICTION.**

52. Although electrical excitation by friction has been generally employed in the construction of ordinary electrical machines, yet other sources of electricity are open to us. Thus the dry electrical pile (25), fig. 12, and the voltaic series (25), fig. 13, may each be considered as electrical machines, although of very limited power for common electrical purposes. The electrical disturbance also induced in a coil of wire by magnetism (25), fig. 14, may be employed as a source of electricity, and has given rise to the magneto-electrical apparatus already alluded to (25). Such instruments, although admirably adapted to some peculiar experimental researches, are quite unavailable in the accumulation of great statical electrical power.

**THEORETICAL VIEWS OF THE ACTION OF ELECTRICAL MACHINES.**

53. The production of electricity by the ordinary electrical machine is explicable upon either the double or single fluid hypotheses (32) (33). According to the first of these (32), we disunite by friction the combined electricities existing in the rubber and glass, and change, according to the improved theory (37), the relative attractive power of these substances for the separated elements, so that the positive force becomes collected on the glass and the negative force on the rubber; but since there is only a certain quantity of electricity in the glass and rubber, the action must cease after a few revolutions of the glass, unless more electricity be furnished for decomposition; and such is found to be the case: very little effect is produced so long as the positive and negative conductors P N, fig. 23, remain insulated. Directly, however, we connect one of these conductors with the ground, the other will be in a position to charge any insulated substance presented to it up to a point of saturation; for the neutral state of the one conductor becomes restored at the expense of the opposite electricity drawn from the mass of the earth, and that of the other

at the expense of the opposite electricity of the insulated body : hence both the mass of the earth and the insulated body may be considered as being electrified by this operation—the one positively, the other negatively, according to the particular conductor, P or N, with which they are associated. But the electricity of the earth being indefinitely great, the result is only sensible on the lesser insulated body, and the action is limited to the quantity of positive or negative electricity it can furnish.

If, instead of connecting one of the conductors with the earth, we associate it with another insulated body, then the effect is sensible in either case, and the action is limited to the quantity of positive or negative electricity which these insulated substances can yield up : such is, in fact, the condition of the positive and negative conductors P N, which yield up negative and positive electricity to the glass and rubber, directly the machine is put in motion, and by which the one, P, becomes charged positively, and the other, N, negatively. If both these conductors therefore be connected with the ground, the result is a continual separation and re-composition of the two forces, and hence a neutral state of the conductors P N. Upon this theory, therefore, the action of the electrical machine consists in a separation and abstraction of one of the electrical elements, leaving the other in excess.

54. The electrical machine, according to the hypothesis of a single elementary fluid, is only a means of changing the quantity of electricity in bodies. By the revolution of the glass against the rubber we continually renew and break the contact between these substances, the consequence of which is (6) an accumulation of electricity on the glass, and an abstraction from the rubber. Hence the insulated conductors P and N become charged, one positively or plus, the other negatively or minus. On this, however, as on the former hypothesis, the action would soon cease if one or both of the conductors remained insulated, since the rubber has only a limited quantity of electricity to yield up, and the glass therefore only a certain

quantity to gain: reciprocally, the glass cannot take up more than a given quantity of electricity, consequently the rubber can only lose a limited quantity. Hence the excitation is confined within these limits. Now imagine two insulated conductors brought near the conductors  $P N$ ; one near the positive conductor  $P$ , the other near the negative conductor  $N$ . The result would be an abstraction of electricity from the body opposed to the negative conductor, which would go to restore the equilibrium in the rubber, and an addition of electricity to the body near the positive conductor, which would relieve or neutralize the electricity of the glass; and thus the action of the machine would be to charge or electrify both these bodies at the expense of each other, the one positively, the other negatively. We have still, however, reached a limit depending on the amount of electricity which the respective bodies can yield up or receive. Such is in fact the condition of the conductors  $P N$ , one of which,  $P$ , receives the excited electricity on the glass, whilst the other,  $N$ , yields up electricity to the rubber. When, therefore, we connect one of these conductors with the earth, the action may become indefinitely great upon any insulated body applied to the other, and required to be electrified, however great its extent, the means of supply or absorption of electricity being without assignable limit. When both the conductors  $P N$  are connected with the mass of the earth, the result is a constant disturbance and renewal of electrical neutrality,—the rubber being supplied with electricity as fast it yields it up to the glass, and the glass yielding it up again to the earth as quickly as it abstracts it from the rubber. Upon this hypothesis, therefore, the operation of the electrical machine is that of a pump,—to withdraw electricity from some bodies and throw it upon others.

55. These two theories apply in nearly the same way to the action of the electrophorus. The cover  $D$ , fig. 26, whilst on the excited plate, acts upon bodies in the same way as the negative conductor  $N$ , supposing the plate to be a resinous electric; and after removal, as the positive conductor

P; that is to say, on the hypothesis of two electric fluids, it first attracts vitreous electricity in consequence of its distant parts being negative by induction (20), and hence will leave any body applied to it negatively charged; and then subsequently, on removal, it attracts resinous electricity from any body presented to it, in consequence of the charge of vitreous electricity it had previously taken up, and so electrifies such body,—if insulated, positively. On the Franklinian, or single fluid hypothesis, the cover first attracts and then gives out electricity; that is to say, it first abstracts electricity from some bodies, and then communicates it to others.

The induced action upon which this state of the cover depends is upon the hypothesis of two fluids derived from the excited electricity of the electrical plate. Supposing the electrical plate to be a resinous plate, the vitreous electricity of the cover is attracted towards it, leaving the resinous electricity repelled to its distant surface. On the Franklinian hypothesis, the electrical plate being deficient of electricity, or minus, the electricity of the cover is drawn towards it so as to equalize the distribution as nearly as possible, hence leaving its distant surface negative or minus.

If the electrical plate be excited with vitreous electricity, then the converse of all this happens, but the theoretical views remain the same.

56. Similar applications of these theories may be made to the electrical pile and to the coil of the magneto-electric machine (fig. 14, p. 23). We have upon the one theory a separation of the combined electricities, and upon the other an unequal distribution. In the electrical column and voltaic series, the opposition of the two metals changes their relative capacities for one of the electrical elements, or for electricity considered as a single elementary fluid, so that either positive or negative electricity exists in excess, or flows upon one of the metals. By the intervening semi-conducting fluid or other substance, this excess is carried on to the next pair of metals, and so throughout the whole series up to the terminating plates.

which become finally charged with different electricities, analogous to the positive and negative conductors of the electrical machine. We have, however, in such arrangements a sort of electromotion, together with a development of electricity by chemical action when the plates are excited by saline and other fluids.

In the coil of the magneto-electric machine a somewhat similar electric motion is produced by the influence of the magnet; and each end of the coil, at the instant of making or breaking contact with the magnet, is in opposite electrical states.

#### THE ELECTRICAL JAR, OR LEYDEN PHIAL.

(57.) The years 1745 and 1746 are remarkable for a scientific discovery of a very marvellous character. Some Dutch philosophers at Leyden finding electricity rapidly disappear from a simple insulated conductor, imagined that if it could be completely enclosed in solid electric matter, a charge might be retained on the conductor for any length of time. This idea, early in the year 1746, they endeavoured to realize. Water being a convenient conductor, they enclosed it in a small glass phial, and having inserted a nail through the cork, suspended it from the prime conductor of the electrical machine so as to convey electricity into the water through the nail. One of the experimenters, Mr. Cunæus, whilst continuing this inquiry, in endeavouring to detach the phial and nail from the electrical machine, received so violent a shock across his arms and breast that it shook his whole frame.

A similar result appears to have been obtained about the year 1745, by Von Kleist, dean of a cathedral in Germany, whilst engaged in an interesting course of experiments on the communication of electricity to glass, an account of which is found in the register of an Academy at Berlin. Having passed a stout brass pin or wire into a common phial containing a small quantity of mercury, electricity was thrown, by means of the electrical machine, upon the interior of the phial, *through the pin*. His account of the phenomena which en-

sued is both instructive and amusing. "As soon (says he) as this little glass with the pin is removed from the electrical machine, a flaming pencil issues from it so long, that I have been able to walk 60 paces in the room with this little burning machine; and if the finger or a piece of money be held against the electrified pin, the stroke coming out is so strong that both arms and shoulders are shaken thereby. Thin-necked glasses have been twice broken by the powerful shock."

Muschenbroek, who subsequently repeated these experiments at Leyden, with water in a thin glass bowl, says, he was so struck in his arms, shoulders, and breast, that he lost his breath, and was two days in recovering from the effects of the blow.

The discovery of such a terrible power in nature soon spread through Europe, and gave immense impulse and éclat to electrical investigations. Sir W. Watson, Smeaton, Bevis, Wilson and Canton, all distinguished members of the Royal Society of London, repeated and extended these experiments; indeed it is to them that we owe the final completion of the Leyden phial, under the present form of the electrical jar, and its consequent practical application to most important researches in Electricity. Sir W. Watson showed that the power of the phial did not depend on the density or quantity of conducting matter either within or without the glass, but that, *cæteris paribus*, it was as the extent of contact between conductors and the outer surface of the glass; on which principle he placed the phial in a small open cylinder of sheet lead. Smeaton now transformed this experiment into the covering of plates of glass with thin metal, leaving on each side an uncovered part, and he found that if after having communicated electricity to one of the surfaces of the plate, he touched both surfaces with his hands at the same time, all the effects of the Leyden or German experiment were obtained. Sir W. Watson, further enlightened by this result, now applied coverings or coatings of metallic leaf to the interior and exterior surfaces of glass jars, leaving a portion of the glass under the mouth of



the jar free, an arrangement both effective and convenient, and in use at this day. Thus the Leyden phial resolved itself into the electrical jar, and became a most important instrument of physical research.

58. The most perfect kind of electrical jar is shown in the annexed figure 27, in which the faintly-shadowed portion *A* marks the position of the exterior and interior coatings, usually tin leaf pasted against the glass: *b* is the uncoated or insulating portion of the jar; *d b* a light metallic rod or tube, terminating in a knob or metallic ball *d*, and passing to the bottom of the jar, where it is supported in a socket of wood, fixed to the coating expanded upon the glass. This is the charging rod, and care is taken to insure a good contact with the metal on the bottom of the jar.

Fig. 27.



When the jar is to be charged, the charging rod *d b* is connected with one of the conductors of the electrical machine, and the exterior coating *A* with the other, either directly or by communication through the ground; and with a view of observing the progress of the charge, a light reed, terminating in a small ball of pith, may be loosely attached to the charging rod by a very short piece of fine silk or cotton thread: this will constitute, with the rod, an electroscope of divergence (39), so that as the charge proceeds the reed will become raised, more or less, into the air, as represented in the figure. If the knob of the jar be placed within about half an inch of the conductor of the machine, a succession of sparks will be observed to pass into the jar: these will become gradually weaker when the jar is fully charged.

59. The charge being thus accumulated, it is again discharged by completing a conducting communication between the exterior coating *A* and the charging rod *d b*: the extent of such

communication, termed an electrical circuit, may be almost indefinite. If the discharge be effected through a short circuit of bent metallic wire, terminating in metallic balls, an intensely brilliant explosion ensues, the light of which is almost insufferable at the instant; and there is heard at the same time a sharp crashing report. If a portion of the living animal frame be included in the circuit, then a powerful shock ensues, the intensity of which depends on the amount of the charge.

Sir W. Watson, assisted by other members of the Royal Society, completed electrical circuits of several miles in extent. In August, 1747, they caused an electrical jar to discharge through a circuit of 4 miles, and found the effect to be instantaneous. The Abbé Nollet passed the charge through the whole community of a convent, forming a line of nearly 6000 feet. The shock was simultaneous.

60. In order to avoid the chance of receiving the shock at the time of discharging an electrical jar, two bent wires are employed, terminating in brass balls, and connected by an intermediate joint, so as to open or shut to any convenient distance, like a pair of compasses. These wires are mounted at the joint upon a glass rod or handle, as shown in fig. 28, and constitute what is termed a discharging rod. The balls are either extended directly between the exterior coating A and the knob of the jar, or one of them is connected with the termination of any extent or kind of circuit proceeding from the exterior coating; and the other applied, as before, to complete the discharge.

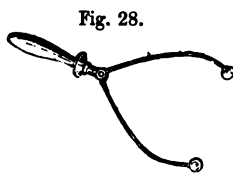


Fig. 28.

61. When a charge is communicated from the prime conductor, the jar is said to be charged positively, or with vitreous electricity. If from the negative conductor, it is said to be charged negatively, or with resinous electricity. In either case it is equally powerful as a means of accumulating and discharging electricity, in a dense and concentrated form, upon

or through any given substance: some idea may be formed of it in observing, that whilst a jar of only moderate size is charging at about half an inch distance from the conductor, several hundred electrical sparks will pass from the machine, and that at the instant of the discharge all these sparks are condensed into one.

62. *Theoretical views of the electric jar.*—It has been seen, Exp. 15 (21), that when two conductors are brought near each other, one of them being insulated and the other connected with the ground, that is, uninsulated, the result is an increased capacity of the insulated conductor for electricity; that is to say, if it be electrified to a certain amount, considered merely as a simple insulated conductor, then, by the approximation of a second uninsulated conductor, it may be electrified in a far greater degree, and yet an electroscope remain insensible to the change. We have already applied this principle to the theory of the condenser (40). As furnishing a primary theoretical explanation of the electrical jar, it may be worth while to illustrate it further in the following way.

*Exp. 26.* Insulate a circular disc of gilded wood A, fig. 29, upon a light rod of varnished glass, and suspend to its upper part, by a filament of silk,

a light reed *t*, terminating in a small pith ball. Electrify this disc A by a spark from a very small Leyden jar, moderately charged. The reed will diverge (15) as shown in figure 5. Whilst thus repelled, let a similar uninsulated disc B, supported on a rod of metal, be

gradually brought near this electrified plate A, the faces of the two plates being parallel to each other. It will be seen that as the uninsulated disc B approaches, the electroscope *t* will collapse, and will, when the plates are near each other, scarcely exhibit any repulsive effect; so that the

Fig. 29.

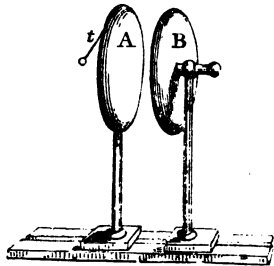


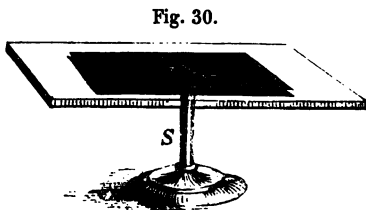
plate A will appear to have lost its charge: now withdraw the neutral plate B, and the electroscope will again diverge.

*Exp. 27.* Whilst the electroscope *t* is in a state of collapse by the influence of the neutral plate, communicate to the plate A, as before, as much electricity as will cause the reed again to diverge to the same extent. Let the neutral plate be again withdrawn, and the electroscope will be thrown off with great violence.

We learn from these experiments, that if an insulated conducting surface be exposed to the influence of a similar surface uninsulated, we may, under this condition, communicate to it a much larger amount of electricity than before, and yet an electroscope be insensible of the change, thereby showing that the capacity of the insulated conductor has been increased.

The conducting plates for these experiments are best constructed of light wood, rounded at the edges and gilded; they may be about 7 inches in diameter and  $\frac{1}{4}$ th of an inch thick; they should be fixed on pieces of wood made to slide on a firm base, as shown in the figure.

63. It is quite apparent that the arrangement just described, fig. 29, is in no sense different from the conditions of the Leyden experiment, and is virtually of the same form as that practised by Smeaton, except that we have given metallic coatings A and B to a plate of air, instead of coating a solid plate of glass. In either case the arrangement is nothing more than the close approximation of an uninsulated to an insulated conductor; the success of the experiment does not therefore depend on the form of the glass, but rather upon its thickness. In applying metallic coatings to opposite sides of a plate of glass, as represented in the annexed figure 30, and connecting one of the sides with the ground by placing it on a metallic support,



Smeaton really fulfilled the essential conditions of the source of electrical accumulation. In this case the upper coating *A* is the insulated conductor, corresponding with the disc *A*, fig. 29, the inferior coating resting on the metallic stand *s*, the uninsulated approximated conductor corresponding with the disc *B*, whilst the intermediate glass is in the same relative position as the particles of air between the plates; that is to say, it is the non-conducting or electric medium.

A square of glass thus prepared has been termed a coated pane, and by the French Electricians, a 'fulminating square.' It is charged and discharged in the same way as the coated jar, by first throwing electricity on the insulated coating *A*, and then joining the opposite sides with the discharging rod.

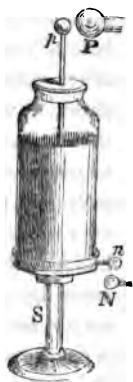
64. Franklin's doctrine of a single elementary electric fluid has obtained, and not without justice, a high degree of general confidence, from its happy elucidation of the phenomena of the electric jar: nothing, perhaps, was ever written on this branch of science which has been so universally read and admired as the 'Observations and Experiments in Electricity' of this distinguished Philosopher. According to the theory of a single elementary fluid, electrical phenomena, as already stated (33), result from a disturbance in the equilibrium of electrical distribution. Now electrics are supposed to be impermeable, or nearly so, to the electric fluid, which cannot move in them except with extreme difficulty, whilst the quantity of electricity they contain can neither be absolutely increased nor diminished: hence, if we by any means force an increased quantity upon any portion of an electric substance, a similar quantity must leave some other portion of it. The Leyden experiment, therefore, according to this hypothesis, consists in the addition of electricity to one surface of the glass, in consequence of which an equal quantity is forced to leave the other surface. Franklin, by a well-conceived and extremely beautiful series of experiments, showed that if either side of a coated jar be electrified positively, the other side is *electrified negatively*, so that there is really no more elec-

tricity in the glass than before, and that the coatings are merely conductors to the charge. To realize this idea more completely, let us imagine that an electrical jar had at first 100 units of electricity, 50 of which occupy a stratum of small depth under one of its surfaces, and 50 a similar stratum under the other. Then, when by the operation of the electrical machine we cause the whole or a part of the 50 units to leave either side and appear on the other, the jar is said to be charged to a greater or less extent. If we collect the total or 100 units on one side, it is said to be charged to saturation. Now the discharge consists in a transfer of the redundant units to the minus surface from whence they were taken, by which the equilibrium of distribution is again restored.

65. The following experiments, resorted to by Franklin in confirmation of this doctrine, are singularly interesting and instructive.

*Exp. 28.* Place an electrical jar, fig. 31, on an insulating support *s*; let the ball *p* of the charging rod be within about  $\frac{1}{4}$  an inch of the prime conductor *P* of the electrical machine. Bring another insulated metallic ball *n*, connected with the negative conductor, within the same distance of a similar ball *n*, projecting from beneath the outer coating *A*. Put the machine slowly in motion, and for every spark which passes between the prime conductor *P* and the charging ball *p*, a similar spark will pass, at the same instant, between the outer coating *n* and the negative conductor *N*.

Fig. 31.

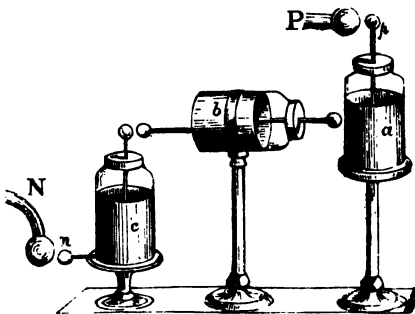


*Exp. 29.* When the jar has become in this way moderately charged, remove it and its insulating support *s* from the balls *P* and *N* of the positive and negative conductors, and having insulated a light metallic ball of about 2 inches in diameter on a glass rod, bring it near the ball of the charging rod *p*; a powerful spark will follow it, and charge

the ball positively. Whilst thus charged, bring the ball near the ball  $n$  connected with the outer coating; the same spark will leave it again, as may be proved by the charged ball having again become neutral: hence precisely the same quantity of electricity taken up from the interior coating has been added to the exterior. In this way, by successive repetitions of the experiment, we may gradually discharge the jar; that is to say, we may take away by small and measured quantities all the electricity communicated to the inner coating, and throw it on the outer. Now if we examine, by means of an electroscope (page 11), the inner and outer coatings of this jar, especially after each alternate contact with the insulated metallic sphere, they will be found in opposite electrical states; that is to say, if the jar be charged from the prime conductor positively, then the outer coating will always evince negative electricity.

*Exp. 30.* Insulate two or three equal and precisely similar jars,  $a$   $b$   $c$ , fig. 32, and arrange them so that the charging ball of the second,  $b$ , may be within half an inch of the outer coating of the first,  $a$ , and the charging ball of the third,  $c$ , within a similar distance of the outer coating of the second,  $b$ . Let the outer coating of this third jar,  $c$ , be placed electrically, by means

Fig. 32.



of a conducting ball  $n$ , within the same distance of an insulated ball  $n$  connected with the negative conductor, as the charging-ball  $p$  is of the positive conductor  $p$ . Put the machine in motion, and directly a spark passes upon the *inner coating* of the first jar  $a$ , a similar spark will appear

to leave the outer coating to charge the second jar *b*: this will again cause a spark to pass from its outer coating upon the charging ball of the third jar *c*, which will throw off the electricity of its outer coating upon the negative conductor *N*, to supply the rubber (43), and is the compensating quantity to the electricity abstracted from it by the glass to be thrown upon the interior of the first jar *a*; so that every jar will be equally and similarly charged, or nearly so, at precisely the same instant, and, by the propagation as it were of electrical impulse, through the series of jars. Now as the second jar *b* has become charged from the outer coating of the first, *a*, and the third jar *c* from the outer coating of the second, *b*, it may be inferred that for every unit of electricity, added to the inner coatings, a similar unit has left the outer, which is what Franklin wished to show. This experiment, although satisfactory when only two or three jars are employed, is greatly embarrassed by extension upon a long series; the accumulated resistance of each jar being at length so considerable as to vitiate the result. When the resistance becomes at all equal to the charging power, the last jars of the series do not become charged in the same degree as the first, and the whole operation is thereby arrested.

66. *Exp.* 31. A coated jar being placed on an insulator, fig. 31, bring the ball *p* of the charging rod near the positive conductor *P* of the machine, and let the negative conductor of the machine be perfectly insulated. Put the machine in motion: little or no charge will have accumulated, even although the outer coating of the jar be connected with the earth. Let the negative conductor and the outer coating be connected by a conducting communication, and the charge immediately proceeds, as already shown (63).

This celebrated experiment was adduced by Franklin as demonstrative of his theory, and as evidence of the charge being nothing more or less than a new and disturbed *distribution of its own electricity* through the agency of the



rubber, the glass, and the prime and negative conductors of the machine.

67. *Exp. 32.* Apply to a glass jar an outer metallic coating only, and give it an interior coating by means of water: insert a brass rod into the water through a cork fitted to the jar, having a varnished glass rod attached to it as an insulating handle: varnish the uncoated interval above the water, inside and out, and let the jar be dry and a little warm: charge the jar as in *Exp. 28*, *fig. 31*: when charged, remove the charging rod by its insulating handle, together with the cork support: take the jar by its outer coating, and carefully pour the water off into another similar jar, prepared with an external coating only: place the charging rod in this: test this jar,—not the least sensible charge will have gone over with the water: let other water be now poured into the originally charged jar, and the charging rod replaced as at first. The whole system may then be discharged in the usual way (58): all the electricity will have remained with the glass.

This was the original form of the experiment, as made by Franklin, in evidence of the disturbed distribution on the glass, and to show that the attached coatings were merely conductors to the charge. At present it is usual to fit open jars with sets of moveable coatings of common tin plate; but the most complete method is to employ a small glass sphere with a long neck, nicely varnished, and give it coatings of dry mercury, which is easily managed.

68. According to the Franklinian hypothesis of a single fluid, the charging of a jar from the negative conductor is the same thing as throwing positive electricity upon the outer coating (65). This may be accomplished either by placing the jar on an insulator, as in *fig. 31*, and connecting the charging rod *p* with the negative conductor *n*, and the outer coating *n* with the positive conductor *p*, or otherwise, by reverting or turning the jar over, so as to rest on the charging rod, and then throwing electricity on the outer coating. When

charged, if we again revert it upon an insulating stand to avoid the shock, it will be charged in precisely the same way as if electricity had been thrown upon the interior coating from the negative conductor, except that at first the excess of electricity upon the inner coating will have been neutralized by the connection of the charging rod with the ground, and there will remain an excess upon the outer or now positive side, being the reverse of the state of the jar when charged directly from the negative conductor in a similar way; in which case there is from the same cause an excess of negative electricity on the interior surface. We have only to abstract a spark therefore, in the first case, from the exterior coating, as in Exp. 29 (65), and we immediately identify the two experiments. It may be likewise further observed, that whether we insulate and connect the opposite coatings directly with the two conductors of the machine, as in fig. 31, or whether the jar remain uninsulated during the time of charging, the result is virtually the same, since the exterior coating opposite the charging side will always communicate with the opposite conductor of the machine through the ground or table on which it is placed (65).

69. Some confusion has frequently been occasioned in all these experiments for want of a clear conception of the electrical conditions of the jar, and the relative position of its inner and outer coatings, which really in no sense virtually differ: all complication will immediately vanish in referring the experiments to a simple plate of coated glass, fig. 30, page 63: here, as is evident, there is no inner and outer coating, and it is quite immaterial which surface we electrify from the positive or negative conductor. We may also further remark, that although in charging coated glass from the positive conductor, we assume, by the Franklinian hypothesis, a communication of redundant electricity to either side whilst in charging it from the negative conductor, we assume, on the contrary, an abstraction of electricity from either side; yet it is quite impossible to determine which is really the posi-

tive and which the negative excitation, and hence it is only by a convenient and arbitrary assumption we refer positive electricity to the excited glass, and minus or negative electricity to the rubber (14). We might, however, as easily imagine the reverse of this, and attribute the redundant electricity to the rubber on the same hypothesis, for any thing we know to the contrary. If we merely consider these opposite states in the light of two absolute forces, which they virtually are, we avoid such arbitrary assumptions, and we have a view of electrical action sufficiently theoretical for a fair exposition of the phenomenon, and sufficiently practical for experimental investigation.

70. Having considered how the hypothesis of Franklin applies to the elucidation of the Leyden experiment, it will be requisite to explain, briefly, the application of the doctrine of two independent fluids (32) to similar phenomena. According to this doctrine, the Leyden experiment consists in a separation of the two electricities resident in the glass, and a transfer of each to its opposite surfaces. Let, for example, the jar be circumstanced as in fig. 31 (65), with its charging rod exposed to the prime conductor, and its outer coating to the negative conductor of the machine. In this case the vitreous or positive electricity of the outer surface, being repelled by the vitreous electricity thrown on the inner surface of the glass, and attracted by the resinous electricity of the rubber, is disengaged and abstracted from the outer coating, leaving the external surface of the glass charged with resinous electricity. In the same way, the resinous electricity of the inner coating being repelled by the resinous electricity now in excess on the outer coating, and attracted by the vitreous electricity of the machine, is also disengaged and abstracted; and thus one surface of the jar becomes charged with vitreous electricity, and the opposite surface with negative electricity. The two electricities of the jar are disunited in the greatest possible degree by the respective actions of the *excited rubber and glass* of the machine, and will tend to re-

combine with a force varying in some direct ratio of the amount of disturbance, and in some inverse ratio of the thickness of the intervening glass by which they are separated. It is this which constitutes the charge as exemplified in the Leyden experiment.

71. Before any conducting communication is established between the interior and exterior surfaces, the tendency of the opposite elements to combine will be exerted in the direction of the intervening glass, and will by their mutual attraction approach as nearly to each other as the resisting intervening electric particles will permit: hence, on removing the coatings, as in Experiment 32 (67), we find the disunited electricities adhering strongly to the surface of the glass under the coating. When, however, we complete a conducting communication between the coatings, and annihilate, as it were, all distance or resistance to the re-union of the two elements, then they re-combine, and the jar is said to be on this hypothesis discharged. If the re-union takes place through the glass itself, which is sometimes the case, then the particles are partly broken through, and fracture of the jar ensues of a very striking and peculiar character.

72. Since the process of charging consists in collecting, by means of the electrical machine, all the vitreous electricity of the jar upon one surface, and all the resinous electricity upon the other, it follows that unless we can bring the machine to act upon both sides of the jar, we cannot obtain a charge: it is requisite, therefore, to establish a communication between each side of the jar and the two conductors of the machine, either directly, as in Exp. 28, or through the medium of the ground: hence it is on this hypothesis that an insulated jar refuses to charge, and that we cannot abstract either of the electricities from one side without at the same time abstracting an equal quantity of the opposite electricity from the other, as may be illustrated by the preceding Experiments 28 and 31, and by the general series of experiments resorted to by Franklin.

73. Although these inquiries show that the solid electric intermediate between the coatings is the means of confining the two electricities to the opposite surfaces of the jar, yet they throw no light on the actual electrical condition of the interior particles of the glass. The beautiful and comparatively recent experiments, however, of Faraday, go far in elucidation of this important question. According to this untiring and profound Philosopher, the particles of non-conducting matter intermediate between opposite electrical powers become, as before explained, fig. 17 (37), placed in a constrained or forced state; the particles assuming positive and negative points which are symmetrically arranged with respect to each other, as roughly represented in fig. 17. This succession of positive and negative attractive powers, exerted as it were in lines between the two limiting coatings, he further shows is accompanied by a repulsive or diverging force in a transverse direction. Now when this condition, termed induction, is permanent, then we have perfect *insulation*; but directly the particles begin to discharge into each other, then more or less of conductivity ensues, and no charge accumulates. If, on the contrary, the transverse action is greater than the particles can support without discharge, then the whole series is, as it were, mechanically deranged or broken up, constituting what he terms 'disruptive discharge,' as in the case of the fracture of a charged electric jar by the power of the two electric forces operating on each other by induction, through its substance, and which is characterized by a peculiar rupture of the glass, a portion of which is reduced occasionally to a fine powder. Hence it is that an interval of air is quite unable to sustain any considerable electrical accumulation as compared with solid electric matter. Faraday has characterized any insulating medium—solid, fluid, or gaseous, through or across which the inductive electric forces act—by the term 'dielectric.'

74. According to these views, not only would the opposite *electricities* be found adhering to the glass beneath the

coatings, as in Exp. 32 (67), but they would have penetrated also into its substance to a greater or less extent, causing a sort of surcharge or residuum after what appears to be a complete discharge of the whole system; and this is really the case.

*Exp. 33.* Let the middle of a circular plate of crown glass, about 8 inches in diameter, be placed between two similar discs of gilded wood of about 5 inches diameter, and about the  $\frac{1}{4}$ th of an inch in thickness: affix a glass handle to one of these, and insulate the whole system upon a slender glass rod, after the way represented in fig. 30. We have then a coated pane with moveable coatings: charge this system by making a temporary communication between the inferior coating and the ground, and throwing electricity on the upper surface: remove now the connection with the ground, and discharge the system with the discharging rod, fig. 28 (60). To all appearance the electrical disturbance has vanished. If, however, we endeavour to detach the upper coating by its insulating handle, an adhesion will frequently arise between the glass and coating, strong enough to lift the whole mass; and if a little time be given, a residuum or return charge is found to occur, sufficient to cause a small second discharge on the application of the charging rod. Faraday, by means of coated lac, observed this return charge after a lapse of ten minutes, and considers it due to the electricity coming forth as it were from the lac, into which it had penetrated in consequence of the forced electrical condition into which the particles had been previously brought.

*Exp. 34.* Complete the discharge as before: remove the upper coating, and subsequently the glass, from off the inferior coating, by carefully lifting it by one point of the edge, so as not to disturb its electrical condition: test the state of the coatings by an electroscope, and it will be found, on the principle of the electrophorous (51), that the upper coating, originally positive, will now appear negative; and

the under coating, originally negative, will be positive, showing evidently the still excited state of the particles of the dielectric.

*Exp. 35.* Deprive the two coatings of all electricity, and replace the system as before, having been careful to preserve the insulated state of the glass plate: complete five or six alternate contacts with the coatings and the finger: first touching one and then the other, a weak spark will be commonly elicited, and if the discharging rod be applied, the system may be again discharged, although in a low degree. As the coatings were rendered perfectly neutral, this subsequent charge could only have arisen from the dielectric particles, much in the way explained by Faraday.

Such are the general theoretical views of the German and Leyden experiments of the years 1745 and 1746, and of the phenomena of the electrical jar. It is true they are not altogether without the region of conjecture, nor are they free from embarrassments, but they possess still powerful claims to attention.

#### THE ELECTRICAL BATTERY.

75. When several coated jars are united electrically by joining together all their charging rods or inner coatings, and placing them on a common conducting base, so as to unite in a similar way all the outer coatings, as represented in the next figure (33), such a combination is termed an electrical battery. When charged from a common source, and discharged in the usual way, they all act together in mass, in the same way as one great jar; and thus we have a means of multiplying electrical accumulation to an almost unlimited degree, provided we can obtain a sufficient charging power. The resulting effect, although not quite in the ratio of the number of jars in combination, on account of the obstructions of the charging rods, &c., will still with a high charging power, aided by jars of sufficient magnitude, have a fair approximation to it. With a battery of 100 jars, each about 13 inches in diameter, 2 feet in

height, and exposing each about  $5\frac{1}{2}$  square feet of coated glass, the celebrated Dutch Electrician, Van Marum, obtained a most tremendous action. When this battery, exposing altogether 550 square feet of coated glass, and charged by the great Teylerian machine (46), was discharged through and upon various kinds of matter, its force was quite irresistible. When passed through steel bars 9 inches in length,  $\frac{1}{2}$  inch wide, and  $\frac{1}{16}$ th thick, the bars became powerfully magnetic: a piece of box-wood, 4 inches in diameter and 4 inches long, was rent in pieces: various metallic substances were melted and dispersed in all directions: an iron wire, 25 feet in length, and about the  $\frac{1}{16}$ th part of an inch in diameter, fell under the shock into red-hot balls thrown in all directions: a piece of tin wire, 8 inches long and  $\frac{1}{8}$ th of an inch in diameter, disappeared in a cloud of blue smoke, from which fell red-hot globules of tin;—these repeatedly bounded from a piece of paper beneath.

76. In the construction of such batteries it has been usual to fit them up in series of small jars, each jar having a separate cover of mahogany, with an attached charging rod and chain. They are placed in partitions, in a case lined with tin-foil, and the jars connected by cross brass rods terminating in balls. The expense of all this is very great; beside that the arrangement is undesirable. The most effective arrangement is to employ jars of as large dimensions as possible, prepared as already explained (58). The fewer jars in a battery the better. They should be grouped on an open conducting base about a central jar, as in No. 1 of the next figure (33), either in groups of five, seven, or any other convenient number. If the battery requires further extension, then unite such groups by joining the charging rods of their central jars. In this way, by the aid of a few common brass wires, and some holes in the charging rods, we may put together in a short time a battery of any required power and of the most efficient description, with very little inconvenience, trouble, or expense.

77. *The management of electrical batteries requires very*

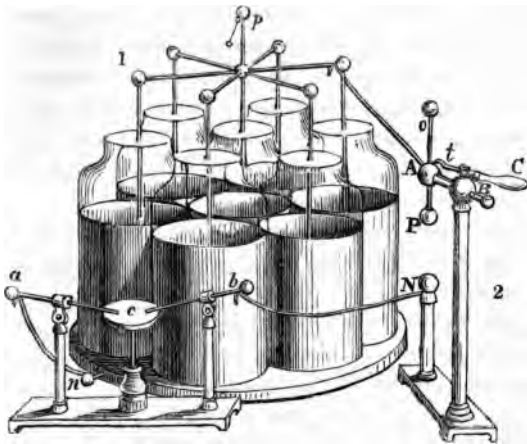


considerable caution. They will be of course easily charged by connecting the central rod with the machine; the charge will then pervade the whole through each conducting ramification: to discharge such a series it will be desirable to employ two insulated balls, one in communication with the central charging rod, and the other with the common conducting base on which the jars are placed, that is to say, with the external coatings, and which balls can be joined at any required instant, as shown in No. 2 of figure 33. In this figure,  $A$  is a ball of brass supported on a rod of varnished glass  $AB$  by a glass pillar and mahogany ball  $B$ . The ball  $A$  has a vertical hole through its centre, so as to admit of a short brass rod  $PO$ , carrying a discharging ball  $P$ , falling freely through it. The rod  $PO$  has two or more small holes drilled in it, by which it can be supported at a given height on the ball  $A$ , by means of a pointed bent wire attached to a hinge joint at  $t$ , and a glass handle  $tc$ . The front of the ball  $A$  receives a direct metallic communication,  $pA$ , with the battery, as shown in the figure. Immediately under the discharging ball  $P$  is a similar ball  $N$ , fixed on a stout pillar of glass, and connected in any required way with the exterior coatings or base of the battery. When the operator requires to discharge the battery through any given circuit, he liberates the bent brass wire support fixed to the joint at  $t$  by a light touch of the glass handle  $c$ . The balls  $P$   $N$  come then together, and the battery discharges without danger to the operator, and always in the same way.

78. When certain substances are exposed to the action of the battery, they are placed on an insulating table between two directing wires, which may be adjusted conveniently for the particular experiment. An instrument of this kind is termed a 'Universal Discharger,' and is represented in No. 3. of fig. 33. The two directing rods  $ab$  are fixed on pillars of glass, and are so arranged as to slide through short spring tubes fixed to the joints, and move in any direction: the insulating table  $c$  is placed between these rods, and is fixed on a glass rod,

supported in a socket of wood so as to slide with friction in some compressed cork, up and down, and be maintained to any required height: the substance to be operated on is to be placed between the directing rods  $a b$  on the table  $c$ , or stretched between them: one of the rods is then connected with the lower insulated ball  $N$  of the battery discharger, and the other with the common conducting base on which the battery rests, as shown in the annexed figure 33, in which it is evident that directly we liberate the ball  $P$ , and cause it to fall on the ball  $N$ , we have completed a circuit  $p A P N b a$ , immediately through the substance exposed to the battery, and extending directly from the inner to the outer coatings.

Fig. 33.



79. It may be proper to observe that several contrivances have been employed for discharging electrical batteries; but that just described is very safe and efficient. Mr. Cuthbertson resorted to a most ingenious method of discharging a battery, by connecting the positive side with a sort of balance set on a knife-edge, and loading one of the arms with a given weight: under the discharging ball of this balance was placed another ball connected with the outside of the battery: the

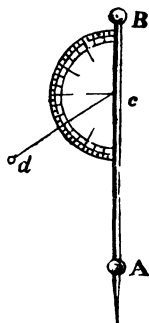
connection of the discharging ball with the charged side being through a second and insulated ball on which it rested, it follows that when the combined attractions and repulsions of these balls are such as to overcome the weight with which the opposite arm of the balance is loaded, then the discharging ball descends and discharges the battery. By a graduated slide and weight, the resisting force can be regulated to any required amount.

#### ELECTROMETERS.

80. An electroscope adapted to the measurement of electrical quantities of any kind is termed an Electrometer, and is employed to determine either the relative quantity of electricity actually in operation,—its attractive or repulsive force under given conditions,—or the comparative effect produced by its discharge in various ways. The different electroscopes already described (39) are converted into electrometers by applying to them some graduated measure of their respective indications. The following may be considered as being amongst the most perfect and available of such instruments.

81. *Quadrant Electrometer.*—This electrometer was invented by Mr. W. Henley, F.R.S., so long since as the year 1772. It has been much employed in certain electrical investigations, especially with the electrical jar, or battery, and appears to be the first instrument of this kind ever used. A short reed *cd*, fig. 34, terminating in a light ball of pith, is set on a delicate axis attached to a vertical conducting rod or stem *AB*. This axis *c* is in the centre of a graduated quadrant or semi-circle, also affixed to the same stem. The stem has a ball at *A*, against which the reed *cd* reposes when not electrified. If charged either by placing the instrument on one of the conductors of the machine, or on the rod of an electrical jar, as in fig. 33,

Fig. 34.

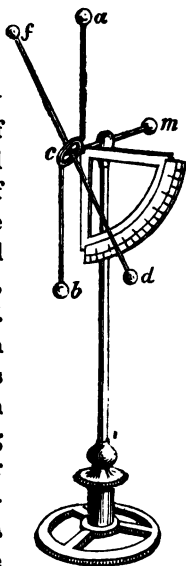


the reed  $c d$  rises into the air, and marks, as an index, on the graduated semicircle the angle of divergence by which the comparative amount of electrification or charge may in some cases be estimated.

82. *Quadrant Electrometer of Double Repulsion.*—This instrument, constructed on the same principle as the former, is represented in the annexed figure 35.

A small elliptical metallic ring  $c$  is set obliquely on a short brass rod  $c m$ , sustained on an insulated support: two light metallic wires,  $c a$ ,  $c b$ , are fixed vertically at the opposite ends of the long diameter of the ring; and these wires terminate in gilded balls of pith or cork. In the direction of the short diameter of the ring, a delicate axis is set on points which by a central transverse pin carries two light reeds  $c d$ ,  $c f$ , forming together one long index  $d f$ . This index also terminates in gilded pith balls: when uncharged, the index reposes against the vertical wire  $c a$ ,  $c b$ : when electrified, either alone, or by connecting the rod  $c m$  with a charged conductor or jar, the index is repulsed in opposite directions, above at  $a$ , and below at  $b$ , by a double repulsion, the amount of divergence

Fig. 35.



being estimated by a graduated quadrant fixed in the centre of the axis of the index, behind the elliptical ring  $c$ . The tendency of the index to a vertical position is regulated by short pieces of reed, placed on the opposite arms of the index, and which slide with friction on them, so as to admit of being placed at various distances from the centre: thus the index diverges with an extremely small force.

83. This class of electrometer, although convenient and useful in many instances, is nevertheless of very limited application in others. We are certainly enabled to say that, *cæteris*

*paribus*, when the angular divergence is the same, the same quantity of electricity is in operation: with a greater divergence, there is a greater quantity,—with a less divergence, less; but how much greater or less it is not easy to determine, since we have with the divergence to take into the account the diminishing force of repulsion as the distance increases, and the simultaneous increasing force of gravity at different angles, together with all the different and varying distances from the centre to the extremities of the repelling arms; and also the variable oblique action of the forces as the index rises; all of which is to a certain extent indeterminate. The common Henley's Quadrant Electrometer, when placed on a charging jar, as in fig. 33 (78), exhibits at first little or no activity. As the charge proceeds, its action rapidly increases: finally, after arriving at an angle of about 60 degrees, the rate of progress of the index is slow, and it will frequently attain a maximum elevation before the charge is complete. We may to a certain extent determine empirically, for a particular instrument, the angular divergence corresponding to given increments of charge, which is perhaps the safest method. The following are the results of some experiments with the electrometer of double repulsion (82) connected with an electrical jar, the quantity of electricity being estimated by the number of revolutions of the machine, or by other known units of measure to be presently described (90).

Measures of Electricity.	Degrees of Divergence.
10	1
20	5
30	12
40	16
50	20
60	28
70	30
80	34
90	36
100	40

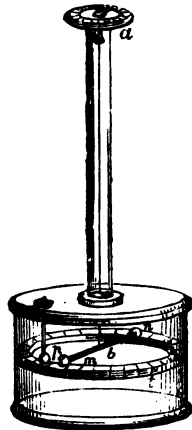
If we reject the first two experiments, and commence with 30 measures, we find the angular divergence of the index nearly as the quantity of electricity: thus the divergence corre-

sponding to the quantities 30, 60, and 90, which are as the numbers 1, 2, 3, is 12, 28, 36, also as the numbers 1, 2, 3, or nearly so; and this law appears to obtain generally within certain limits. Thus the quantities 50 and 100 give the divergences 20 and 40, which are both as 1 : 2.

Mr. Achard states, that to estimate truly the repulsive force by Henley's electrometer, the graduations should be according to a scale of arcs, the tangents of which are in arithmetical progression. The truth of this, however, can only be ascertained experimentally.

84. *The Balance of Torsion.*—M. Coulomb, a distinguished member of the Royal Academy of Sciences at Paris, describes, in a memoir presented to the Academy in the year 1785, a method of balancing the force of electrical repulsion against the re-active force of a fine wire, suspended vertically, and twisted more or less from its quiescent position. This has been termed Coulomb's Electrometer, or the Electrical Torsion Balance. It is represented generally in the annexed figure 36. A fine wire or silver thread *a b* is attached to a rigid pin at *a*, and carries at its lower extremity *b* a small weight, and a transverse arm or lever *m n*: a small gilded disc of paper or a gilded pith ball, insulated on a thread of lac, is attached to one end *m* of this arm, and a paper vane, to arrest the oscillations, is fixed to the other extremity: the whole is enclosed in a cage of glass. Immediately next the ball *m* there is another similar ball *p*, insulated and supported by the cover of the glass cage: the centre of this ball corresponds with zero of a graduated circle *p t n*, surrounding the cylindrical glass cage. When the ball *p* is electrified, or charged, and placed in the cage through a hole in the cover,

Fig. 36.



so as to touch the ball  $m$  of the lever, then the balls repulse each other (15); the lever  $m n$  is turned upon its centre, and the suspensory wire becomes more or less twisted: by this a re-active force is produced, and a balance obtained to the electrical repulsion at a given point. Now Coulomb proved that this re-active force, or tendency of the wire to return to its previous position, was exactly proportionate to the angle of its torsion, or twist: hence the angle of divergence of the small insulated ball  $m$  of the lever  $m n$  is the measure of the repulsive force, in terms of the angular torsion of the wire, referred to the same angular distance between the balls. Thus, supposing by a charge communicated to the ball  $p$ , the balance had taken place at an angle of  $36^\circ$ , as read off on the graduated circle  $p t n$ , then  $36^\circ$  would represent the repulsive force at that angular distance in terms of the re-active force of torsion, for the wire must have been twisted to that same degree. If we require to find the value of the repulsive force at some other angle, say  $18^\circ$ , then it becomes necessary to impress a new force of torsion on the wire, such as will act against the repulsive force, and maintain the balls  $p m$  at that angle. Now this is effected by turning the central pin at  $a$ , to which the wire is fixed, in opposition to the direction of the repulsive force, until we oblige the balls to rest at the given angle: this is the new force of torsion. For the measurement of this, there is a graduated circle and index, with a milled head at the termination of the suspension pin  $a$ , by which the increased torsion is produced and measured. Thus, supposing that in order to maintain the balls at  $18^\circ$  we had twisted back the wire  $a b$   $126^\circ$ , as measured by the graduated circle at  $a$ , then  $126^\circ$ , together with  $18^\circ$ , the angular torsion at  $18^\circ$  would be the total force at that angle, and we should have the numbers 36 and 144 for the relative values of the repulsive forces at the angular distances of  $36^\circ$  and  $18^\circ$ .

In every experiment with this instrument the charge must be communicated to the ball  $p$ . To estimate the electrical *state of any charged body*, the electrified ball is first brought

into contact with it, and then transferred to the balance: the assumption is, that the ball becomes charged in a similar ratio to that of the point of contact of the given electrified substance. A more detailed account of this elegant instrument, together with certain requisite corrections and precautions, will be found in the 'Memoirs of the Royal Academy of Sciences' at Paris, for the year 1785. The use of it is somewhat tedious and difficult, and demands rather more than ordinary skill in electrical manipulation.

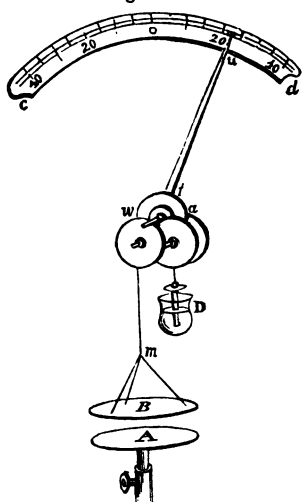
85. *The Bifilar Balance.*—In this electrometer, an account of which will be found in the 'Transactions of the Royal Society' for 1836, a re-active force is obtained by means of a lever at the extremity of two parallel and vertical threads of unspun silk, suspended within a quarter of an inch of each other from a fixed point: the threads are stretched more or less by a small weight, and the repulsive force is caused to operate much in the same way as in Coulomb's Balance of Torsion. As the threads turn or tend to turn, as it were, upon each other, the stretching weight becomes raised by a small quantity, and thus gravity is brought to re-act against the repulsive force in operation. Its delicacy is extremely great, and will render sensible a force of the  $\frac{1}{100000}$ th part of a grain. There are facilities in the mechanical detail of this instrument for examining electrical forces, and testing the electrical state of charged conductors, not found in the Balance of Torsion.

86. *The Hydrostatic Electrometer.*—In this instrument, represented generally in the annexed figure 37, the force of electrical attraction is balanced against the position of a small cylinder of wood, partially immersed in water. A light circular disc *A* of gilded wood is fixed on an insulating rod at the extremity of a graduated slide, by which it may be raised or depressed to any required distance: opposed, and directly over this, is a similar disc *B*, insulated by a suspensory silk thread, and attached to a silver thread *t m* passing over a quarter of the circumference of a wheel *a t w*. This



disc is counterpoised by a short cylinder of wood, suspended in a similar way from a silk line  $t$   $a$  passing round the opposite side of the wheel, and partly floating in water contained in a small glass vessel  $D$ : the wheel at  $w$  is supported on friction wheels, to give it great freedom of motion, and carries an index of light reed, moveable over a graduated arc  $c$   $u$   $d$ : the centre of this arc,  $u$ , is marked zero, and the graduations extend to an angle of  $30^\circ$  on each side, divided into 60 parts: from zero to each extremity,  $c$  and  $d$ , the counterpoise may be so adjusted, either by small weights, or by

Fig. 37.



a regulating screw, with which the glass vessel  $D$ , containing the water, is connected, that when in equilibrium at a certain immersion of the float, the index  $t$   $u$  is at zero of the scale. It follows from this arrangement, that so soon as any new force is brought to operate on either side, suppose the force of gravity by the addition of a few grains' weight, then the cylindrical float  $D$  either sinks or rises in the water, until the fluid it continues or ceases to displace balances the weight added; and this is shown by the position of the index on one of the arcs  $u$   $c$  or  $u$   $d$ , according to the direction of the motion; so that any force operating between the circular discs  $A$   $B$  may be referred to a known standard of weight determined experimentally in terms of the divisions of the arc  $c$   $d$ .

In the application of this instrument to electrical inquiries, the suspended disc  $B$  is connected with its silver thread of suspension by a very fine wire, and electricity communicated to the fixed disc  $A$ , supposing the force to be estimated is an

attractive force. For repulsive forces we allow it to hang by its varnished silk thread, and charge it with a similar electricity to that of the disc *A*, either by bringing the discs in contact, or by other insulating temporary communication, or by a separate and temporary charge. In either case the forces are shown on the arcs *u d* or *u c*, and the distance of their action estimated by the amount which the disc *B* had been previously depressed from contact with the disc *A*, together with the addition or subtraction of the distance through which the suspended disc *B* has moved by the force in action, either towards or from *A*, known by the measured circumference of the wheel *a t w*, and the degrees it has turned in either direction.

A more particular description of this electrometer will be found in the 'Philosophical Transactions of the Royal Society' for the year 1839, and also for 1834. Quantities of electricity producing certain forces of attraction between the discs are in proportion to the square roots of these forces: thus, if in any two experiments, with a constant distance between the discs *A B*, the forces in degrees of the arc are found as 1 : 4 or as 1 : 9, then the quantity of electricity accumulated would be as 1 : 2 and as 1 : 3 respectively: hence the forces are as the square of the quantity of electricity in action. By always maintaining the same charge in *B*, and varying the distance *A B*, the forces will vary in an inverse ratio of the squares of the distances: thus, if in any two experiments the distances are as 1 : 2 or as 1 : 3, the forces will decrease in the proportion of 1 : 4 or 1 : 9 respectively. Such are the laws which determine the value of the indications of this electrometer: it is free from many sources of error and difficulties which embarrass the use of the delicate balance of Coulomb or the Bifilar balance, only requiring a careful adjustment or estimate of the respective distances between the discs, and which is not difficult to arrive at. Care also must be taken to discharge the air and discs for each new experiment, since the conditions of the attraction

are precisely those of the Leyden experiment (64). The discs are merely coatings to a plate of air, and if we move them much out of their place, by further separation, the charge remains behind: hence it is desirable to regulate the distance at first, or as nearly as may be. The support on which the instrument is mounted is for this purpose furnished with regulating rack-work and a milled hand lever, by which the whole of the machinery, with the float, may be raised or depressed through any required distance, estimated on a scale divided into inches and tenths.

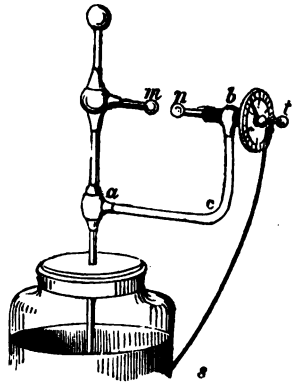
87. *The Scale-beam Electrometer.*—The common scale-beam furnishes an exact measure of electrical forces, and has been applied to the purposes of an electrometer with considerable effect: the particular arrangements are described in the 'Philosophical Transactions' for the year 1834; they are very similar to those of the electrometer just described, with the exception, that a common balance is substituted for the wheel-work, the suspended disc counterpoised, and the force of the attraction estimated by weights placed in the scale-pan: the laws of the instrument are in all other respects the same.

88. Electrometers more especially applicable to the Leyden jar, in combination with the preceding, are of great moment in electrical investigations; particularly those by which a precise estimate of the quantity of accumulated electricity may be attained: several methods have been resorted to for the construction of such instruments, including the principle of the fusion of metallic wires by the subsequent discharge. Most of these are defective. The most available and satisfactory measures are Lane's Discharging Electrometer, and what has been termed the Unit Measure: the following is a brief explanation of these two essential instruments.

89. *Discharging Electrometer.*—This electrometer was contrived about the year 1767, by Mr. Thomas Lane, a medical practitioner of London, his object being to obtain repeated *discharges from an electrical jar of a given force.* The instru-

ment consists of a bent arm of glass, *a c b*, fig. 38, attached to the charging rod: this arm sustains a screw-slide and tube carrying a rod and ball *n t*, connected with the outer coating of the jar by a wire *s*, the ball *n* being, by means of a graduated circle and index at *t*, set at a measured distance from a similar ball *m*, projecting from the charging rod: these balls thus regulated, by withdrawing the ball *n* from contact with *m* through a given distance, say two-tenths

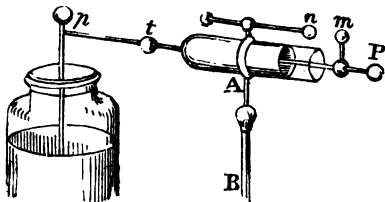
Fig. 38.



of an inch, the charge is allowed to proceed. When the increasing force is sufficiently powerful, an explosive or disruptive discharge ensues between the balls *m n*, and it is demonstrable that the relative quantity of electricity accumulated at the instant of this discharge is directly as the distance between the points of contact of the balls: thus, when the discharge takes place at a distance of four-tenths of an inch, the quantity of electricity accumulated would be double that producing a discharge at two-tenths of an inch, and so on.

90. *The Unit Measure.*—This species of electrometer, for the measurement of quantity, consists of a small Leyden phial, *A*, fig. 39, about five inches in length, and three-fourths of an inch in diameter, fixed horizontally on a long insulator *B*, and interposed between the machine and the jar or battery to be charged: this small phial is coated to within two inches of the top, so as to expose about seven square inches of coated glass; the charging rod is con-

Fig. 39.



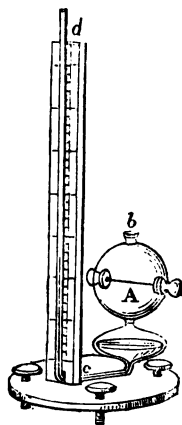
necting rod is con-

nected at  $P$  with the charging conductor of the electrical machine, and the outer coating with the jar or battery by a rod  $t p$ . When the machine is set in motion, this small jar begins to charge, and by the law of the Leyden phial (64), as much electricity as is thrown upon the inner coating by the electrical machine, leaves the outer coating to be distributed over the inner surface of the jar. When the accumulation in the phial has reached a certain height, as measured by two small balls  $m n$ , connected with the inner and outer coatings on the principle of Lane's electrometer, just described, a disruptive discharge or explosion ensues, which marks a certain quantity of electricity received as a charge by the battery in terms of the small jar, charged to a given degree: this explosion, for all practical purposes, sufficiently discharges the phial, and places it in a state to mark a similar quantity, at each instant of the explosion, as the charge proceeds: thus the small jar becomes a sort of unit of measure, showing how many phials of electricity have, as it were, been poured into the charged jar or battery: it is only requisite to fix the distance between the exploding balls  $m n$ , by means of a graduated slide on the charging rod, so as to determine the magnitude of this unit, and assign precisely its comparative relative value.

91. *The Thermo-Electrometer.*—The effect of the electrical discharge on metallic bodies is to raise their temperature to a greater or less degree, and in many instances to render metallic wires red-hot, and to dissipate them in a shower of melted globules. The fusion of wire has accordingly been resorted to as a measure of the force and extent of electrical accumulations on coated glass. Independently of this method being uncertain and tedious, it is in many instances quite inapplicable to many refined inquiries. The thermo-electrometer, whilst it avoids all destruction of the metal, indicates at the same time the comparative heating effect of the discharge, and admits of an accurate estimate of the force in operation. It consists of an air thermometer  $A C D$ , fig. 40, having a fine

wire of platinum passed air-tight across its bulb *A*, which is screwed also air-tight on a small open vessel containing a coloured liquid, and soldered at the extremity of a bent glass tube *A c d*. The long vertical leg *c d* of this tube is supported by a graduated scale of inches and tenths, on a convenient foot *c*, the lower part of which is marked zero, at the point where the coloured liquid in the short leg finds its level. There is a small screw-valve at *b*, to admit of an opening with the external air, so as to adjust the fluid to zero.

Fig. 40.



When an electrical accumulation from a jar or battery is passed through the wire in the ball *A*, the temperature of the wire is more or less raised, which causes the air to expand and press the coloured fluid up in the long leg *c d*, the altitude being measured on the graduated scale. In this way a comparative numerical value of the heating effect of the discharge may be arrived at; and it is found that the height to which the fluid rises in the leg *c d* is, *ceteris paribus*, as the square of the quantity of electricity discharged through the wire. The delicacy of this electrometer will depend on the size of the platinum wire, which, for ordinary purposes, may be from the  $\frac{1}{80}$ th to the  $\frac{1}{100}$ th of an inch in diameter, and about 3 inches in length, corresponding with the diameter of the ball of the thermometer. A more detailed account of this instrument is given in the 'Philosophical Transactions' for the year 1827, and also in the 'Edinburgh Philosophical Transactions' for 1832.

## IV.

### ELECTRICAL ACTIONS AT A DISTANCE.

Experimental Inquiries of Faraday—His views of Electrical Actions at a Distance—Theory of Induction reviewed—Ordinary Electrical Attractions and Repulsions considered—Laws regulating these Forces.

92. The operation of those mysterious and intangible powers in nature, by the agency of which masses of matter influence or act upon each other at very sensible and even very remote distances, is a subject of physical research replete with the most exciting interest, whether we refer such actions to the question of gravity as exerted between the sun and planets at distances of many millions of miles, or to the influences of bodies electrically charged through comparatively insignificant distances of only a few inches. In either case the fact is equally wonderful and important. Philosophers, for a long time unable to assign any sufficient cause in explanation of the action of bodies on each other at a distance, have been content to consider such phenomena in the light of a mere matter of fact—upon which they have been led to rest a theory, without inquiry into the cause. The profound and admirable electrical researches, however, of Faraday, have placed us at least greatly in advance of this position, as regards the apparently distant action of bodies on each other through the agency of electricity.

93. Before entering upon a particular analysis of the laws regulating such actions, it may be useful, for a clear view of this subject, to revert once again to the results of these most admirable inquiries, inasmuch as they afford a fair view of the nature of the operation by which an electrical force, originating in a certain place, is propagated and sustained at a distance, appearing there, in another place, as a force of a similar kind.

We have already seen, in Exp. 16 (22), that the electrical

phenomena of attraction and repulsion are altogether dependent on a preparative principle, termed induction, by which a substance is first rendered attractable or repellent, and then, attracted or repelled. Now this principle of induction is of the utmost generality in electrical action,—it is the essential function of all electrical development; so that in fact we cannot proceed with any investigation of the laws of electrical action without some practical knowledge of its nature.

94. According to Faraday, the immediate effect of any substance electrically charged is to bring the particles of matter adjacent to it into a peculiar constrained condition, consisting of a new distribution of the electrical forces resident in them, and by which these forces become placed in a new and certain relative position in regard to the electrified substance. These proximate particles, thus affected electrically, now act on the next particles contiguous or adjacent to them, and these on the next, and so on, through a given series, until the electrical forces throughout the whole become symmetrically arranged in a succession of positive and negative points, as already explained, fig. 17 (37), or in other words, become polarized; thus propagating the original force to a distance where it appears to be sustained as a force of the same kind, equal in amount, but opposite as to direction, Exp. 14, fig. 8 (20). Now in metallic and other good conducting bodies this polarization of intermediate particles does not subsist for an instant, in consequence of the particles communicating the opposed forces from one to the other, by which the whole condition is lowered; it is, in fact, this act of discharge from particle to particle which constitutes conduction. Metallic or other conductors, therefore, will only exhibit the polarized state as a whole; as in the instance of the faces *b c* of the cylinder *B*, fig. 8, page 15. Now this result is altogether independent of the mass of the body, and requires no sensible thickness for its mere development; a piece of the thinnest leaf gold may become positive on one surface, and negative on the other, and that without the least interference of the two

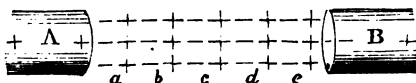


electrical forces. Hence all charge will necessarily be found on the surface of conductors (23), since it is only there that the surrounding resisting dielectric medium, capable of sustaining induction, and upon which the charge depends, is found to commence. If the conductor be hollow, or contain air, still there can be no induction, because of the opposing actions in all directions from the interior surface of the charged body. In the case, however, of electric substances forming dielectric media, as in the instance of the intermediate air or glass between limiting conducting planes, figs. 29 and 30, pages 62 and 63, the thickness of the stratum has a most important influence. In such media the forces cannot discharge, as it were, into each other, as in the former case; and the result is a permanently existing polarization throughout the whole series, constituting what has been termed Insulation (8), (11), and by which a sort of propagation of force is obtained, by virtually the same particle of electricity, throughout a series of contiguous particles of matter, until the force reaches some limiting conducting surface, and appears there at a distance from the point in which it originated. By the term contiguous particles, we are to understand, following particles, or next particles of matter, without any relation to the question of nearness, or of indefinitely small distances between such particles: by polarity we understand, also, such a disposition of force as enables the same particle of matter to acquire opposite powers in different parts, as represented in fig. 17 (37). Moreover, it follows, that since the force in any point of such a series is made up of the forces acting in all directions, that is, the resultant of such forces,—we may conceive the existence of a relative force transverse or oblique to the line of the direct inductive force, equivalent to a dilatation or repulsion of such lines, and the force of which may be expressed by the term *tension*.

95. We see, then, according to these views, that the first effect of an electrically charged or excited body upon an insu-  
medium is upon the particles of the medium immedi-

ately next it: these operate in a similar way upon the following or contiguous particles, until the force extends to some distant body; and there is probably no distance so great as to be without the reach of this progressive propagation, although with the same constraining or originating force the polarization goes on more easily as the extent of the medium is lessened, since there will be in this case fewer particles in the line of action opposing a united resistance to the forced or polarized state; and reciprocally we must ever consider this state as forced or constrained: it is sustained purely by the force of the originating electrical charge, and sinks back or returns to its normal or quiescent condition directly the originating force is removed. The annexed diagram, fig. 41, may serve to elucidate these principles more completely. Let A be

Fig. 41.



a charged conductor, charged either positively or negatively; B a distant neutral and similar conductor; and *a b c d e* intermediate particles of air, or any other insulating dielectric medium: then if A be charged positively, we obtain a series of successive intermediate positive and negative forces symmetrically disposed, the negative after the positive, until we reach the *proximate* face of the conductor B; upon which, by the law of the series, a similar force will appear to that exerted by A, but of an opposite kind as to direction—that is, it will be negative; but as this progressive insulation of the particles or forces cannot exist in the conductor B, we only there obtain the polarized state as it were bodily, that is, where the insulating medium begins or continues, so that the next companion force is found towards its more distant parts;—thus whilst the intervening dielectric medium is polarized, particle after particle, the limiting conductor is polarized as a whole: but if we can in a distant conductor B, obtain in this way a similar

force in effect, although opposite in direction to that of the permanently charged body  $A$ , it may be inferred, as we have shown experimentally (21), that this newly developed force should also operate or re-act by a sort of return action upon the original conductor in a very similar way, by which the state of polarization would be further exalted, and the force in the near surface of the charged body  $A$  strengthened, since the return action would conspire with the original power of polarization as to the position of the intermediate forces, much in the same way as would arise supposing the distant conductor  $B$  to be also permanently charged with a force opposite in direction to that of  $A$ ; in which case it is evident that both conductors would act similarly.

96. Such, then, we may conceive as being the existing state of things in every case of electrical action at a distance; so that attraction by electrical agency is really the consequence of charge, as in the Leyden experiment: we require for its development two oppositely electrified surfaces, usually limiting conductors, with an intermediate dielectric medium, the particles of which become polarized in a similar way in each direction. Repulsion, on the other hand, originates in a similar electrical arrangement; but in this case the intermediate polarization is not the same in both directions, but subversive.

97. It may be perceived from these several considerations, that the laws regulating electrical action at a distance are of a very complex and embarrassing character, since the forces are derived entirely from the previous operation of induction. If we take, for example, the thin suspended disc  $B$  of the electrometer, fig. 37, page 84, and allow it to remain perfectly insulated by its thread of varnished silk, the attractive force, on communicating any amount of charge to the opposed plate  $A$ , will be extremely small, the extent of electrical change by induction which it can suffer being very limited. If, however, we increase its thickness, or give it conducting extension of any kind, either by erecting on its upper surface a short me-

tallic rod, or placing a light hollow cylinder on it, then the attractive force becomes more freely developed, the bodily polarization or separation of the forces being then more complete. If it be uninsulated, the capability of bodily polarization is the greatest possible, and then the attractive force of which the two plates are susceptible, under given conditions of charge and distance, becomes fully apparent. It would hence be only in the case of attractive forces between charged and uninsulated conductors, as in the conditions of the Leyden experiment, or otherwise, between two conductors in each of which the opposite forces were completely developed by an independent charge, that we may expect to find a uniform and regular law of action. In any other case in which the inductive charge upon the attracting surface becomes impeded, or in any way cramped or interfered with, the uniform law would be liable to great disturbance, and the forces might appear as being not subject to any regular law whatever,—a conclusion arrived at by some of the early inquirers in this department of science. If this be true of attractive forces, it is still more apparent in the forces of repulsion, since here the force is sustained no longer than the permanency of the perfect resisting polarization, the inductions being subversive of each other, as acting from similarly charged bodies in exactly opposite directions, so that the real tendency of the repulsive force is virtually a tendency to induce attractive force, as before : hence it follows, that whenever the polarization in either direction, however small, is affected, and the resistance of the particles to a given extent overcome, then any uniform law of the repulsive force is disturbed immediately ; and as this would be more likely to ensue at lesser distances, in which the united resistance of the dielectric particles becomes diminished, cases may arise in which the repulsive force might vanish, and be superseded by a weak attractive force, which is really found to happen ; and it is by no means uncommon to find similarly electrified bodies repel at one distance and attract at another. This circumstance it is which renders all investigations with electrometers

of repulsion in many cases precarious and unsatisfactory. Electrical actions at a distance therefore are, as already observed, very embarrassing. The agency upon which they depend, unlike gravity or other central forces, does not depend on the mass of bodies and on some general law of distance, but operates, under certain conditions, exclusively between their surfaces through the intervening particles of a resisting medium.

98. In the consideration of ordinary electrical attraction or repulsion, three primary, or more elementary conditions, claim consideration: 1°. the thickness and polarization of the intervening dielectric medium: 2°. the direct and reflected polarized disturbance of the limiting conductors: 3°. the force with which they tend to approach or recede from each other, all impediments to motion being removed.

In all experimental inquiries into the laws of these electrical forces it is essential to have the most perfect instruments of measure, and to provide for an accurate means of obtaining and determining the quantity of electricity originating a charge. The electrometers described (83) (84) (85) (86) are perfectly adapted to such inquiries, and we may obtain given quantities of electricity originating a charge, either by the communication of a certain number of very small measures through a unit measure (90) to a simple insulated conductor, or an electrical jar, or otherwise, by the transfer of small quantities of electricity taken upon an intermediate or insulated carrier disc from one of the coatings of an insulated jar charged with a given measured quantity, as practised in Exp. 29 (65). A jar exposing about 100 square inches of coating will furnish a series of small charges so nearly alike, that for all practical purposes they may be taken as identical. We have only to be careful in the selection of the carrier conductor; this should be a small insulated disc or plate, which, after contact with the jar, must be brought fairly in contact with the insulated conductor to be charged: it will then, by the great differences of the extent of surface and the in-

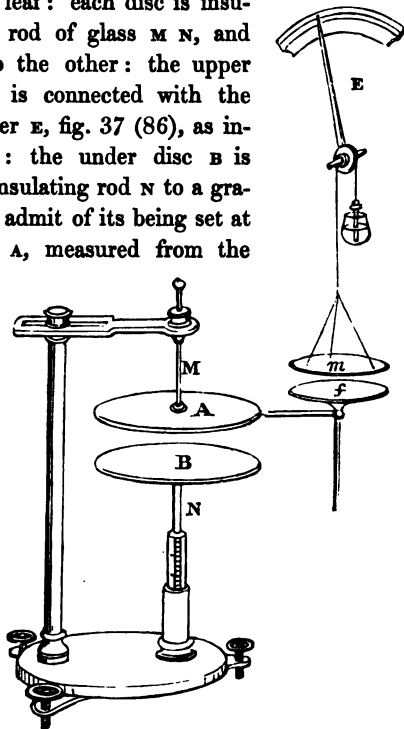
fluence of opposed neutral bodies, be so completely deprived of its electricity, that the quantity it may be supposed to retain on removal will for any required extent of charge become indefinitely small. It will be also requisite to neutralize the opposite coating of the insulated charging jar after each contact (65).

99. This premised, we are prepared to understand the following experiments: *A B*, fig. 42, are two circular discs of light wood, about 12 inches in diameter, and covered with metallic leaf: each disc is insulated on a varnished rod of glass *M N*, and placed parallel one to the other: the upper disc *A* is fixed, and is connected with the hydrostatic electrometer *E*, fig. 37 (86), as indicated in the figure: the under disc *B* is attached through its insulating rod *N* to a graduated slider, so as to admit of its being set at given distances from *A*, measured from the points of contact. In this arrangement we have, as is evident, all the conditions of the Leyden experiment, and full means of examining experimentally the laws regulating the inductive action, as represented in fig. 37, page 84.

*Exp. 36.* Connect the plate *B* with the ground, and separate the two plates by a

given measured distance, suppose  $\cdot 3$  of an inch, and set the attracting plates *m f* of the electrometer also at a measured

Fig. 42.



distance, suppose  $\cdot 4$  of an inch : under these conditions charge the insulated plate by successive experiments, with double, treble, &c. quantities of electricity. It will be found that the march of the electrometer in degrees will be as the squares of these quantities ; that is to say, with twice the quantity of electricity the force indicated by the electrometer will be four times as great,—with three times the quantity, nine times as great, and so on, which is the law before alluded to (86) as affecting the instrument as a measure of quantity : the following are examples of numerical experimental results at the distances just given :

Quantity.	Force.
1	2·5
2	10
3	23

In order, therefore, to obtain the actual relative quantities of electricity affecting the electrometer, we must represent them by the proportion of the square roots of the indicated forces.

*Exp. 37.* Charge the insulated plate A with a given measure of electricity, and let the insulated plate B be placed in successive experiments at different distances from A, say at the successive distances  $\cdot 3$ ,  $\cdot 6$ ,  $\cdot 9$  of an inch, the upper plate being charged with the same quantity of electricity at each experiment. The attracting discs of the electrometer being set at  $\cdot 4$  of an inch apart, the indicated degrees of force developed in the upper plate will be as the squares of these distances directly ; that is to say, as the distance between the plates A B decreases, the force of the upper plate on the electrometer increases, and as the squares of these distances, as seen in the following results :

Distance of Plates.	Force.
$\cdot 3$	3
$\cdot 6$	12
$\cdot 9$	27

Now this experiment may be taken to measure the return or reflected induction of the uninsulated limiting conductor on the upper charged plate ; and since the quantities of electricity

affecting the electrometer are as the square roots of the degrees of force (Exp. 36), it follows that these quantities were directly proportioned to the distances between the two plates; that is to say, when the distance is double, the quantity of electricity affecting the electrometer is also double.

*Exp. 38.* This law, although true for the case of a perfectly uninsulated opposed plate, is not the same when we limit that condition: on taking the opposed neutral plate insulated, it is found that the electrometer only varies in a direct simple ratio of the distances;—hence the quantities of electricity operating upon it are as the square roots of the distances. Thus, at the distances  $\cdot 3$ ,  $\cdot 6$  and  $\cdot 9$  of an inch, the forces were  $3^\circ$ ,  $7^\circ$  and  $10^\circ$ , instead of  $3^\circ$ ,  $12^\circ$  and  $27^\circ$ , as before,—the one being as the simple distance, and the other as the square of the distance, or nearly so; so that it is apparent we have, by altering the electrical capacity of the opposed plate, completely changed the law of this peculiar influence.

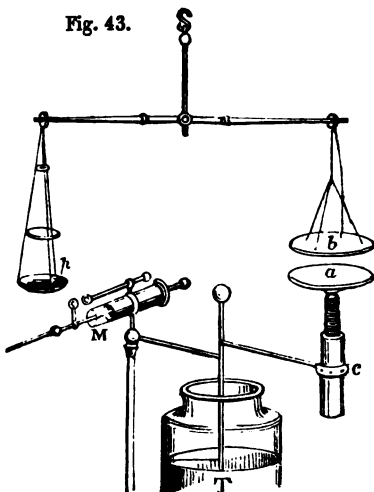
100. If we conceive the whole force exerted between two conductors circumstanced as in the conditions of the Leyden experiment, and free to obey any tendency towards each other, to be compounded of these two inductive actions, together with the distance or thickness of the intervening dielectric; then, supposing these actions to be uniformly progressive, and to proceed according to the law just given (Exp. 37), it would follow that in decreasing the distance between two such conductors we double the inductive force, and diminish the thickness of the dielectric stratum of particles at the same time, by which we further increase the force of polarization; and hence the total force between the limiting conductors, tending to pull as it were the bodies together, would increase in some inverse ratio greater than that of the simple distance between them: on the other hand, if the inductive charges be limited or restrained, the force may be only as the simple inverse ratio of the distance, that is, of the thickness of the dielectric medium between them; and such is really found to be the case.

*Exp. 39.* Place the discs *a b*, fig. 43, of the scale-beam



electrometer (87) at a given distance, by means of the graduated slide at *c*: counterpoise accurately the suspended disc by weights placed in the scale-pan at *p*: connect the insulated disc *a* with a large jar *τ*, by a connecting rod at *c*, and charge the jar with measured quantities of electricity through a unit measure *M*: place a given weight in grains in the

Fig. 43.



scale-pan *p*, and find the number of measures required to turn the beam: take this as a unit of quantity: now change the distances *a b*, and it will be found that the weights required to balance the same unit of quantity will be in the simple inverse ratio of the squares of the distances precisely: the scale-beam will turn with the greatest precision.

Thus when the distances are 4, 8, 12, 16 divisions of the slide at *c*, graduated into tenths of an inch, the weights required to balance 30 measures thrown upon the jar *τ*, and communicating with the lower disc *a*, are 16, 4, 1.75 and 1 grains respectively: thus when the distance is increased to twice the extent, the force has decreased to one-fourth; when it is three times the first distance, the force has diminished to one-ninth, or nearly so, and so on.

*Exp. 40.* In the former experiment the disc *b* is a simple plate, perfectly free by metallic communication with the scale-beam. If we now substitute for it a light gilded cylinder, and insulate it from the balance by a varnished silk line, we

shall then limit the inductive polarization of which the terminating surface is susceptible, and the forces will approach nearer the inverse ratio of the simple distance between the surfaces  $a b$ : in most cases they will be exactly in this ratio. Hence it is clear that the general law regulating electrical attraction exerted between distant conductors is altogether a compound result of two inductive elementary conditions. If one be limited or constant, it will vary with the other.

These experiments may be also made with the hydrostatic electrometer before described (86), the force being taken in degrees.

101. The arrangement represented in fig. 42, page 97, is directly applicable to a perfect analysis of the law of electrical accumulation in the Leyden jar, involving essentially the conditions of induction.

1°. Imagine for an instant that whilst the insulated plate  $A$  is charged with a given quantity of electricity, and at a given distance from the free plate  $B$ , the electrometer  $E$  indicates a certain number of degrees, say 16 degrees, it were possible to increase the area of these plates  $A$  and  $B$ , say to twice the amount; then it is clear that the same charge being disposed over *twice the area*, there could only be one-half the quantity of electricity in any one point; but the electrometer being affected in the proportion of the square of the quantity communicated to its lower plate  $f$ , fig. 42, it would then only indicate 4 degrees; and this is found to be really the case,—for by varying the magnitude of the discs, and preserving the same charge, we obtain that result. Imagine then with this increased or double extent of conducting surface or coating, and the electrometer at 4°, we now proceed to double the quantity of electricity; we should, by Exp. 39, bring up the electrometer to 16°, as before. Hence the charge which can accumulate under the *same* degree of attraction or *electrometer* intensity, is as the opposed areas directly; so that if the area and quantity of electricity vary together, the electrometer will remain unchanged; and hence, as is found to be the case, a given

number of degrees of an electrometer may with electrical jars of different magnitudes represent any accumulation whatever.

2°. Imagine that with a given distance between the plates  $A B$ , and a given quantity of electricity, and whilst the force evinced by the electrometer is a given number of degrees, say 4 degrees, we had increased the distance of the plates; suppose we had doubled it, then the attractive force between these plates would become reduced to one-fourth by Exp. 39. Imagine now that the quantity of electricity is doubled also; then the force between the plates  $A B$  would, by Exp. 39, be the same as before, since it varies with the square of the quantity. Hence the accumulation between two coatings is, under the same attractive force exerted between them, directly as their distances apart.

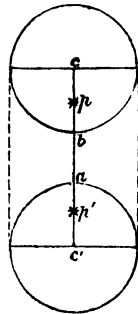
Now all this time the indications of the electrometer  $x$  connected with the charged plate  $A$  are directly as the square of the accumulated quantity and the square of the distances between the plates, Exp. 36, 37. If then with the quantity 1 we suppose the distance between the plates to become twice as great, the electrometer intensity will be 4 times as great; that is, if it before indicated 4 degrees, it will now indicate 16 degrees (Exp. 37): suppose at this instant we could take away one-half the electricity, that is, reduce the original quantity to one-half, the electrometer will then show the same intensity as at first, that is, 4° (Exp. 37). Hence it follows that the charge which can accumulate under a given intensity of an electrometer connected with the charged side, is directly as the distance between the opposed coatings, that is, the thickness of the dielectric stratum.

Now these results are quite in accordance with the deductions arrived at by the learned and ingenious Mr. Cavendish by a very different and totally distinct method of inquiry, so long since as the year 1775, who states, in the 66th volume of the 'Philosophical Transactions,' that "the quantity of electricity which coated glass can receive under the *same degree* of *electrification* is as the area of the coating directly, and as the

thickness of the glass inversely." But the conditions of the two plates *A B*, fig. 42, are precisely those of the Leyden jar, the only difference being in the dielectric medium, which in this case is *air* instead of *glass*; and although Faraday has shown that in respect of all such media there are differences of inductive capacity, yet for the same medium the specific inductive capacity is constant, and consequently the laws of electrical accumulation and attraction, as determinable in that medium, would not be far different from the laws in any other medium.

102. We may from these results determine, under common circumstances, the attractive forces between electrified conductors of any form, by considering the opposed portions as limiting conductors to an intermediate dielectric medium, as in the case of the Leyden experiment; in which case the force of attraction will be as the extent of the opposed areas directly, and as the squares of the distances inversely (100). Take for example the case of two spheres, a problem which has hitherto greatly perplexed Philosophers and Mathematicians, considered upon the theory of electrical distribution. We may here, by a simple mathematical analysis, assign two points *p p'*, fig. 44, within the surfaces of the opposed hemispheres in which we may conceive the whole force to be collected, and to be the same as if proceeding from every portion of the hemisphere. The position of these points will depend on the variable distances *a b*. Now the force between the two spheres under induction, tending to pull them together, will be in the inverse ratio of the squares of the distances *p p'*; that is to say, if we double this distance, the force will be reduced to one-fourth; if we treble it, to one-ninth, and so on. It will also be as the distance *a b* multiplied into *c c'*; that is, as the distance between the nearest points multiplied into the distance between the centres of the spheres.

Fig. 44.



*Example.*—Two conducting spheres, each an inch radius, being applied to the scale-beam electrometer, as in the case of the planes, fig. 43, page 100, the beam lifted with a weight of 6 grains of resistance, when the distance  $ab$  was  $\cdot 5$ , and distances  $pp'$   $1\cdot 11$ ,—the distances being measured in inches and tenths,—on increasing the distance  $ab$  to one inch, the distance  $pp'$  becomes increased to  $1\cdot 73$ : at these distances a resistance of  $2\cdot 5$  grains was required to balance the beam. When distance  $ab$  was increased to 2 inches, distance  $pp'$  becomes  $2\cdot 83$ : in this case the resistance was reduced to  $\cdot 75$  of a grain. In all these cases, therefore, the force varies exactly according to the law just stated; that is to say, the forces 6,  $2\cdot 5$  and  $\cdot 75$  are inversely proportional to the squares of the distances  $1\cdot 11$ ,  $1\cdot 73$  and  $2\cdot 83$ , as also to the simple distances  $\cdot 5 \times 2\cdot 5$ ,  $1 \times 3$  and  $2 \times 4$ .\*

Professor W. Thomson, of Glasgow College, in a most interesting communication to the Physical Section of the British Association assembled at Oxford in June, 1847, placed this problem in a very elegant way, under the principle of optical reflections,—considering that in every case of electrical attraction, certain reflections or electrical images of force, as it were, may be conceived to be produced within the opposed conductors, which in the case of two oppositely charged bodies were continued in infinitum, much in the way of reflections between two mirrors; and by a rapidly converging series obtainable for such cases he considers the problem to come perfectly under the dominion of analysis.

103. The laws of electrical repulsion were first submitted to rigid experimental examination by the celebrated French Philosopher Coulomb, who showed by the Balance of Torsion already described, page 81, fig. 36, that two small spheres, similarly charged, repulsed each other, in certain cases, with a force inversely proportional to the squares of the distances be-

\* A more detailed analysis of this problem will be found in the 'Philosophical Transactions of the Royal Society' for 1834, p. 240.

tween their centres. Thus the re-active force of torsion (84) at distances  $36^\circ$  and  $18^\circ$ , were 36 and 144; so that with a decrease of distance to one-half, the force had increased in the proportion of 1:4. The electrical inquiries of Coulomb have deservedly engaged the attention, as they have challenged the admiration of scientific men: Biot, Poisson, and other of the French Mathematicians, rest their mathematical theory of electricity entirely upon his researches.

In prosecuting this subject experimentally, Coulomb has mainly directed his attention to an hypothetical distribution of electricity upon the surfaces of bodies, where it is supposed to be confined in a stratum of greater or less density by atmospheric pressure, and to exist as it were within a hollow vase of air of the form of the body. He endeavours to deduce experimentally the general laws of this hypothetical distribution, by touching electrified bodies with a small insulated carrier plate, termed the 'proof plane,' and which he subsequently transfers to the balance of torsion. This plane, on removal from the electrified body, is considered as an element of the surface, and to be in all respects identical with it: the laws of electrical distribution are given for spheres, plates, cylinders, and bodies of various forms, and the proportion in which a charge is shared between such bodies determined, and the thickness of the supposed electrical stratum at different points of these bodies assigned. This stratum in cylinders and plates is greatest and most dense towards the extremities or edges: at the point of contact of two unequal spheres it is nothing. In the extreme opposite points the ratio of the density increases in proportion to the diminished size of the lesser sphere, but it never surpasses a given limit. When the spheres are separated, the limit of ratio of the densities of the stratum in each is  $\frac{5}{3}$ .

The advance of modern researches certainly renders the views of Electricity entertained by the French Mathematicians somewhat questionable. It is not that they cast the least doubt on the intellectual ingenuity and profound thought of the great

Experimentalist upon which their particular theory is built, but only on the hypothetical evidence upon which some of the experiments are based. It is very doubtful whether the proof plane can be altogether considered as an element of the surface of an electrified body to which it is applied,—whether it fairly represents the actual amount of electricity accumulated in that point; and if it did, whether the laws of repulsive forces are so uniform and invariable with all charges, and at all distances, as to enable us to deduce therefrom the ratio of the intensities. It is well known that proof planes of variable thickness come away charged from the same point of electrified bodies in different proportions,—the power to take up electricity being entirely determined by the induction of which it is susceptible: it can be further shown, that notwithstanding the presence of similar or dissimilar electricities, inductive forces tend to arise in two opposed conductors, similar to those which would arise supposing one of the bodies in a neutral state, thereby, with certain charges and at certain distances, decreasing the amount of the repulsive force, and disturbing the law of action: all question of the accuracy of past deductions, therefore, upon the evidence of the proof plane, is not altogether unpardonable; nevertheless, the beautiful memoirs of the French Philosophers, and of Coulomb especially, upon this branch of Physics, will be always regarded as the most splendid efforts of genius and intelligence.

104. *Tension and Intensity.*—It may be proper, before closing our remarks on electrical actions at a distance, to advert to two terms frequently employed in Electricity, and which have given rise to some considerable discussion as to the sense in which they are to be received: these are the terms ‘electrical *tension*’ and ‘electrical *intensity*.’ Now with respect to the signification of such terms, inasmuch as they are abstractedly considered as mere signs of ideas, they may be taken to represent any previous definition we may think proper to assign to *them*. *Their particular and direct application, however, to the*

representation of electrical actions, is tolerably clear and determinate.

The term *tension*, in its general acceptation, applies to the case of re-active or resisting force, however derived, whether to the re-active force of an elastic fluid, such as air heaving out under compression, or to the re-active force of a strained or twisted wire, as in the instance of a stretched musical string, or a wire employed in a balance of torsion (84). In either instance there is a force set up in these bodies by which they tend to recover their normal or quiescent state; and the amount of this force is virtually the tension or degree of suffering to which they are exposed. If we conceive, therefore, for an instant, according to the French theory, that electricity is a certain force exerted by an elastic fluid, capable of compression, then, like any other elastic fluid, such as steam or air, it would exhibit a certain amount of tension or re-active power; and this would be as the density or the number of particles confined in a given space: such would be the signification in this case of the term electrical tension.

But the term may be also and equally well applied to the condition of polarized molecules of a dielectric interposed between two limiting conductors, as represented in fig. 41, p. 93, and to the state of induction generally. In this case it expresses the re-active force of the particles constrained to assume a new condition in their electrical relations, and the amount of suffering they endure in their forced deviation from their normal or quiescent state. The higher the companion and separated powers are exalted, the greater will be the degree of tension they endure. In a similar way, the lateral or transverse force of dilatation upon the representative line of induction is a kind of lateral tension or stretching of these forces, tending to throw the particles asunder, all of which may be conceived to increase up to the limit of the power of endurance: all this is fairly expressed by the general term *tension*, and which term thus becomes representative, either of the particular condition of the electrical agency itself, or of the



re-active state of the inductive force of the molecules of a dielectric when charged by induction.

Now the term *intensity*, although of the same class, is still of a somewhat different character from *tension*; it rather applies to the degree or amount of resistance: it would be no superfluity of language to speak, for example, of the *intensity* of the tension or re-active power, as indicative of its greater or lesser amount in degree; just as we say, the *intensity* of the heat of the sun, the intensity of light, &c. In its application, however, to ordinary electrical phenomena, it has a proper and marked position assigned to it, being peculiarly expressive of the activity of an-electrified conductor, as shown by an electroscope or electrometer connected with it, as in the case of the electrometer  $\mathfrak{E}$ , and the opposed plates  $\mathfrak{A}$   $\mathfrak{B}$ , fig. 42, p. 97, and other instances which have been given of the application of electroscopes and electrometers. Thus the charge communicated to a jar or battery may be taken in terms of the quadrant electrometer, or any other indicator, which may be said to express the *intensity* of the charge, in which sense we speak of the jar being charged to a given *intensity*; but what renders this term as particularly necessary and distinctive, is the fact that the activity, or what we have called the intensity of the charge, thus exhibited, is as the *square* of the *quantity* of electricity accumulated (99); whereas the *tension* or force on the dielectric particles actually constituting the charge itself between the limiting conducting surface, is as the quantity only, as is also the *tension* of the electrical agency itself when restrained to a given space. In the case of charged glass or other dielectric, the electrometer indicates the activity of the uncompensated electricity, or the free action, as it were, of the charged surface. This is one thing; but the *tension* or degree of power in the molecules of the intervening dielectric, tending to break down the induction by a species of mechanical violence, as in the case of a fracture of a charged jar, is another; and hence the two terms, tension and intensity, are under these *limitations fairly and distinctly separable.*

## V.

### ON ELECTRICAL DISCHARGE.

Various Forms of Discharge—Laws of Disruptive Discharge—Length of Electrical Spark—Influence of Pointed Bodies—Brush and Glow Discharge—Theoretical Views of Disruptive Discharge, and the Action of Points—Convective Discharge—Discharge by Conduction—Theory of the Action of Conducting Bodies.

105. The return of a charged system of electrics and conductors (59) to its normal or quiescent state constitutes electrical discharge, and is a phenomenon directly opposed to insulation: this return may be effected in various ways, giving rise to very different effects, and constituting different kinds of discharge.

The most palpable and violent form of electrical discharge is that which has been termed 'disruptive discharge,' in which the particles of the intervening dielectric become more or less displaced, and the electrical polarization of the molecules, fig. 41, p. 93, raised to a degree past endurance; so that the forces re-combine with a sort of convulsive effort, causing a powerful extrication of light and heat, and a most irresistible expansive power. The common electrical spark, as drawn from the conductor of the electrical machine, and the dense concentrated explosion of the Leyden jar, between the balls of Lane's electrometer, furnish good illustrations of this form of discharge. Now it is of little consequence to the immediate result, whether this takes place either *directly* between the limiting conductors of the system, as in the case of the breaking down the intermediate dielectric medium itself, or whether it occurs in some other direction between these limiting conductors, as in the case of an *exterior* circuit, such as produced in applying the discharging rod to the discharge of the *electric jar* (60). In either case the polarized con-

ditions of the intervening glass vanishes, in consequence of the neutralization of the forces sustaining the induction. In the case of the exterior circuit, the return of the intermediate dielectric molecules to their normal condition is generally effected without any violence; but even here we find an occasional participation in the mechanical effect by which the jar, at the instant of the discharge, is as it were forcibly penetrated in some point between the coatings, and a fracture is the consequence.

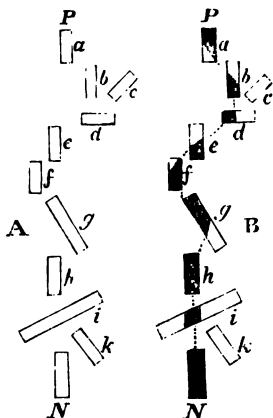
106. The laws of disruptive discharge are marked and decisive. If the insulation immediately between the limiting conductors sustaining the induction be sufficiently powerful, and the electrical forces allowed to determine their own path of discharge in some other direction, and in which the insulation is not equal to restrain them, then discharge ensues in that direction, and its course will be determined through a line or lines in which there is the least resistance, that is to say, in which there is the least amount of insulating power: thus, if a coated pane or jar, charged to a high intensity, be so circumstanced as to discharge as it were through a fortuitous admixture of electrics and conductors more or less perfect, the discharge will find out for itself a path of least resistance, seizing upon some bodies and avoiding others, according as such bodies happened to be convenient, or in any way useful in facilitating its progress,—a result arising as a necessary consequence out of the principle of induction, which ranges the tension (104) in a determinate way throughout the whole line or lines of discharge, and by exalting the electrical state of the particles in that line or lines beyond the tension of the adjacent particles, determines the course of the action; and this it is which gives to electrical discharge a sort of foresight, or, as it were, faculty of perception of the easiest course to be followed.

The following are instructive and important mechanical illustrations of this result:

*Exp. 41.* Let some detached fragments of leaf gold, a b c d,

&c., fig. 45, A, be laid down on paper in any casual and convenient way, producing with the paper a detached series of bad and good conductors: charge about 10 square feet of coated glass to a considerable intensity; for example, to the full intensity of Henley's electrometer (81), and having placed the extremes P N of the series between the wires of the universal discharger (78), allow the charge of the battery to find its path across the interrupted circuit: the result will be as represented in fig. 45, B. Portions of the leaf gold, unable to withstand the disruptive effect of the shock, will be burned up, thereby showing the course of the discharge; and it will be

Fig 45.



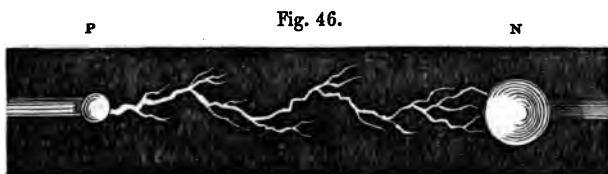
perceived by the black portions in the figure, copied from an actual experiment, that the line of inductive force  $P b d e f g h N$  is the shortest and least resisting possible line which could be traced between the points P N. It is here remarkable that not only are the fragments  $c k$  untouched, being, from their relative position, of no use in facilitating the progress of discharge, but even portions of the remaining pieces are also left out, and we have only the pieces  $h N$  totally destroyed, as being in all their parts completely essential to the course of the induction.

107. The distance through which disruptive discharge is obtained, as between two discharging balls or other conductors, has been called the striking distance. This striking distance, or length of spark, is very variable and very dependent, not only on the intensity of the charge, but on the form of the conducting bodies: the larger the conductors, the greater the electrical charge required to pass through a given distance.

for in this case, as before remarked, the intensity is diminished in the inverse ratio of the square of the surface (101). Now it is by exalting the intensity of the discharging point that the greatest length of spark is obtained: sparks from electrical machines are obtained of 10 inches or a foot in length, in rapid succession, amounting to a current. In this case it is usual to affix a ball of about two inches in diameter to the conductor, so as to project 3 or 4 inches from it, and then to present a large ball to this, either connected with the earth or with the opposite conductor. By crowding, as it were, upon the small ball, the inductive influence of the large ball, all the lines of which terminate upon it, and at the same time collecting upon this ball, as a point of discharge, a large proportion of the electricity of the conductor, the intensity of the small ball is greatly exalted, and it strikes through a longer distance, producing, by the resisting intervening air, a crooked or zig-zag spark. The striking distance of the electrical discharge in the case of a single electrical jar is usually confined to extreme distances of about an inch: by the peculiar arrangement, however, of a series of jars in the way shown in fig. 32, page 66, a long striking distance is obtained between the outer coating of the last jar and the knob of the first. Dove, at a meeting of the Royal Academy of Sciences at Berlin, in June, 1847, showed that the length of the spark by this succession of positive and negative surfaces varies with the square of the number of jars. Mr. I. Baggs, also, in a communication to the Royal Society, in January, 1848, describes a method of charging and placing the jars by which a disruptive spark of unusual length and brilliancy is easily produced. In this experiment the jars are each charged separately and to the same intensity, then quickly placed in series of positive and negative surfaces, very near, but not so as to touch.

108. The character of these disruptive sparks or flashes depends almost entirely on the form, area, and electrical intensity of the discharging surfaces, as also on the kind of *electricity on the conductor* in which the spark originates. If we

present a large smooth uninsulated metallic ball, of about 3 or 4 inches in diameter, to the rounded prime conductor of a powerful electrical machine,—short, brilliant, and perfectly straight sparks will pass between them, accompanied by a sharp snapping sound, but the ball must be brought near the conductor. If the same ball be presented to the negative conductor, the sparks will be far less powerful and dense; they will become small, and of a pointed character. If we now attach a smaller ball to the prime conductor, of about an inch or a little more in diameter, allowing it to project about 3 inches into the air, and present the large ball to this, much longer sparks are obtained, but less brilliant than before, and of a crooked or zig-zag form. In transferring this experiment to the negative conductor, the length of the spark is very considerably diminished; frequently it will not be above one-sixth of the length. Long sparks produced from the prime conductor in this way, or by the arrangement of jars just described, will be a foot or more in length: they are attended by lateral branches or divergences, frequently of a violet colour, presenting to the eye an extremely beautiful appearance. The annexed fig. 46 may be taken to represent this kind of dis-



ruptive spark, and which, especially in the long discharge by a series of jars, is very imitative of what has been termed forked lightning.

Electrical sparks are more brilliant between good conductors than between imperfect or less perfect conductors: hence metals are almost exclusively employed in cases where long or brilliant short sparks are required.

109. It has been proved both theoretically and experimen-

tally that the actual force in operation between the discharging surfaces at the instant of disruption is the same for all distances: thus, if the force between the balls of Lane's discharging electrometer be at the instant of discharge at a given distance considered as unity, or 1, and the balls be now supposed to be placed at *twice* that distance, then the force would become reduced in the nearest points of discharge to  $\frac{1}{4}$ th (100), since it varies as the squares of the distance inversely. If at this double distance, however, we double the quantity of electricity, then this force, reduced to  $\frac{1}{4}$ th by increasing the distance, is again raised up to unity as before (99), since it increases as the square of the quantity, and is hence, with twice the quantity, four times as great: the spark therefore will again occur: hence the influence of distance is only to vary the quantity of electricity requisite to produce a given force equivalent to cause a disruptive spark at that distance, but it does not affect the actual force at the instant of discharge.

110. *Influence of pointed bodies, and action of points in modifying disruptive discharge.*—When we continue to diminish the extent of surface originating a disruptive spark, and finally arrive at a small termination or point projecting freely into the air, most important and very curious results ensue: instead of a brilliant explosion, stars and brushes of light appear to arise or settle on the points, attended by currents of wind, whilst the distance at which very small balls or points are observed to operate is in many cases considerable: thus, in the instance of the great machine at Haarlem (46), a point appeared luminous at 28 feet from the conductor. If a short brass rod with a pointed rough end project from the prime conductor of a powerful machine, it will send out a full brush of luminous electrical rays, especially on presenting to it a flat imperfect conductor; and if a small ball be substituted for this projecting point, its surface, if the machine be powerful, will present a sort of phosphorescent luminous glow, apparently covering the whole surface. Faraday considers both these

phenomena as variations of disruptive discharge, and has not inaptly termed one, *brush* discharge, the other, *glow* discharge.

111. Although both brush and glow discharge have been pretty nearly identified as to cause, and as derived from the same electrical source, yet under ordinary circumstances they are in appearance totally distinct, and have a striking relation to the kind of electricity producing them: thus, according to the Franklinian theory, when electricity passes off a point, it is generally productive of brush discharge. When received on a point, the appearance is that of a glow or star or pencil of light. *Brush discharge* commences, as in the annexed figure 47, in a short conical brush root, which terminates in pale quivering ramifications, attended, when the discharge is powerful, by a subdued roaring sound, shown by Wheatstone to arise out of distinct and successive small explosions, and may be considered as an intermitting series of short sparks between metal and air, or between any good and bad conductor. The discharge always commences at the root of the brush, and is complete at the point of the conductor before more distant particles of air arrive at the same degree of tension (104): hence the discharge is progressive.

Fig. 47.



*Glow discharge*, on the contrary, is a more quiet and almost perfectly continuous result, depending on the charging of portions of air in contact with the surface of discharge. By diminishing the pressure, the glow can be caused to pervade a large extent of surface. Brass balls of 2 inches in diameter will become covered with a luminous glow when exposed to the action of a powerful machine under a reduced atmospheric pressure of about 5 inches of the mercurial gauge. The essential difference between brush and glow discharge appears to be in the kind of action upon the particles of the dielectric medium. In the brush discharge, these particles are operated on by a momentary intermitting action,



whereas the glow is a constant renewal or permanence of the same action without stop. In either case the particles of the air or other dielectric in contact with the conducting surface continue to be charged, and have their electric tension highly raised. We may by certain artificial arrangements convert the one into the other: any circumstance which tends to facilitate the charge of the air, and preserve at the same time the degree of tension of the dielectric particles, produces glow; whereas by resisting the charge of the particles so as to favour previous accumulation, and a consequent sinking of tension by discharge, we produce intermissions or brush;—thus rarefaction of the air, or the presentation of a pointed conductor, favours *glow*; whereas condensation of the air, and the presentation of large surfaces, will convert the glow into a brush.

112. Franklin first noticed the influence of pointed conductors on electrically charged bodies; he showed, that when presented to them, their charges became dissipated even at considerable distances, and with surprising rapidity. He charged an iron shot about 4 inches in diameter, and observed that on presenting to it an uninsulated pointed needle its attractive force on a small thread immediately ceased. He further observed that this influence of pointed bodies was also exerted when projecting from the charged body itself: the charge became rapidly dissipated by the projecting point; at the same time a current of aerial particles set off from the point, capable of impressing motion upon light models moveable upon a central axis, and fitted with vanes. This current has been termed the 'electrical aura' or gale. The re-active force of this current upon the point itself, and from which it appears to flow, is so great as to give the point motion in a reverse direction when free to move. The following experiments are very illustrative of these curious and important facts.

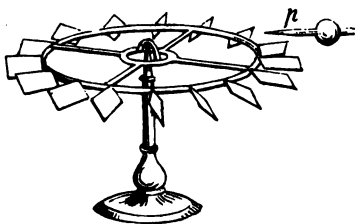
*Exp.* 42. Place an uninsulated metallic ball within about 2 inches of the prime conductor of an electrical machine, and whilst a series of strong sparks continues to pass upon

it, present a pointed wire to the conductor; the sparks will instantly cease, even when the point is at more than twice the distance of the ball from the conductor, where it will in the dark appear as a star of light.

*Exp. 43.* Attach the pointed wire to the prime conductor itself; sparks can be no longer obtained upon the uninsulated ball, whilst a divided brush of rays will appear to issue from it, and the electrical effect will be transmitted to a point at a still greater distance.

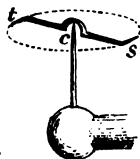
*Exp. 44.* Fix a series of card-paper vanes on the circumference of a light wheel, as in fig. 48; poise the wheel on a central point, and expose the vanes, which should be set a little oblique to the plane of the wheel, to a current of electrical air proceeding from a charged point *p*, projecting either from the positive or negative conductor of an electrical machine. When the machine is set in motion, the poised wheel will turn upon its centre.

Fig. 48.



*Exp. 45.* Bend the extremities of a light pointed wire, *sct*, fig. 49, at right angles to the wire, but in opposite directions: let the whole be poised on a central point, and placed in communication with either conductor of the electrical machine. When the machine is put in motion, the bent wire will fly round in a direction reverse to that of the points *s t*, and will appear in the dark as a ring of light, in consequence of the luminous particles of air receding from this point.

Fig. 49.



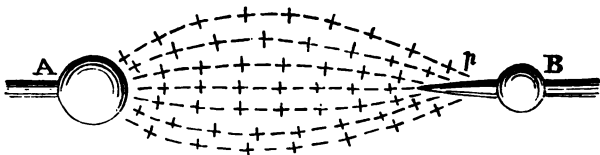
In all these experiments it is requisite that the point project freely into the air. If it be sheltered by being beneath the surface, then its electrical effect as a point ceases.

113. *Theoretical views of disruptive discharge and the action of pointed bodies.*—In the theory of induction (95) it is assumed that the particles of the dielectric are in a certain state of tension which rises higher and higher in each particle as the induction is raised higher and higher, either by the closer approximation of the inducing surfaces, increase of the charge, or variation of form, &c.: the sustaining of this tension (104) constitutes insulation; and when the tension surpasses the insulating power, the grand close of the existing phenomena is disruptive discharge. According to Faraday, the peculiar condition of the molecules of the dielectric necessary to the induction and insulation is equally essential to the final phenomena of discharge by disruption of the intervening dielectric medium. The theory does not assume that *all* the particles are equally affected as to tension; discharge occurs not when *all* the particles have attained a certain tension (104), but when the tension of a particular particle upon which the whole of the equilibrium depends has been pressed beyond endurance, and hence gives way: all the particles in this case must give way also; because, being all associated in the induction, it is the sum of the whole resistance which constitutes the equivalent insulation. Now the insulation is at the greatest possible endurance when aided by this single particle: if the assistance, therefore, of that one particle be withdrawn, the whole barrier is overturned. This happens more readily at a diminished distance between the limiting conductors A B, fig. 42, page 97, since there are fewer intermediate particles to resist the tension of the polarized state; and conversely, with an increased distance, the tension is more completely sustained: hence the law of Lane's discharging electrometer (89); for it is evident that the number of intermediate dielectric particles uniting their forces to sustain the insulation diminish and increase with the distance; so that the quantity of electricity to produce discharge is in the same ratio: hence also the law of surfaces relative to the quantity of electricity completing a charge (101).

By expanding the charge over a greater extent of surface, the number of molecules employed in sustaining the induction and insulation is increased, and hence the tension upon the previous particles intermediate between the lesser surfaces must be lowered. Conversely, by diminishing the surfaces of the limiting conductors (fig. 42) we throw the resistance upon fewer intermediate particles; disruptive discharge, therefore, with a given quantity of electricity, is less likely to happen with large opposed surfaces than with small, as shown in the phenomena of tension and intensity (104). The spark-striking distance, &c. (108), will therefore on this theory be dependent on the discharge of a few particles of the dielectric occupying a very limited space, in consequence of which the polarized inductive state of the whole series is lowered, and the molecules return to their previous or normal condition in the inverse order in which they left it; whilst their powers to propagate or continue the *discharging* operation from the point where the subversion of the insulation first occurred become now united. A good mechanical illustration of this may be derived by standing a series of thin rectangular pieces of wood upright, and near each other: if we overturn one at the end, the next, the next, and so on, must follow, and each in succession becomes pressed upon by the forces of all the preceding, which now unite to complete the downfall of the whole series in a given direction. The few particles originating the discharge are generally next one of the terminating conductors: in this point of subversion, however, they are not merely pushed aside, but they assume for the time an extreme tension, and the powers discharge throughout the series with violence and explosive force: the ultimate effect is the same as if we had put a discharging wire in place of the dielectric particles, and operated by conduction immediately between the limiting conducting surfaces A.B, fig. 42. The tension of the particles of the dielectric next the points in the limiting conductors being greater than those in the middle of the series, it is hence in these points that the disruptive effect commences; so that when

these conductors terminate in mere points or small surfaces, the tension upon the particles of the dielectric in contact with them is excessively increased; in fact, all the lines of inductive force may be supposed to concentrate upon a pointed conducting body, thus: let *A*, fig. 50, be the terminating spherical

Fig. 50.



surface of an uninsulated conductor, and *p* a point projecting from an opposed charged conductor *B*; then the lines of inductive force will concentrate, as it were, upon the point *p*, as represented in the figure: the point therefore becomes the source of an active mechanical force, and preserves its predominance over the other portions of the conductor behind it by a continued discharge of the accumulated electricity: hence currents of wind arise by the recession of the charged particles of air, and which are in every way favoured by the shape and position of the rod immediately behind the point. If the point be more or less central to the walls of a room without any more immediately opposed conductor, or be exposed to the induction of any other substance in its vicinity, still the same result ensues, since there is no distance so great as to limit the operation of this inductive action. The theory applies, by the converse of this, to an uninsulated point opposed to a charged body.

114. *Convective Discharge*.—In the production of currents by pointed conductors, the dielectric aerial particles necessarily carry away with them the electricity of the charged body, and so, by neutralizing the oppositely induced force in some other distant body, complete a discharge of the accumulated electricity. This species of discharge is not a communication of force, but a carrying away of force, as it were: the particles in this

case do not remain in place,—they travel. Faraday has hence termed this species of action ‘convective discharge;’ and this term is applied to every case in which discharge is effected by the transmission of electrified particles of matter, whether conducting or non-conducting.

115. *Discharge by Conduction.*—When the particles of an intervening dielectric *communicate* their forces, and lower the tension of the charged stratum, we have then discharge by conduction, or, as termed by Faraday, *conductive* discharge. This kind of power is common to all substances: the question is a mere question of time. In some substances, such as the metals, this communication of force takes place with extreme rapidity; in others, such as air, shell lac, &c., the process is difficult and slow,—so slow as to admit of such substances being considered as insulators. Conduction and insulation approach, therefore, very near each other, and arise both out of one common condition of matter. It is this kind of discharge by conduction which allows of the charge penetrating the particles of a solid dielectric, and causes in the charging and discharging of coated glass a spontaneous renewal of the charge. Faraday has given a very elegant and beautiful practical illustration of this kind of discharge. The substance spermaceti is found to be a dielectric through which induction can take place,—that is, its particles may become polarized; but it is also a very slow conductor, even when the electric force has travelled, as it were, through it to some distance: by communication of force from particle to particle we can, by removing the inductive or constraining power, cause it to return, as it were, upon its path, and re-appear again in its former position. This may be effected by giving two conjoined plates of spermaceti metallic coatings, one on each opposite exterior surface, and then, after charging and discharging the system, separating the plates, and examining their electrical condition. In this case, although *previously* to the separation after discharge, no kind of electrical indication could be obtained externally, yet, after the separation,

one half exhibits positive electricity and the other half negative electricity; so that on removal from each other's *inductive* influence, the two forces re-appear upon the surface under the attached coatings. The action, therefore, of an insulating dielectric, as in charged coated glass, is ultimately the same in promoting discharge, as that of the wire which discharges it.

116. *Nature of Conduction.*—We obtain from these considerations some little insight into the nature of conduction and the action of conducting bodies (9). These bodies are subject to the general laws of induction through contiguous particles in common with electrics, by which they are brought into a state of tension or polarity; but being in this state, the particles communicate their forces and promote discharge so rapidly, that the state of tension or polarity vanishes as soon as induced: hence the comparative differences in insulating and conducting power, which admit of one class of substances being considered as insulators, and another class as conductors (12). All substances promote discharge by the communication of forces, but the capability of this action, in a *greater* or *lesser* degree, renders them better or worse conductors—worse or better insulators.\* Thus, contrary to what might have been anticipated, insulation and conduction stand side by side as kindred phenomena.

117. The progress of electrical discharge by conduction through metallic or other substances involves the idea of velocity, and hence attempts have been made at various times to determine the rate of motion. We have already noticed the more early attempts of Watson and other members of the Royal Society, in 1748, to determine the velocity of the ordinary electrical discharge from coated glass, and their failure in deducing any thing like a numerical value of it. In more recent periods, however, this important question has been investigated experimentally with more success. Wheatstone, in 1834, by a beautiful and conclusive series of experiments, showed that the

\* Faraday's 'Electrical Researches.'

velocity of an electrical discharge through a wire of half a mile in length was at the rate of 576,000 miles in a second of time. This fact was deduced by catching in a mirror, whilst revolving on a horizontal axis at the rate of 800 times in a second, three electrical sparks produced by the discharge of an electrical jar in an interrupted circuit, the interruptions being at each end and in the middle of the conducting wire. Now it was observable that the centre spark fell out of the line of the other sparks by half a degree of the circle, and had hence experienced retardation, from which it was not difficult to compute the time of discharge through the wire; for the angular motion of the image being by an optical law double that of the mirror, the time of motion through half a degree is given, when the time of a whole revolution is known. By this process it was found that the centre spark occurred later than the others by at least the one-millionth part of a second; it was nearly the 1,152,000th part of a second, giving a velocity of about 576,000 miles in a second, supposing the current to have passed from one end of the wire to the other, or 288,000 miles, supposing it to have traversed one-half the wire.



## VI.

### AGENCIES OF ELECTRICITY.

**Mechanical Effects—Agency of Electricity in evolving Light and Heat—Luminous and Phosphorescent Effects—Chemical Agency—Electrical Currents—Magnetic Agency.**

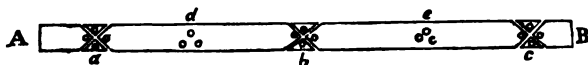
118. *Mechanical Agency.*—The transmission of electricity through substances is commonly attended by some mechanical effect: this is most apparent when the electrical discharge pervades bad and imperfect conductors; such bodies are not unfrequently rent in pieces; even good conductors, such as the metals, suffer expansion and other mechanical action to a considerable amount. If a heavy discharge be thrown upon a small metallic wire, it will become crippled, as it were, throughout its length; and when a similar discharge is passed through a capillary tube containing mercury, the tube will be shattered in atoms, such is the expansive force of the metal. In the passage of electricity through imperfect conductors, the particles become separated by expansion, causing a compression of surrounding particles. It is this which by the collapse of the air causes the sharp snapping sound of the electric spark, and which, in the discharges of a powerful battery, amounts to a stunning report.

When the electrical discharge is less sudden and condensed, a more progressive action ensues, and these violent effects become subdued, as we have seen in the case of glow and brush discharge, and the production of currents of wind (111); whilst in the conductive discharge by metallic bodies of large dimensions, as compared with the quantity of electricity transmitted, mechanical action is no longer apparent.

The following experiments are highly instructive, and illustrative of the mechanical agency of electricity in bad and *imperfect conductors*.

*Exp.* 46. A slip of tin-foil, *A B*, fig. 51, about 18 inches in length and half an inch wide, is attached by a little

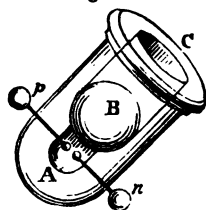
Fig. 51.



paste to the surface of a dry piece of wood, and small cross cuts *a b c* made on it with a sharp-edged knife; a few common wafers are placed over these cuts, and other wafers *d e* immediately between them on the continuous portions of the metal. When an electrical shock from a charged jar is passed from *A* to *B*, the small wafers will be thrown with violence off the disjointed portions *a b c* of the metal, whilst those on the continuous parts will remain undisturbed.

*Exp.* 47. *A C*, fig. 52, is a small mortar, turned out of hard wood or ivory, having two wires *p n* passed air-tight through the sides of the chamber *A*, and terminating in small balls placed within a short distance of each other. A ball of cork *B* is inserted immediately over the chamber, fitting loosely in the mortar, so as to avoid any considerable friction. When an electrical jar is discharged through *p n*, the ball *B* will be violently expelled by the expansion of the air in the chamber at *A*.

Fig. 52.



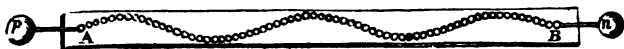
If the disruptive spark be produced within a drop of ether or water, these fluids will be converted into vapour, and the result is then more apparent: such is the expansive force thus produced, that few substances can resist it. Beccaria succeeded in fracturing to atoms a solid ball of glass of 2 inches in diameter by means of an electrical spark passed through a drop of water contained in a small cavity within the centre of the ball. Solid matter, such as stones, wood, loaf sugar, and other brittle imperfect conductors, are rent in pieces by an electrical discharge between wires placed within them.

The more progressive action of disruptive discharge through bad and imperfect conductors is frequently attended, as we have shown (112), by currents and other mechanical impulses capable of giving motion to light float wheels and other models delicately mounted on pivots.

119. *Agency of electricity in the evolution of heat and light.*—The discharge of electricity through insulating or imperfect conducting matter is invariably attended by an evolution of heat and light, to a greater or less extent. This is most apparent in the transmission of the electrical discharge through an interrupted metallic circuit, whether the circuit be continued through air or other gases, or through imperfectly conducting fluids: in air, brilliant and dazzling light arises, as in the case of the ordinary electrical spark; in water, a bright spark may also be obtained between wires placed near each other; in oil, alcohol, and ether, brilliant light is produced, attended by considerable heat. The great expansibility, however, of alcohol and ether, renders experiments of this kind somewhat hazardous.

The evolution of light and heat by electrical agency admits of the following simple and elegant illustrations :

Fig. 53.



*Exp. 48.* Attach a series of very small circular spots of tin-foil to the surface of a strip of glass, so as to be extremely near each other, and forming any kind of device, as *A B*, fig. 53: we have thus a series of metallic interruptions, in air, immediately over a solid electrical substance: if under these circumstances a current of electricity from a powerful machine be caused to pass between the positive and negative conductors *p n*, through this interrupted circuit, it will become beautifully luminous, especially if small sparks be taken on two balls *p* and *n* connected with the metallic chain

**A B.** In this way very brilliant pictures in luminous sparks of dazzling brightness may be obtained by means of disjointed tin-foil pasted on glass, and then covered with good insulating varnish: the foil should be laid on in narrow lines or strips, and then cut across with a sharp knife over the outline of the picture required, being so connected as to admit of a conduction of electricity throughout the total length, that is, by joining their alternate extremities, as represented in fig. 54. When small cuts are made in these

Fig. 54.



lines, representing any required figure, and electricity transmitted from balls *p n* at the extremities of the series, the whole figure will appear in dazzling light. Small dots of tin-foil attached in a spiral form round a tube of glass produce a brilliant effect, and may frequently be caused to extend over several feet in length: a series of metallic beads strung on a line of silk (41) has also a singularly beautiful appearance.

*Exp.* 49. Insert two wires in a ball of ivory or box-wood, and pass the charge of a small Leyden jar through its centre: the ball will become for the time luminous, and appear of a crimson or scarlet colour.

Oranges, apples, and other fruits, as also eggs and sugar, may be rendered luminous in this way. The universal discharger, already described (78), is well adapted to such experiments.

The tendency of electricity to evolve light in interrupted circuits is such that even apparent contact does not prevent it. If the shock of a jar exposing about 4 square feet of coated glass, fully charged, be passed over a small iron chain, luminous and brilliant scintillations take place at every link of the chain.

producing an extremely fine effect, especially when the chain is suspended in festoons upon insulating supports.

*Exp. 50.* Fill a common wine-glass with cold water nearly to the brim, and pour on its surface a thin stratum of ether; connect the water with the prime conductor of the electrical machine by means of a wire; put the machine in action, and draw a spark from the water through the ether by means of a brass ball or by the knuckle; the ether will be immediately inflamed.

Highly rectified spirit may also be inflamed by means of a powerful spark from the electrical machine, especially if it be gently warmed and placed in a metallic cup. The electrical spark will set fire in this way to various kinds of inflammable matter, such as resin, cotton wool, phosphorus, gunpowder, and other detonating compounds, all of which may by adequate arrangements be readily inflamed.

The firing of common gunpowder by electricity is best effected by placing a glass tube full of water in the circuit, so as to diminish the violent expansive effect of the spark, by which the grains are scattered without explosion: to prevent this, the gunpowder is usually enclosed in a cartridge, and two wires inserted within it; but even then the experiment is not always successful. When the discharge is transmitted through a few inches of water contained in a tube of glass, loose gunpowder placed between the wires of the universal discharger inflames instantly.

The heating effects of electricity upon metals and other good conductors are also very interesting and important. We have before shown (116), that some amount of resistance to electrical transmission always takes place, even in the best conductors, which resistance may be referred to the same source as the resistance through bad conducting matter, that is to say, momentary tension and insulation: hence similar effects in the progression of electrical action through their



51. Fix a strip of silver or gold leaf on paper, and

subject it to the shock of about 8 square feet of coated glass, fully charged; the metal will disappear with a bright flash.

When a powerful shock is passed through a slender iron wire, the metal becomes heated, and may be dispersed in red-hot balls, producing in the dark an extremely brilliant effect. All the metals, when drawn into fine wire, may be caused to burn in a similar way; and the most refractory, such as platina, gold, and silver, are converted into earth-like powders of various colours, termed *oxides*.

Low degrees of heat, elicited in metals by the transmission of electricity through them, may be estimated by names of the thermo-electrometer before described, fig. 40, p. 89.

From numerous experiments by Brook, Van Marum, and Cuthbertson, and subsequent experiments with the thermo-electrometer, it is found that the action of electricity in heating metallic bodies increases in the ratio of the square of the increased power, that is to say, as the square of the quantity of electricity discharged, without any relation to the intensity (104) or extent of coated glass on which the charge is accumulated: and since the resistance decreases with the number of conducting particles, it is further observed, that the heating effect of a given charge is in the inverse ratio of the square of the diameter of the wire.

120. The agency of electricity in evolving heat and light in bodies through which it passes is powerfully and wonderfully apparent in the discharge of the Voltaic battery alluded to in page 22, fig. 13. When an extensive series of plates, excited by an acid solution, discharges through points of charcoal, attached to stout wires connected with the opposite extremities of the battery, the heat and light evolved is most intense. With 2000 series of 4-inch plates, Sir Humphry Davy obtained an arched stream of light, of nearly 4 inches in length: fragments of diamond, on being introduced into it, disappeared; and thick wire of platina, one of the most refractory of the metals, fused readily: all the metals in thin laminæ, such as gold

and silver leaf, burned vividly: when fine iron or steel wire was made to join the opposite ends of the battery, it immediately ignited, and stout platina wire was kept at a white heat. The late ever-to-be-lamented Professor Daniell, by his new Voltaic battery, exceeded even these effects. With this battery the arc of electrical flame between points of charcoal was so intense and in such volume that the eyes of the spectators were seriously affected and inflamed, even though guarded by thick grey glasses: the Professor's face became scorched by the heat, as when exposed to a meridian sun: the rays, when collected into a focus, burned a hole readily through paper at many feet distant, and a bar of platinum of  $\frac{1}{8}$ th of an inch square, together with other highly infusible metals, such as rhodium, iridium, and titanium, were easily melted. Gold leaf burned with a vivid white light, and silver leaf with a light of brilliant emerald green.

121. We have not sufficient knowledge of the nature of the electrical light to say on what its presence depends; but from the fact that both the light and heat attendant on electrical action vary with the resistance to discharge, it has been supposed, and with much reason, that both are evolved from the medium in which the discharge occurs, by the mechanical agency of electricity in compressing their particles (118),—a result consequent on the same compressing effect produced in many other ways.

In condensed air the light is white and brilliant; in rarefied air it is divided and faint; in air highly rarefied it is not unfrequently of a violet colour. In a similar way the density of various gases has a material influence on the luminous effect: thus, in carbonic acid gas the light is white and vivid; in hydrogen gas, as in highly rarefied air, it is red and faint.

Electric light of great brilliancy exhibits, like the sun-light, all the prismatic colours when decomposed by the prism, and may be caused to display them separately by the intervention of different media. If a powerful discharge be passed between *two wires inserted* in a soft piece of deal, in the direction

of the fibres, the colour of the light varies with the depth of the points beneath the surface. If one of the points be inserted rather deeper than the other, all the prismatic colours appear. In this experiment the depth beneath the surface of the wood may be from  $\frac{1}{8}$ th to  $\frac{3}{8}$ ths of an inch.

122. *Phosphorescent Effects of Electricity*.—When the light evolved by the electrical discharge is very intense, a subdued luminous or phosphorescent effect continues to glow upon the surfaces of various bodies over which the discharge has passed. Calcined oyster-shells exhibit this effect in a high degree: selenite shines for a few seconds with a bright green light, and calcareous spar remains luminous.

The substances employed in these experiments should be placed between the wires of the universal discharger (78), and the eyes closed at the instant of discharge, so as to avoid the dazzling light of the spark. The luminous glow left on the surface of common chalk thus treated affords a beautiful illustration of the phosphorescent effects of electricity.

*Exp. 52.* Place a flat piece of dry chalk on the universal discharger, and set the pointed wires on its surface, at 2 inches apart: discharge a large electrical jar, fully charged, through the wires; a streak of light will remain on the chalk, which will continue for some time.

The particles of brittle substances, such as loaf sugar, when dispersed by a powerful shock, appear luminous for many seconds.

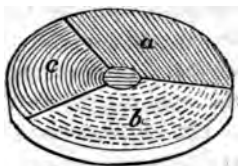
123. In all these and the preceding experiments, the duration of the immediate light of the spark is very limited and transient. Wheatstone, by viewing the reflection of electrical sparks in his revolving mirror (117), has clearly shown that the duration of the light does not exceed the one-millionth part of a second, so that objects in extremely rapid motion are seen by this evanescent light as if fixed and at rest. In evidence of this he has given the following very elegant experiment.

*Exp. 53.* A plane circular disc, fig. 55, having the three primitive colours,—yellow, red, and blue,—painted on it,



and occupying three proportionate sections of the disc *a b c*, was caused to revolve on a centre with extreme velocity, so that the colours, by an optical result, blended nearly into white, and were hence quite undistinguishable. The room being dark-

Fig. 55.



ened, and the light of a spark from an electrical jar allowed to fall on the disc, the three colours were as apparent at the instant as if the disc were at rest. The light had in fact been produced, and had again vanished before either of the colours had turned through a sensible space. The most rapid motion, therefore, producible by art is virtually rest, when taken in relation to the velocity of the electric light.

A very effective experiment may be made with a similar disc by merely placing on it three spots,—blue, red, and yellow,—one inside the other, and which, when the disc is caused to revolve, form three distinct circles. Now, however rapid the motion, the spots are seen, by the light of the electric spark, as being perfectly at rest; and if successive sparks be produced, they appear to be merely changing their relative position on the disc.

124. *Chemical Agency of Electricity.*—The agency of electricity in effecting chemical changes will be found of still greater extent and importance than its mere mechanical action, although probably intimately associated with it. Not only are the most refractory metals converted into oxides, but oxides already formed may be decomposed, and the metal revived and restored to its former condition. When a succession of small electrical shocks is passed through oxide of tin, placed within a clean tube of glass, the tube becomes stained with metallic tin. If we subject common vermilion, a compound of sulphur and mercury, to the charge of a moderate-sized jar, the mercury is easily reproduced in its metallic state. When the electrical spark is taken in various fluids, the fluids

may be decomposed and separated into their constituent elements. Thus water, by the action of the electric spark, may be converted into hydrogen and oxygen gases. This very important fact was first announced by some Dutch chemists, and was subsequently confirmed by one of the most distinguished of British Philosophers, Dr. Wollaston. Compound gaseous bodies, likewise, are chemically acted on in a similar way: thus it was observed by Priestley and Cavendish, that when a portion of common air was exposed for a considerable time to a succession of small electrical discharges, the bulk of the air became diminished, and its constituents, oxygen and nitrogen, combined in the proportions required to produce nitric acid, which was accordingly found in the vessel in which the air was confined.

125. The most powerful and available source, however, of this electro-chemical agency is found in the series of Volta already referred to (25). When substances are subjected to the continuous electrical discharge of this apparatus, very few compounds resist its power. As a general law of the decomposing agency, it is found that oxygen and its compounds, acids, &c., are determined upon the zinc or positive end of the series; alkalis, hydrogen, and other inflammable matter, upon the copper or negative end. Alkalis being thus found determined to the negative extremity, were suspected by Davy to contain an inflammable element, and he finally succeeded in realizing this conjecture: by means of a powerful series, potassa and soda were resolved into two constituent elements, viz. oxygen gas, which appeared at the zinc extremity, and a highly inflammable kind of metal, which was rendered up at the copper or negative extremity,—a discovery which shed an extraordinary degree of lustre on British Chemistry.

In the electricity of the Voltaic battery, therefore, we possess a most powerful chemical agency, by which the elements of bodies may be separated and transferred to distant points. The following experiment is another fine example of this important fact.

*Exp. 54.* In the centre glass *c* of the three glass vessels *A C B*, fig. 56, is a solution of sulphate of potassa, one of the neutral salts: *A* and *B* contain a light infusion of the blue colour in distilled water, which is highly sensible of the presence of alkali, or acid, its colour being immediately affected thereby: the three glasses are connected by filaments of moistened cotton, as shown in the

Fig. 56.



figure and wires *P N* from the opposite ends of the Voltaic series, terminating in gold or platinum points, inserted in the glasses *A B*. The result is, that the neutral salt in the glass *c* is soon decomposed into potassa and sulphuric acid; the acid is transferred to the positive cup *A*, and turns the blue infusion red, whilst the alkali becomes transferred to the negative cup *B*, and turns the blue infusion green.

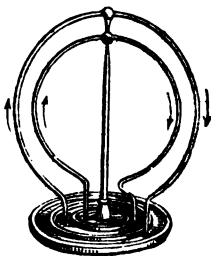
When metallic salts are subjected to this species of electrical action, they are speedily decomposed and the metal revived: thus a plate of silver connected with the negative wire of the battery will, when plunged into a solution of sulphate of copper (blue vitriol), become coated with metallic copper directly we immerse the positive wire *P* in the solution.

126. *Electrical Currents.*—Another important agency of electricity consists in its progressive, or, as it is usually termed, current action, as exhibited in the phenomena of conduction and discharge. When we discharge an excited Voltaic series, or an electric jar, through any substance, whether metallic, moist conductors, or air or other elastic media, this progressive or current force always arises, and the substances transmitting the electricity exhibit extraordinary and very peculiar powers. Now it is an important feature of an electrical current, that the two electrical forces are every where in it; the current is in respect of these forces the same in every part: we have never a current of one force only, so stated by Faraday, it is quite indivisible, and may be

conceived of as an axis of power in every part of which the two forces are present. Electricity as a chemical agent appears to operate directly through the medium of this current force. Thus in the decomposition of water and other bodies an electrical current is established, sufficiently powerful to separate the constituent elements which then interchange the two forces, and operate in lowering the tension by a species of convective discharge. The fluid, whilst under the influence of the battery, may be considered as any other dielectric substance in a state of tension (95). Faraday has termed such substances as decompose in this way *electrolytes*. The determining action, he further shows, exists within them, and not at the poles or doors of the battery; the chemical force of the current being directly as the quantity of electricity transmitted.

About the year 1820, the celebrated French Philosopher Ampère discovered the mutual attractions and repulsions of electrical currents, and showed that metallic wires, if free to move, attract each other when transmitting electrical currents in the same direction, and repulse each other when the currents passing through them are in opposite directions. When two wire rings are freely suspended one over the other, as in fig. 57, so as to turn readily in a vertical plane, and electrical currents transmitted through them, through the intervention of a little mercury in which their extremities terminate, the rings will be observed to separate and turn round upon their pivots until the currents passing through them flow in the same direction. Faraday has since shown that wires transmitting electrical currents induce momentary currents in other wires near them: thus the phenomena of attraction, repulsion, and induction, are common both to statical and dynamical electricity, that is to say, to electricity both at rest and in motion.

Fig. 57.



127. *Magnetic Agency of Electricity*.—The agency of elec-

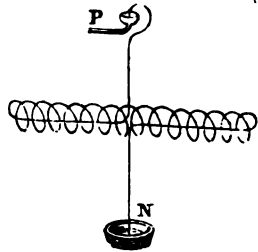
tricity in imparting polarity to iron and steel was observed by Franklin, and further verified by Van Marum with the great machine and battery in the Teylerian Museum at Haarlem (46). By this apparatus, pieces of watch-spring 6 inches in length were rendered powerfully magnetic: when the discharge was passed through the steel placed perpendicularly, the lower end acquired a north polarity, that is to say, if suspended in a horizontal position, that end pointed to the north; if placed horizontally in the magnetic meridian, the end pointing north acquired a north polarity, whichever extremity was connected with the negative side of the battery.

The most important discovery, however, in this department of Electricity is of comparatively recent date. This discovery, one of the greatest of modern times, is due to Professor Ørsted, of Copenhagen, who was led, about the year 1819, to investigate the peculiar condition of a wire uniting the terminating plates of the Voltaic battery, so as to form a closed circuit. The magnetic needle, when placed above or below this wire, termed by Ørsted a conjunctive wire, immediately deviates from its meridian according to certain laws, and tends to place itself at right angles to the wire. In all these deviations the pole of the needle over which the negative electricity enters turns towards the west, and the pole under which it enters turns towards the east. On further investigation, it appears that the transmission of the electrical current through the wire is attended by a transverse action exerted in a direction at right angles to the direction of the current, which action is always the same in kind and direction: this force in relation to current electricity is what lateral tension is to statical electricity (94).

The distinguished French Philosopher Ampère was amongst the first to pursue these new inquiries, and he succeeded in imparting to wires transmitting electrical currents all the properties of common magnets. When a ring of wire, suspended in a vertical plane and free to move, transmits a current of *electricity*, it places itself at right angles to the magnetic

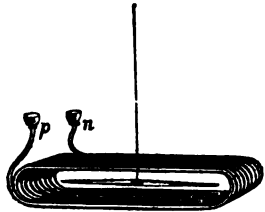
meridian, and becomes obedient to the attractive and repulsive forces of magnetized steel. Having investigated this magnetic agency of electricity experimentally, he concluded, from a profound mathematical analysis, that the condition of a common magnet was the same as that of a body about which electrical currents circulated in planes perpendicular to its axis,—a deduction confirmed by an extremely elegant and instructive experiment. A long copper wire covered with silk thread being turned into a helix, and the extremities returned through the axis and brought out at the centre, was freely suspended upon a fine point P, as in the annexed fig. 58. When an electrical current transversed this helix from P to N, the helix acted in every respect as an ordinary compass needle.

Fig. 58.



128. A magnetic needle delicately suspended and placed within a longitudinal coil of copper wire covered with silk thread, as represented in the annexed figure 59, becomes extremely sensitive of the least current transmitted through the coil from *p* to *n*; for the wire passing many times above and below the needle, tends to move its poles with the united influence of the whole, and in the same direction; so that the effect of a single wire becomes multiplied in nearly the proportion of the number of times the coil passes above and below the needle. A needle thus circumstanced, with a divided circle to measure the angle of deviation, constitutes an instrument termed a *galvanometer*, or, as it was first termed, *electro-magnetic multiplier*.

Fig. 59.



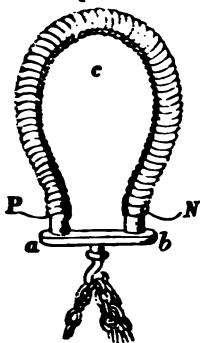
Faraday, by means of a delicate instrument of this kind, succeeded in identifying common and Voltaic electricity as a

source of electro-magnetic action. When a highly charged and powerful battery of about 25 square feet of coated glass was discharged through the galvanometer in an attenuated and rapid stream, by means of a pointed conductor and a moist hemp line, deviations of the needle were obtained, as in the original experiment of Ersted with the Voltaic series.

129. One of the most surprising results of the magnetic agency of electricity is the enormous power induced in soft iron by the circulation of electrical currents about its surface.

*Exp. 55.* Let  $acb$ , fig. 60, be a cylindrical bar of soft iron, bent so as to bring the extremities  $a$   $b$  near each other, and  $P$  and  $N$  copper wire covered with silk thread wound round its surface from  $P$  to  $N$ ; then, on uniting the extremities  $P$   $N$  of this coil with the terminating plates of a Voltaic battery, the iron immediately becomes magnetic, and will sustain by attraction at its extremities  $a$   $b$  a considerable weight when affixed to a square iron rod, joining the extremities or induced poles  $a$   $b$ .

Fig. 60.



By superposing heliacal coils in this way upon the surface of a soft iron bar, and uniting the extremities into one common terminator  $P$   $N$  on each side, temporary magnets have been produced capable of sustaining more than a ton weight; the magnetic power, however, is lost, or nearly so, directly the currents cease to circulate through the coils.

Small needles of hardened steel become permanently and instantly magnetized by placing them within helices of this kind, transmitting electrical currents: the polarity acquired at each end depends on the direction in which the current circulates, whether from left to right, as in the direction of the hands of a watch, or the contrary. "Let a person," observes Faraday, "imagine he is looking down upon the dipping *needle or north magnetic pole of the earth, and then let him*

think upon the direction of the motion of the hands of a watch, or of a screw moving direct, currents in that direction round a needle would make it into such a magnet as the dipping needle;" that is to say, the extremity towards the opposite end of the helix would be a north pole.

The magnetic, as in the case of the chemical and heating agency of electricity, is, *cæteris paribus*, directly as the quantity of electricity transmitted, without reference to the intensity.

By the operation of these powers, and the peculiar kind of tangential force they display, electrified wires and magnets are caused to revolve round each other, and a vast variety of most interesting and wonderful effects obtained, constituting a most beautiful department of science, termed *electro-magnetism*.

130. *Magneto-electricity*.—Magnetism having been thus found to arise out of electrical action, the conclusion was irresistible, that electricity should also be derived from magnetism. Although this conclusion lay dormant for several years, it was finally completely verified by Faraday in the course of the year 1831, being about twelve years subsequent to Oersted's celebrated discovery (127). When a piece of soft iron surrounded by coils of copper wire is brought into or removed from contact with the poles of a magnet, electrical currents are produced in the wire of considerable magnitude, as shown in the action of the magneto-electrical machine, fig. 14, p. 23. Hence has arisen a new and most valuable branch of electrical science, termed *magneto-electricity*; the essential principle of which consists in the converse operation of magnets, either permanent or temporarily produced in the common electricity of non-magnetic bodies.

131. The following is a remarkably elegant experimental illustration of magnetism induced by electricity, and the simultaneous and reciprocal excitation of electricity by the induced magnetism.

*Exp.* 56. Let two coils of copper wire, covered with silk



thread, be wound round opposite semicircles of a soft iron ring of about 6 inches in diameter, leaving a separating portion of the iron between the coils ; then, if the extremities of one of the coils be connected with the Voltaic battery, fig. 13, page 22, a current is established through that coil (126) which magnetizes the iron beneath (129). At this instant a current of electricity arises in the coil on the opposite semicircle by the flow of magnetism through the ring, and which, if the battery be sufficiently powerful, will cause a spark to appear between the extremities of this coil, if duly placed near each other. If put in connection with the electro-magnetic multiplier (128), the needle is violently deflected. It is, however, to be observed that this action is temporary, as in the case of the electro-magnetic machine, page 23, and exists only at the instant of making or breaking the contact of the opposite coil with the Voltaic battery.

132. A permanent electrical current may be induced by the direct agency of a magnet on metals by causing the metal or magnet to revolve near each other, and in the following way. Let a circular plate of copper be mounted on a horizontal axis, and its edge placed just within the poles of a powerful compound magnet, so as to revolve between the poles on a vertical plane. If we now cause the plate to revolve by turning round a winch attached to the axis, electrical currents will flow between the centre and circumference of the plate in the direction of the radii, that is to say, in a direction transverse to the direction of the motion of the plate. This may be made evident by attaching two wires by a spring pressure, one to the centre, the other to the circumference of the plate at the point immediately between the poles of the magnet, and bringing the opposite extremities of these wires in connection with the electro-magnetic multiplier (128): the pressure of the extremities of the wires against the plate must in this case be such as to insure a perfect contact without impeding the motion of the plate.

133. A similar result ensues in the rotation or motion of a

magnet near a metal at rest,—an effect admitting of most instructive illustration by the phenomena observed by M. Arago, in the year 1825, in the course of some experiments on the reciprocal action exerted between magnets and non-ferruginous bodies. This eminent French Philosopher found, that when a magnetic needle was caused to oscillate within a ring of copper, the amplitude of the vibrations became rapidly reduced, and the needle speedily reduced to rest; and also, that when a magnet was put in rapid rotation beneath a light copper disc freely suspended, the copper disc became dragged round with the magnet. Similar results were obtained when a copper or other metallic plate was caused to rotate rapidly beneath or near an ordinary compass needle: the needle first began to deviate from its meridian, and if the rotation was sufficiently rapid, was finally dragged round by the plate so as to acquire the same rapid circular motion.

134. All these effects were at first suspected to arise out of a temporary and vanishing magnetic impression made on the non-magnetic substance; but both Ampère and Arago considered the action as being always repulsive, and not such as would be likely to arise from any species of attraction: hence they deemed it to be a result of some new force as yet undiscovered. Such has subsequently proved to be the fact,—the force in question being the force of electrical currents induced in the metal by the magnet transverse to the direction of the motion. Thus, when a copper plate revolves near a magnet, electrical currents are induced, as just shown (132), between the centre and circumference, in the direction of the radii of the plate, the action of which is similar to that of currents produced in a wire joining the extremities of the Voltaic series (127): a transverse force is produced which causes both the magnet and plate to revolve in the same direction;—hence these bodies exhibit no action on each other when at rest, since the electrical currents then cease.

## VII.

### NATURAL ELECTRICITY.

Electricity of the Atmosphere—Lightning and Thunder—Meteors—  
Aurora Borealis—Water-spouts—Whirlwinds and Earthquakes.

135. *Electricity of the Atmosphere.*—Electrical science, in the early periods of its history, appears to have been limited to the phenomena of attraction and repulsion developed in certain bodies, by exciting into action some curious and hidden principle which such bodies were supposed to contain. No sooner, however, had the brilliant discoveries of the eighteenth century been achieved, than a far wider and apparently unlimited field of investigation presented itself. Electricity now became associated with those great and mysterious agencies upon which the natural operations of the material universe were supposed to depend. Dr. Wall, who, in the year 1705, observed the light, together with the crackling sound attendant on electrical excitation, became impressed with its miniature resemblance to the phenomena of lightning and thunder. Grey observes ('Philosophical Transactions,' 1735), that the electrical fire seems to be of the same nature with thunder and lightning. The Abbé Nollet, in 1745, speaks of "thunder and lightning being in the hands of Nature what electricity is in ours," and speculates on the probability of a thunder-cloud being an "electrified body, depending on the same mechanism as that of charged conductors." The great discovery of the Leyden phial, and the several investigations of Franklin, especially those relative to the action of pointed bodies, went still further in confirmation of these conjectures, and in the association of various luminous phenomena of the atmosphere with ordinary *electrical action*. Watson, in the 48th volume of the 'Philo-

sophical Transactions,' enumerates certain appearances recorded by the ancients, evidently depending on electricity in the air. Pliny tells us in his *Natural History*, that stars settle with an audible sound on the sail-yards of ships. Seneca describes the spears of the soldiers in the Roman camp as being on fire. The old historian, Herodotus, states that the Thracians disarmed the heavens of their thunder by launching their arrows into the air. Such phenomena are evidently nothing more than natural interpretations of the action of pointed bodies, and of certain forms of disruptive discharge (110), (111). Important coincidences of this kind could not fail to arrest the attention of those distinguished men who, about the middle of the eighteenth century, had surprised mankind by their great and novel discoveries in electricity, and to convince them of the prevalence of electrical agency in the ordinary operations of Nature.

In the year 1749, Franklin proposed two methods by which he supposed electricity might be drawn from the clouds; and at his suggestion an insulated pointed rod of iron, 40 feet long, was erected by Dalibard, in 1752, at Marley la Ville, near Paris, so as to project freely into the air. On the 10th May, 1752, electrical sparks were obtained from this rod, attended by the usual snapping sound. Franklin having noticed the great similarity between lightning and the ordinary electrical discharge, suggested the employment of pointed conductors in this way, for collecting electricity from the clouds and air. Lightning, he observes, is generally crooked and waving; it strikes the most prominent bodies, takes the readiest conductor, sets fire to inflammable matter, rends bodies in pieces, destroys animal life, and affects the magnetic needle; all of which are likewise effects of the ordinary electrical battery. His conviction, therefore, of the identity of the agency of lightning with that of common electricity amounted in his own mind to a reality. Tired of waiting for the erection of a tall spire in Philadelphia, upon which he had proposed to place a pointed conductor, this distinguished Philosopher, in June, 1752, found

a readier access into the higher regions of the air by means of a common kite. The kite had a pointed wire affixed to it, whilst the line of the kite was insulated by a silk cord connected with its lower end, and which terminated in a common key, thus made subservient to the purposes of an insulated conductor. After patiently awaiting the passage of several clouds over the kite, he had the unspeakable delight to observe some of the loose fibres of the hemp string bristle upwards and repulse each other (15); and finally, as the conducting power of the string became increased by the fall of rain on it (9), electrical sparks were drawn from the insulated key placed at the extremity of the string. Thus became realized, by actual experiment, one of the most beautiful and important discoveries in the history of science. Romas, in France, repeated Franklin's experiments, and, as stated by the Academy, according to an original conception of his own. In June, 1753, he raised a kite 550 feet into the air during the prevalence of thunder-clouds; the string had a copper wire round it, and was attached below to an insulated iron tube. The effects not only astonished but greatly endangered the spectators; flashes of fire, a foot long and 3 inches wide, passed from the insulated conductor communicating with the string, attended by loud reports heard at a distance of 500 feet, and produced the sensation of the spider's web (46) upon the faces of the spectators. Three straws, one of them a foot in length, were observed standing erect upon the ground towards the string: these at last began to dance up and down (13). At this instant a roaring noise ensued, similar to that of a forge bellows; the large straws now became violently attracted and repelled by the insulated tube, and three distinct, ringing, sharp reports ensued, similar to the sound of earthen jars dashed in pieces on a stone pavement. A spindle-shaped flash accompanied each of these discharges; the long straw now began to follow up the kite string with extreme rapidity, being sometimes attracted and sometimes repelled: it *ascended in this way* about 300 feet: the kite itself appeared

as if surrounded by a luminous cylinder 3 or 4 inches in diameter.

A celebrated Russian, Professor Richmann, of St. Petersburg, lost his life whilst prosecuting similar experiments. Having erected an insulated or pointed iron rod on the top of his house, he hastened there from the Academy of Sciences, in August, 1753, to observe the amount of charge communicated to a large quadrant electrometer (81) connected with the rod. At this instant, whilst stooping to examine the index, a large globe of bluish fire struck him on the head, and he instantly expired.

Saussure and other celebrated French Philosophers have subsequently made extensive experiments on the electricity of the atmosphere, and have shown that the air is generally more or less charged with electricity, either positively or negatively. Our talented countryman, Mr. Crosse, of Bromfield, near Taunton, in Somersetshire, has greatly contributed to our knowledge of atmospheric electricity, and has elicited some very important and general facts. His apparatus consists of more than a mile of insulated wire, extended between pointed conductors raised high into the air upon tall masts upwards of 100 feet in length. It appears, by experiments with this apparatus, that the electricity of the atmosphere has a daily flowing and ebbing period, like the sea, being found to increase and decrease in force twice every 24 hours. Generally, the electricity obtained from the air is positive, especially in clear weather; but during the fall of rain, fogs, snow, and storms, and especially during the passage of certain clouds, the apparatus is frequently electrified negatively. We are also in possession of numerous valuable observations made at the Kew Observatory by Mr. Ronalds, for the British Association. We are indebted to Mr. Ronalds for a high perfection of the means of observation, and, together with Mr. Crosse, for the following general results:

1. The electricity of the air is always positive,—fullest during the night,—increases after sunrise,—diminishes towards

noon,—increases again towards sunset,—and then decreases towards night,—after which it again increases.

2. The electrical state of the apparatus is disturbed by fogs, rain, hail, sleet, and snow. It is negative when these first approach, and then changes frequently to positive, with subsequent continued changes every three or four minutes.

3. Clouds also, as they approach, disturb the apparatus in a similar way, and produce sparks from the insulated conductor in rapid succession, so that an explosive stream of electricity rushes to the receiving ball, which it is wise to let pass off into the earth. Similarly powerful effects frequently attend a driving fog and heavy smart rain.

136. Electricity being thus intimately associated with meteorological changes in the atmosphere, it is thus further evident that thunder, and lightning, and several other meteoric appearances, are dependent on the ordinary operations of electrical agency. If we consider attentively the electrical conditions of a thunder-storm, we may observe in them all the elements of the Leyden experiment: the atmosphere, in fact, becomes a great coated pane or fulminating square (63), of which the charged cloud is the insulated, and the surface of the earth the uninsulated, terminating conducting planes;—the phenomena of thunder and lightning are neither more nor less than disruptive discharges through the intervening air, on the principles we have already explained (84), the magnitude of the effect depending on the tension. It has been well observed by Franklin, “If two gun-barrels strike at two inches distance and make a loud report, at how great a distance may 10,000 acres of electrified cloud strike, and how loud must be that crack?” We may further remark, that all the causes which operate in modifying the phenomena of the common electrified spark, also operate in giving variety to the phenomena of lightning. Thus lightning is often waving, or of a crooked zig-zag appearance: at other times it is straight and brilliant: when occurring near the observer, the light is quite intolerable *to the eye.* Sailors are accustomed to call the dividing zig-zag

spark 'forked lightning:' when it does not divide, it often affects the eye by a sort of ripple of light,—this they term 'chain lightning:' when a vivid spark occurs, but concealed from the eye by interposed masses of clouds, the light is so reflected from more distant masses as to illuminate the whole of the hemisphere,—this has been termed 'sheet lightning.' Arago, however, and other Philosophers, have given this term to electrical discharges spread out into broad alternated flashes, such as we observe in the flashes of a summer evening. The phenomenon termed 'globular lightning,' presenting an appearance of a ball of fire either in motion or at rest, is referable to glow discharge (111), which commences and proceeds for a short time previously to the more condensed disruptive discharge of the whole system. In this way balls of fire have been observed to roll along the sea and ground, or become stationary previously to a burst of thunder and lightning, according as the cloud upon which the discharge depends is at rest or in motion.

137. The noise called thunder may be referred to the collapse or mechanical and violent compression of the air by the disruptive discharge (105), and to the reflected or successive echoes of the sound reverberating between the opposed surfaces of clouds or land, and which has been called the roll of the thunder: thus, when a cloud covers the horizon, the noise of cannon fired at sea is often attended by a long-continued roll like thunder.

As the motion of sound is extremely slow compared with that of light, being not more than about 1000 feet in a second of time, whilst light travels at the rate of 190,000 miles in a second, we are enabled by neglecting the evanescent time of the light to calculate the distance of the point in which the disruptive discharge begins, merely by multiplying the number of seconds which elapse between the light and the sound of the thunder by 1090, the actual rate of motion of the sound in a second. Thus an interval of 5 seconds would give 5450 feet for the distance of the thunder-cloud from the observer, being rather more than a mile.



138. The effects of electrical discharge under the form of thunder and lightning are similar to those already noticed (118), the mechanical effects being of a most stupendous character. Wood and other resisting matter is rent, and scattered in all directions: nothing appears to stand against it: thus rocks are split open and scattered, and trees of enormous size, especially the oak, rent asunder. In November, 1790, the *Elephant*, of 74 guns, was struck at Portsmouth by an electrical discharge from the atmosphere, which entirely shook and shivered her mainmast, weighing about 18 tons: all the iron hoops and mouldings were burst open and scattered around: some of the hoops were half an inch thick and 5 inches wide: the mast itself consisted of a mass of wood 3 feet in diameter and 110 feet long.

139. *Meteors*.—All those ordinary meteors found on the masts and sail-yards of ships and other pointed bodies are, without any question, pure effects of atmospheric electricity depending on the action of pointed bodies projecting into electrified air (113). These appearances have been termed by Spanish sailors the ‘Fires of St. Elmo,’ and in superstitious periods were supposed to proceed from the body of the Saint. In the record of the second voyage of Columbus we find that “during the night of Saturday (October, 1493) Saint Elmo appeared on the top-gallant masts with seven lighted tapers.” The Italians refer these appearances to St. Peter and St. Nicholas: the Portuguese call them ‘Corpo Santos;’ hence, probably, the term used by English sailors of ‘Comazants.’\* The action of electrical points may hence be adduced in explanation of many of those luminous phenomena of the atmosphere found occurring near the earth’s surface. Other meteors, however, although suspected to originate in electrical action, do not appear to be so clearly referable to it. Amongst these may be classed the well-known phenomenon of the

\* ‘Tomlinson on the Thunder-storm,’ published by the Committee of General Literature and Education appointed by the Society for Promoting *Christian Knowledge*.

'shooting-star:' this appearance, however, has still been successfully imitated by passing the shock of an intensely charged jar between balls enclosed within the extremities of a long glass tube exhausted of its air. In the present state of our knowledge, however, of such meteors, we cannot pretend to decide on the question of their electrical origin.

140. *Aurora Borealis*.—The magnificent phenomenon termed Aurora Borealis, or Northern Lights, may be certainly classed amongst meteors depending on ordinary electrical action, and may be referred to the flashing of electricity through air more or less rarefied, and at variable heights above the surface of the earth. If a pointed conductor be caused to discharge electricity within an exhausted glass receiver, streamers of white and coloured light, and diffuse or pervading flashes, are produced, closely imitating the phenomenon termed Aurora Borealis. A receiver of 6 inches in diameter, and 10 feet high, becomes filled with light in this way under the action of a powerful electrical machine; the light and colour depending on the rarity of the air, the amount and kind of vapour it contains, and the substance and form of the conducting body from which the electricity is transmitted. By a careful management of the experiment, beautiful violet and red streamers, and splendid coruscations, together with diffuse glow, are easily obtained. By far the greater number of appearances occasionally observed in these latitudes, and referred to the phenomenon of the Aurora Borealis, may be traced to the presence of dense masses of electrical clouds yielding up electricity into the atmosphere above them. Both glow and brush discharge (111), diffuse and infinitely varied, and of a vast extent, is produced in this way, and is often observed to proceed from behind the masses of clouds in long shooting streamers, attended by a beautiful glow of diffuse light, varying from green to deep purple, violet, and red, and which sometimes covers the hemisphere. This light has often the appearance of luminous vapour, quite transparent, so that the stars are visible through it: this is

also the case with the diffuse light artificially produced in an exhausted receiver, in the way just stated. Such phenomena are commonly followed by wind, rain, and unsettled weather.

In Siberia and high Northern latitudes, the meteor termed *Aurora Borealis* assumes a character of the greatest imaginable splendour. Here the electrical phenomena are most distinctly marked;—here we find beams and rays of light moving with greater or less velocity;—these are termed in the Shetland isles *Merry Dancers*;—also vast columns of light, arches and crowns of various colours, the lower extremities of which frequently quiver with a fiery red colour, and the upper with orange and violet. In Siberia the aurora begins with single bright pillars, which rush about from place to place with great velocity, and finally cover the whole sky; the streamers then meet in the zenith, and appear as if covering the surface of the earth with a vast tent of light glittering with gold, rubies, and sapphires. Brilliant luminous coruscations also frequently occur, accompanied by a crackling sound, like the crackling of sticks when fractured, and very like the ordinary sound of electrical excitation: the noise sometimes is that of a rushing hissing sound, as if the largest fire-works were in action. At this time, the dogs of the Siberian hunters are said to crouch with terror on the ground, and will not move. All these phenomena are purely electrical, and are well produced in turning round a large electrical plate, detached from the conductors of the machine.

The more early observers of the aurora supposed it to occur at very considerable elevations above the surface of the earth, in regions where the air was indefinitely rare. Euler estimated the altitude at some thousands of miles. This meteor, however, by more recent observation, certainly takes place at far less elevations above the earth than is generally supposed, and does not in any case probably reach the sensible limit of the atmosphere. Franklin, at Fort Enterprize, in February, 1821, determined the altitude of an aurora to be less than the elevation of that of denser clouds,—a fact confirmed by Parry in

his third voyage. Lieutenants Sherer and Ross, who accompanied Captain Parry, were, together with Captain Ross, simultaneously surprised at seeing a bright ray of the aurora shoot down from a general mass of light in the heavens, within a distance of them less than 3000 yards.

141. *Water-spouts and Whirlwinds.*—The agencies of electricity in the production of certain atmospheric meteors has been, and not without some show of reason, extended to the phenomena of water-spouts and whirlwinds, which are supposed to arise from the operation of electrical attraction,—water-spouts at sea being what whirlwinds are on land. These have been known to tear up trees and scatter bodies in all directions, and are attended by a rumbling noise. Water-spouts have the form of a speaking-trumpet, with the broad end near the clouds. All these appearances occur in months most liable to thunder-storms, and closely resemble what might be expected from the prolongation of protuberances of electrified clouds towards the sea, occasioning thereby a mutual attraction between the water and the clouds. These appearances are said to be dispersed by mariners by presenting to them sharp-pointed conductors.

*Earthquakes.*—Dr. Stukely, in some very interesting memoirs printed in the 'Transactions of the Royal Society' for 1749 and 1750, has ingeniously referred the phenomena of earthquakes to electrical action within the earth, and endeavours to support this conjecture by reference to the vast extent and mass of earth simultaneously shaken,—from the prevalence of electrical phenomena which attend them, such as coruscations, thunder, lightning, and fire-balls of various kinds. At the time of the great earthquake in London, in 1749, such appearances were in abundance. A sound was observed to roll from the River Thames to Temple Bar before the houses ceased to nod, as when an electrical shock is accompanied by the report of the discharge,—whilst all the mechanical phenomena correspond to the peculiar vibrating motion of disruptive discharge in the substance of imperfect conducting matter. In the con-

cussion at Daventry, in Northamptonshire, in September, 1750, the motion was felt throughout a space of 100 miles in length and 40 in breadth, and 4000 square miles of surface were convulsed in an instant.

That a high amount of electrical action attends these wonderful operations of nature is quite certain, but in the present state of our knowledge of such phenomena, their reference to pure electrical action as a primary cause must necessarily be considered to rest on a very hypothetical basis.

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## VIII.

### CONCLUDING REMARKS.

#### Practical Applications of Electricity.

142. In the foregoing pages we have been anxious to direct the reader's attention to the *principles* of electrical science rather than to its practical applications. When the principles are fairly established, practical applications are comparatively easy, and of less importance to be brought before the general reader in minute detail. In the brief remainder of our space we can only just glance at a few of the examples of *power* which electrical *knowledge* has supplied.

The first and most obvious practical result of Franklin's discoveries was the *lightning conductor* for the protection of buildings and ships from the violent effects of the disruptive discharge. The first lightning conductors consisted merely of metallic rods or chains proceeding from the highest point of the building or the ship, in a direct line to the earth or to the sea: but this was not found in all cases sufficient. Instances have occurred in which the conductors have been fused or shattered,—and hence arose a prejudice against their use, under the idea that they did more harm than good, by inviting *the destruction they were intended to prevent*. An attentive

examination, however, of numerous cases of damage from lightning has shown that the path of discharge from the cloud to the earth is always in the line of least resistance. This line may not be the shortest lineal distance, but it is in all cases the shortest electrical distance; that is, the lightning picks out the best conductors in its transit to the earth, selecting with the nicest discrimination metal cramps and fastenings (106), and by its expansive power shattering and destroying inferior conducting substances, such as wood, brick-work, and stone. Now in order to arm a fabric effectually against these effects, the conductor ought in the first place to be of metal; but as metals vary greatly in their conducting power,\* one of the best conducting metals ought to be selected. For this purpose copper has many advantages over any other metal. In the second place, to prevent any possibility of fusion, the conductor ought to be capacious. Now it has been found that a copper rod three-quarters of an inch in diameter, or an equal quantity of copper, under any other form, is capable of resisting the heating effect of any discharge of lightning whose effects have hitherto been recorded. Thirdly, the conductor should consist of several branches, with pointed rods projecting freely into the air from distant summits of the building, and connected by horizontal branches passing along the ridges of roofs, and from these sending off other branches to the ground. All great masses of metal occurring near the conductor should be bound up with it, and all the terminating branches should be connected with a spring of water, a drain, or some other conducting channel. In a ship, each mast should have its own capacious conductor permanently fixed and connected with bands of copper passing through the sides of the ship under the deck-beams, and with the large bolts leading through the keels and keelson, and including, by other connections, all the principal metallic masses employed in the construction of the hull. Under such a system, a discharge of lightning falling

\* Taking lead as unity, the conducting power of tin is 2; iron, 2.4; zinc, 4; and copper, 12.

on a house or a ship finds its way to the earth or to the sea without the possibility of damage. The great principle in applying such conductors is to place the ship or building in the same electrical condition it would assume supposing the whole were a solid mass of metal, or as nearly as may be, and the conductor should be so applied that a discharge of lightning falling on the general mass cannot enter upon any circuit of which the conductor does not form a part.

143. The heating power of electricity has been turned to account in promoting the efficacy and lessening the danger of blasting on land and under water. Instead of the common fuse, as ordinarily employed, a fine wire is passed through a charge of gunpowder, properly secured, and the ends of the wire are connected with the terminal wires of a Voltaic battery: the moment the circuit is completed the fine wire is ignited, and the powder explodes.

144. The lighting of public streets and buildings by Voltaic electricity is an application of the Voltaic battery which has occupied the minds of practical men during many years past, from the time, indeed, when Davy produced the magnificent arc of light from his large battery at the Royal Institution. So long as Voltaic batteries were costly and their effects only of short duration, this application was deemed impracticable; but when *constant* batteries, as they are called, were invented at a cheap rate, the attempts to apply the light from the charcoal points to the purposes of illumination became numerous. It is now some years since M. Archereau exhibited his splendid Voltaic light in the streets of Paris, and at the time we are writing Mr. Staite is dazzling the metropolis in a similar way. These gentlemen seem to have overcome the difficulties which prevented their predecessors from obtaining a *continuous* light; so that we may hope shortly to see the light brought into use. Its intensity seems eminently qualified for light-houses, railway signals, &c.

145. The constant battery invented by the late Professor *Daniell* has also given rise to an art which, within a very short

period, rose to an eminent position among the useful arts and manufactures of this country, namely, *electro-metallurgy*, *electrotype*, *galvano-plastic*, &c. The principle upon which these arts rest, is that of the constant battery, which is an arrangement of copper and zinc, but is excited by *two* fluids instead of one, as in the common form of battery, fig. 13, p. 22. The two fluids are separated by a diaphragm which prevents them from mixing, but allows the electric current to pass. There is first a rod of zinc immersed in dilute sulphuric acid contained in the porous diaphragm, surrounding which is a solution of sulphate of copper contained in a copper cell. By this arrangement the solution of zinc is kept away from the copper, and the hydrogen, instead of escaping, passes through the porous diaphragm with the current, when it combines with the oxygen of the oxide of copper in the solution of the sulphate, thereby leaving pure metallic copper, which lines the interior of the copper cell. By peculiar contrivances this copper can be made to deposit itself as fast as it is formed on moulds of wax rubbed over with black lead to give a conducting surface; or the copper can be deposited on the cylinders and rollers used in calico printing, working out the pattern with the greatest fidelity; also in surface printing, etching, and various other applications. When the solution of some salt of gold or silver is used, instead of the solution of copper, a thin film of either of the precious metals can by Voltaic action be deposited on articles and ornaments previously cast and finished in some inferior metal. In this way basket-work, fruits, leaves, flowers, busts, grapes, statues, and medallions have received coatings of copper (125) and other metals: \* even Daguerreotype plates have been copied by this means. By writing on a metallic surface with a peculiar kind of varnish, and then depositing the copper over the lines, a plate fit for printing is obtained. By Voltaic arrangements metals can be assayed; iron covered with thin films of lead, thereby pre-

\* By Mr. Woolrich's application of magneto-electricity (130) to electroplating, the Voltaic battery can be altogether dispensed with.



venting rust; pins can be tinned, and numerous other applications are daily being made.

146. But perhaps the most wonderful application of electricity to the purposes of life is the facility it affords to persons, separated by hundreds of miles, to hold instant communication, by night or by day, giving them the power, as it were, to annihilate space (117), enabling them to consult, admonish, inform, condole with each other, as if they were in the same room, and, having ended their conversation, to turn aside, and one to find himself in London and the other in Edinburgh. There is nothing in fiction more wonderful than this; yet the means are apparently so inadequate, resting as they do,—first, upon the simple principle discovered by Ørsted, in 1819, that a magnetic needle, free to rotate about its centre, when brought near to a wire through which an electric current is passing, tends to place itself at right angles to that wire (127), the direction of its motion following a certain law; and, secondly, that a piece of soft iron is rendered magnetic during the transmission of an electric current along a wire coiled spirally round it, when placed near the wire which connects the poles of a Voltaic battery (25). In the earliest form of electric telegraph a number of magnetic needles were thus arranged at the two extremities of a line of railway, and also at some of the intermediate stations. Each needle had its own wire,\* so that any deflection produced on any one needle at any part of the line caused a similar deflection in all the other needles connected with the same wire. Thus, by operating upon two or more wires at once, or in rapid succession, the needles could be thrown into certain positions, which, by previous arrangement, should be made to represent certain symbols,

\* In the earlier forms of electric telegraph, it was thought necessary, in order to close the current, that each needle should be furnished with a return wire; but it has been abundantly proved, by experiment, that water, or even the moisture contained in the earth, is sufficient to transmit the current back to the battery. Hence, in all telegraphic arrangements, *the return wire is now dispensed with.*

letters, or words. In the latter forms of telegraph, Professor Wheatstone has availed himself of the power of the Voltaic current to confer a magnetic condition upon soft iron, which is destroyed the moment the current is suspended or cut off. A number of constant cells is used to convert cylinders of soft iron (2 inches long by  $\frac{1}{4}$  an inch in diameter) at the distant station into electro-magnets. Whenever contact is made, the keeper of the magnet is attracted; and when contact is broken, the keeper is removed by a spring. In one form of the instrument, two drivers attached to the keeper act upon a toothed wheel, and convert the alternate into a circular motion, which is transmitted to an axis bearing a signal-disc or *indicator*. In this case the resistance of the wires, as compared with the electromotive powers of the batteries, is not great; but where the resistance is great, the keeper has merely to move a detent which liberates the toothed wheel, and allows motion to be given to the indicator by a clock movement. A *commutator* at one station is furnished with a disc corresponding to the indicator, so that when any sign of one is brought by the hand of the attendant to the place of observation, the corresponding sign is exposed by the other at the distant station. Each station has a commutator and indicator, all four being included in the one circuit of a wire each way. *Time* may also be transmitted instead of signals, and hence we get what Professor Wheatstone calls the *electric clock*. For this purpose, the indicator is fixed, and furnished with a clock face, the axis carrying an index or hand: the communicating disc is moved round by the oscillations of a pendulum. In this way one good clock can be made to communicate its own time to a series of *skeleton* clocks at any distance.

Various contrivances have been made for *registering* or *printing* the signals. For example, each letter of the indicating disc is attached to a spring radiating from the centre; when the letter is brought, by the action of the instrument, to the proper place for indicating a required signal, a hammer, acted on by clock-work, set free by a second electro-magnet,

strikes the letter upon a pad of manifold writing paper and fair paper, and so registers the signal. A cylinder rotating on a spiral axis exposes fresh surfaces of paper. The same current which works these telegraphs also rings an alarm to call attention.

147. Amongst other applications of electricity to useful purposes, the application of magneto-electrical action (133) in arresting the oscillations of the compass card on board ship is not the least important. We have seen that electrical currents are excited in non-magnetic metals, such as copper, zinc, &c., when placed near a magnet in motion, or when themselves set in motion near a magnet (134) : these currents so tend to arrest the motion that if an ordinary magnetic needle be caused to oscillate within a ring of copper, the amplitude of the oscillations rapidly diminishes, and the needle is speedily reduced to rest. On this principle, the common compass card employed at sea is placed within a dense ring of copper,—the poles or extremities of the magnet being near the interior of the ring ; this, with some valuable and judicious improvements in the construction and mounting of the needle, so fetters the vibrations, that even although the instrument be extremely sensible of the least motion, and of the action of the magnetic force of the earth, yet the compass card is found steady in the heaviest sea, and under the violent motion of steam boats when struggling with a gale ; so that instead of wandering about, as was frequently found to happen in compasses of the ordinary kind, the card remains steady, and yet sensible to the least directive force of the earth.

148. *Electricity applied as a moving power.* The operation of Voltaic electricity in magnetizing iron (129), and the disappearance of the excited magnetism directly its action is suspended, or nearly so, has furnished a means of obtaining to a certain extent a considerable moving force applicable to the purposes of machinery ; and although in all the attempts hitherto made, engines of great practical value have not been obtained, yet

very considerable advances have been made and are still making in electrical machines.

The general principles resorted to in the construction of electro-magnetic engines are these,—either a rapid change of polarity in masses of iron surrounded by spiral coils (129), so as to cause them to alternately attract and repel other electro-magnets brought within their influence, or otherwise a rapid magnetizing and demagnetizing of masses of iron in a similar way, without any change of polarity by which an attractive force is brought to act upon other masses of iron, so long as the attraction is operative in pulling them onward, and no longer. In both these cases a rotatory motion is obtained by fixing the attracted masses on the circumference of a wheel, and placing the wheel so as to admit of the operation of the electro-magnets upon the extremities of its radii, as in any other similar case of the application of a moving force to the circumference.

Professor Jacobi, of St. Petersburg, by means of an engine on this principle, succeeded, in the years 1838 and 1839, in propelling a boat upon the Neva at the rate of 4 miles an hour. This boat was 28 feet in length, about 7 feet wide, and drew nearly 3 feet water. It contained ten persons; the engine was worked by a Voltaic battery of sixty-four pairs of platinum plates, excited with nitric and sulphuric acid, and propelled the vessel through the medium of paddle-wheels. Mr. Llewelyn exhibited to the members of the British Association, in August \* last, a similar experiment on a lake at his beautiful residence near Swansea. By means of an electro-magnetic engine, contrived with singular skill and ingenuity, he propelled a small boat with considerable force through the medium of a screw propeller. Jacobi subsequently applied his engine to working machinery, but not with any great success.

In 1842, Mr. Davidson constructed an electro-magnetic locomotive engine, which was tried on the Edinburgh and

\* Meeting of the British Association at Swansea, August, 1848.

Glasgow Railway, the carriage of which was 16 feet in length and 6 feet wide, weighing above five tons, including the batteries and magnets. It was propelled at the rate of about 4 miles an hour.

Wheatstone, Talbot, Hearder, and several others engaged in this branch of science, have given models of electro-magnetic engines which display great powers of invention; and although as yet the perfection of such engines and their practical and commercial advantages remain in great uncertainty, still it is to be considered that the development of the principles upon which they have been constructed is of very recent date, and the question altogether in its infancy. The speed obtained by Professor Jacobi, of 4 miles an hour with a boat on the Neva, certainly exceeded that resulting from the first attempts to propel vessels by steam. We have, therefore, yet to hope for vast improvements in electric machines. When we consider that an electro-magnet is now exhibiting in London, which attracts a mass of iron at one-eighth of an inch distance with a force of 1344 lbs., and requires no less than 4764 lbs. (more than 2 tons) to separate the contact, it is really difficult to assign limits to the application of a moving force derived from the agency of electricity.

THE END.

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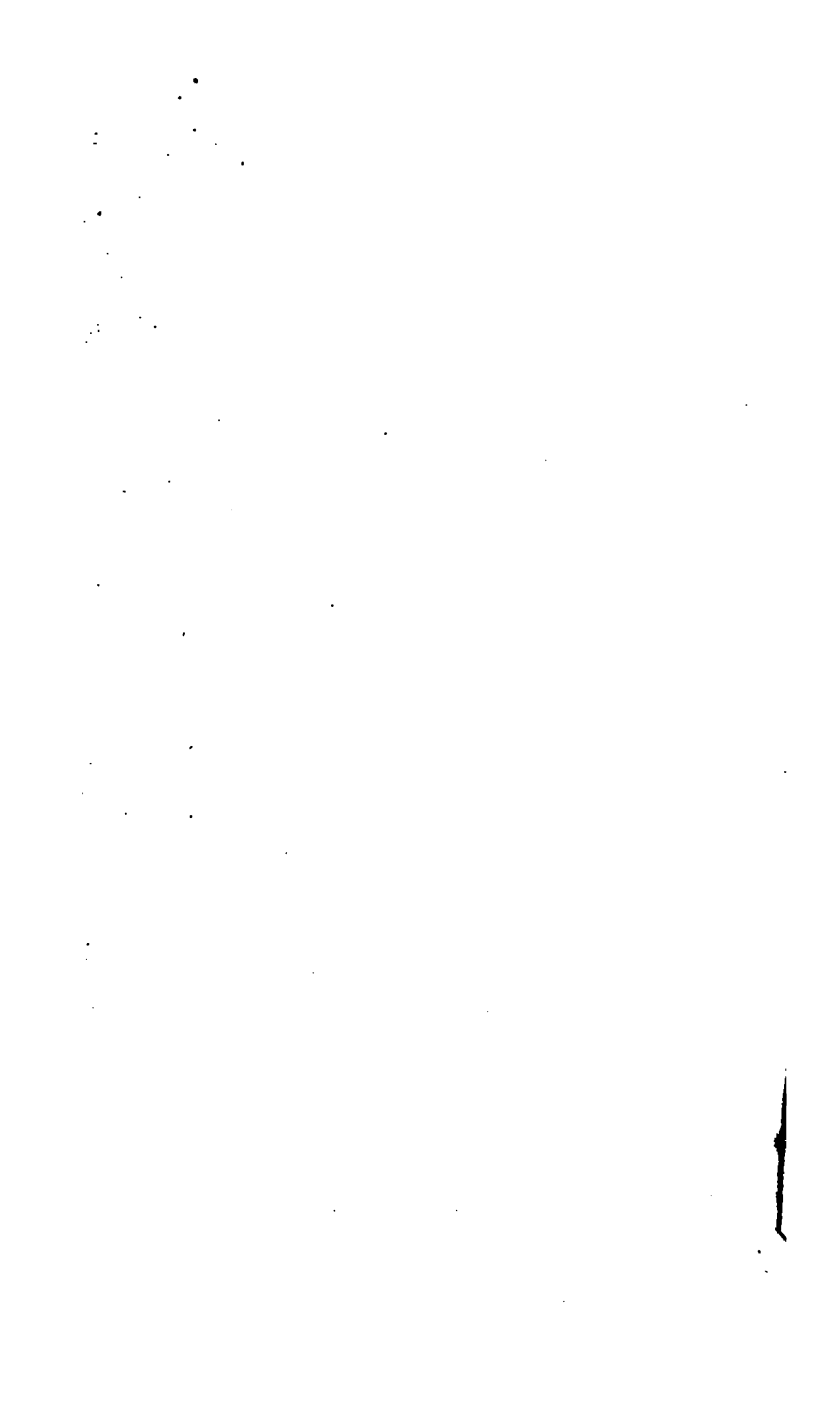
CORRECTIONS.

Page 98, line 10 from the bottom, for "decreases," read "increases."

Page 99, line 11 from the bottom, for "double," read "increase."

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